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The Trusted Integrator for Sustainable Solutions

REMOVAL SUPPORT TEAM 3
EPA CONTRACT EP-S2-14-01

February 4, 2019

Mr. Gez Bushra, On-Scene Coordinator
U.S. Environmental Protection Agency, Region II
Removal Action Branch
2890 Woodbridge Avenue
Edison, NJ 08837

EPA CONTRACT No: EP-S2-14-01

TDD No: TO-0370-0117

DC No: RST3-05-D-0139

**SUBJECT: SITE-SPECIFIC UFP QUALITY ASSURANCE PROJECT PLAN
PORT REFINERY
RYE BROOK, WESTCHESTER COUNTY, NEW YORK**

Dear Mr. Bushra,

Enclosed please find the Site-Specific Uniform Federal Policy (UFP) Quality Assurance Project Plan (QAPP) for the Removal Action activities to be conducted at the Port Refinery Site (the Site) located in Rye Brook, Westchester County, New York. This plan covers the air sampling to be conducted at the Site on February 5, 2019.

If you have any questions or comments, please do not hesitate to contact me at (908) 565-2974.

Sincerely,

WESTON SOLUTIONS, INC.

For: Michael Lang
RST 3 Site Project Manager

Enclosure:
cc: TDD File: TO-0370-0117

an employee-owned company



In association with Scientific and Environmental Associates, Inc.,
Environmental Compliance Consultants, Inc., Avatar Environmental, LLC,
On-Site Environmental, Inc., and Sovereign Consulting, Inc.

SITE-SPECIFIC UFP QUALITY ASSURANCE PROJECT PLAN

PORT REFINERY SITE

Rye Brook, Westchester County, New York

SSID No.: 027T

EPA ID No: NYD986954048

DC No: RST3-05-D-0138

TDD No: TO-0370-0117

EPA Contract No: EP-S2-14-01

Prepared for:

U.S. Environmental Protection Agency, Region II
2890 Woodbridge Avenue
Edison, New Jersey 08837

Prepared by:

Removal Support Team 3
Weston Solutions, Inc.
Federal East Division
Edison, New Jersey 08837

February 2019

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ATTACHMENTS

ATTACHMENT A: Site Location Map

ATTACHMENT B: Sampling SOPs

- ERT/SERAS SOP# 2001: *General Field Sampling Guidelines*
- ERT/SERAS SOP# 2008: *General Air Sampling Guidelines*

ATTACHMENT C: Analytical SOPs

- NIOSH Method 6009

ATTACHMENT D: EPA RSLs for Residential Air

LIST OF ACRONYMS

ADR	Automated Data Review
ANSETS	Analytical Services Tracking System
AOC	Acknowledgment of Completion
ASTM	American Society for Testing and Materials
CEO	Chief Executive Officer
CERCLA	Comprehensive Environmental Response, Compensation and Liability Act
CLP	Contract Laboratory Program
CFM	Contract Financial Manager
CO	Contract Officer
COI	Conflict of Interest
COO	Chief Operations Officer
CRDL	Contract Required Detection Limit
CRTL	Core Response Team Leader
CRQL	Contract Required Quantitation Limit
CQLOSS	Corporate Quality Leadership and Operations Support Services
CWA	Clean Water Act
DCN	Document Control Number
DESA	Division of Environmental Science and Assessment
DI	Deionized Water
DPO	Deputy Project Officer
DQI	Data Quality Indicator
DQO	Data Quality Objective
EM	Equipment Manager
EDD	Electronic Data deliverable
ENVL	Environmental Unit Leader
EPA	Environmental Protection Agency
ERT	Environmental Response Team
FASTAC	Field and Analytical Services Teaming Advisory Committee
GC/ECD	Gas Chromatography/Electron Capture Detector
GC/MS	Gas Chromatography/Mass Spectrometry
HASP	Health and Safety Plan
HRS	Hazard Ranking System
HSO	Health and Safety Officer
ITM	Information Technology Manager
LEL	Lower Explosive Limit
MSA	Mine Safety Appliances
MS/MSD	Matrix Spike/Matrix Spike Duplicate
NELAC	National Environmental Laboratory Accreditation Conference
NELAP	National Environmental Laboratory Accreditation Program
NIOSH	National Institute for Occupational Safety and Health
NIST	National Institute of Standards and Technology
OSC	On-Scene Coordinator
OSHA	Occupational Safety and Health Administration

LIST OF ACRONYMS (Concluded)

OSWER	Office of Solid Waste and Emergency Response
PARCCS	Precision, Accuracy, Representativeness, Completeness, Comparability, Sensitivity
PAH	Polynuclear Aromatic Hydrocarbons
PCB	Polychlorinated Biphenyls
PIO	Public Information Officer
PM	Program Manager
PO	Project Officer
PRP	Potentially Responsible Party
PT	Proficiency Testing
QA	Quality Assurance
QAL	Quality Assurance Leader
QAPP	Quality Assurance Project Plan
QMP	Quality Management Plan
QA/QC	Quality Assurance/Quality Control
QC	Quality Control
RC	Readiness Coordinator
RCRA	Resource Conservation and Recovery Act
RPD	Relative Percent Difference
RSCC	Regional Sample Control Coordinator
RST	Removal Support Team
SARA	Superfund Amendments and Reauthorization Act
SEDD	Staged Electronic Data Deliverable
SOP	Standard Operating Practice
SOW	Statement of Work
SPM	Site Project Manager
START	Superfund Technical Assessment and Response Team
STR	Sampling Trip Report
TAL	Target Analyte List
TCL	Total Compound List
TDD	Technical Direction Document
TDL	Technical Direction Letter
TO	Task Order
TQM	Total Quality Management
TSCA	Toxic Substances Control Act
UFP	Uniform Federal Policy
VOA	Volatile Organic Analysis

TABLE 1: Crosswalk: UFP-QAPP Workbook to 2106-G-05 QAPP

Optimized UFP-QAPP Worksheets		2106-G-05 QAPP Guidance Section	
A. Project Management and Objectives			
1 & 2	Title and Approval Page	2.2.1	Title, Version, and Approval/Sign-Off
3 & 5	Project Organization and QAPP Distribution	2.2.3	Distribution List
		2.2.4	Project Organization and Schedule
4, 7, & 8	Personnel Qualifications and Sign-Off Sheet	2.2.1	Title, Version, and Approval/Sign-Off
		2.2.7	Special Training Requirements and Certifications
6	Communication Pathways	2.2.4	Project Organization and Schedule
9	Project Planning Session Summary	2.2.5	Project Background, Overview, and Intended Use of Data
10	Conceptual Site Model (CSM)	2.2.5	Project Background, Overview, and Intended Use of Data
11	Project/Data Quality Objectives	2.2.6	Data/Project Quality Objectives and Measurement Performance Criteria
12	Measurement Performance Criteria	2.2.6	Data/Project Quality Objectives and Measurement Performance Criteria
13	Secondary Data Uses and Limitations	Chapter 3	QAPP ELEMENTS FOR EVALUATING EXISTING DATA
14 & 16	Project Tasks & Schedule	2.2.4	Project Organization and Schedule
15	Project Action Limits and Laboratory-Specific Detection/Quantitation Limits	2.2.6	Data/Project Quality Objectives and Measurement Performance Criteria
B. Measurement/Data Acquisition			
17	Sampling Design and Rationale	2.3.1	Sample Collection Procedure, Experimental Design, and Sampling Tasks
18	Sampling Locations and Methods	2.3.1	Sample Collection Procedure, Experimental Design, and Sampling Tasks
		2.3.2	Sampling Procedures and Requirements
19 & 30	Sample Containers, Preservation, and Hold Times	2.3.2	Sampling Procedures and Requirements
20	Field Quality Control (QC) Sample Summary	2.3.5	QC Requirements
21	Field Standard Operating Procedures (SOPs)	2.3.2	Sampling Procedures and Requirements

TABLE 1: Crosswalk: UFP-QAPP Workbook to 2106-G-05 QAPP (Concluded)

Optimized UFP-QAPP Worksheets		2106-G-05 QAPP Guidance Section	
B. Measurement/Data Acquisition			
22	Field Equipment Calibration, Maintenance, Testing, and Inspection	2.3.6	Instrument/Equipment Testing, Calibration and Maintenance Requirements, Supplies and Consumables
23	Analytical SOPs	2.3.4	Analytical Methods Requirements and Task Description
24	Analytical Instrument Calibration	2.3.6	Instrument/Equipment Testing, Calibration and Maintenance Requirements, Supplies and Consumables
25	Analytical Instrument and Equipment Maintenance, Testing, and Inspection	2.3.6	Instrument/Equipment Testing, Calibration and Maintenance Requirements, Supplies and Consumables
26 & 27	Sample Handling, Custody, and Disposal	2.3.3	Sample Handling, Custody Procedures, and Documentation
28	Analytical QC and Corrective Action	2.3.5	QC Requirements
29	Project Documents and Records	2.2.8	Document and Records Requirements
C. Assessment/Oversight			
31, 32, & 33	Assessments and Corrective Action	2.4	ASSESSMENTS AND DATA REVIEW (CHECK)
		2.5.5	Reports to Management
D. Data Review			
34	Data Verification and Validation Inputs	2.5.1	Data Verification and Validation Targets and Methods
35	Data Verification Procedures	2.5.1	Data Verification and Validation Targets and Methods
36	Data Validation Procedures	2.5.1	Data Verification and Validation Targets and Methods
37	Data Usability Assessment	2.5.2	Quantitative and Qualitative Evaluations of Usability
		2.5.3	Potential Limitations on Data Interpretation
		2.5.4	Reconciliation with Project Requirements

a) **Site Name/Project Name:** Port Refinery Site

b) **Site Location/EPA ID No.:** Rye Brook, Westchester County / NYD986954048

c) **Contract/Work Assignment Number:** EP-S2-14-01 / TDD#: 0370-0117

Weston Solutions, Inc.
1090 King Georges Post Road, Suite 201
Edison, New Jersey 08837

2/4/2019
Date

2/4/2019
Date

2/4/2019
Date

02/05/19
Date

Date _____

3

QAPP Worksheet #1& 2: Title and Approval Page (Concluded)

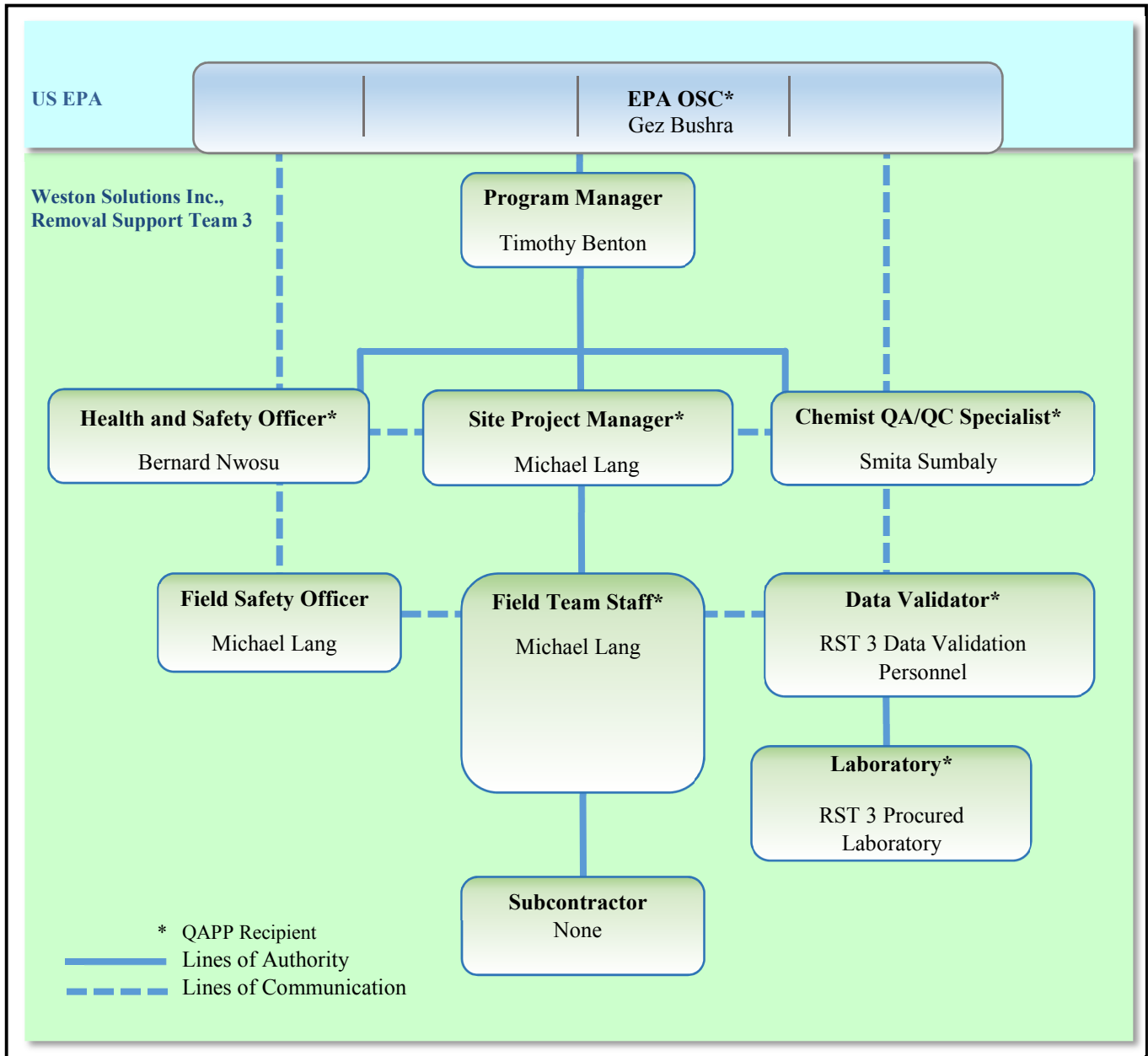
3. List Plans and reports from previous investigation relevant to this project.

Not applicable

Exclusions:

Not Applicable

QAPP Worksheet #3 & 5: Project Organization and QAPP Distribution



Acronyms:

EPA – U.S. Environmental Protection Agency
OSC – On-Scene Coordinator
RST 3 – Removal Support Team 3
QA/QC – Quality Assurance/Quality Control

QAPP Worksheet #3 & 5: Project Organizational and QAPP Distribution (Concluded)

QAPP Recipient	Title	Organization	Telephone Number	Fax Number	E-mail Address	Document Control Number
Gez Bushra	OSC	EPA, Region II	(732) 690-7055	(732) 906-6182	bushra.gezahegne@epa.gov	RST3-05-D-0139
Michael Lang	SPM	Weston Solutions, Inc., RST 3	(908) 565-2974	(732) 225-7037	michael.lang@westonsolutions.com	RST3-05-D-0139
Smita Sumbaly	QAO	Weston Solutions, Inc., RST 3	(732) 585-4410	(732) 225-7037	S.Sumbaly@westonsolutions.com	RST3-05-D-0139
Bernard Nwosu	HSO	Weston Solutions, Inc., RST 3	(732) 585-4413	(732) 225-7037	Ben.Nwosu@westonsolutions.com	RST3-05-D-0139
Site TDD File	RST 3 Site TDD File	Weston Solutions, Inc., RST 3	Not Applicable	Not Applicable	Not Applicable	-

EPA – U.S. Environmental Protection Agency
 OSC – On-Scene Coordinator
 SPM – Site Project Manager
 RST 3 – Removal Support Team 3
 QAO – Quality Assurance Officer
 HSO – Health & Safety Officer
 TDD – Technical Direction Document

QAPP Worksheet #4, 7 & 8: Personnel Qualification and Sign-off Sheet

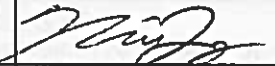
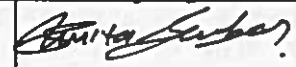
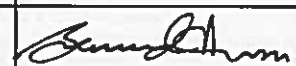
Project Function	Specialized Training By Title or Description of Course	Training Provider	Training Date	Personnel / Groups Receiving Training	Personnel Titles / Organizational Affiliation	Location of Training Records / Certificates ¹	Date of Training
[Specify location of training records and certificates for samplers]							
QAPP Training	This training is presented to new employees and Site project Manager's to introduce the provisions, requirements, and responsibilities detailed in the QAPP. The training presents the relationship between the QAPPs, SOPs, and work plans. QAPP refresher training will be presented to all employees following a major QAPP revision.	QA Officer	As needed	All RST 3 members upon initial employment and as refresher training	Weston Solutions, Inc., RST 3	Within Division	February 2018
Health and Safety Training	Health and safety training will be provided to ensure compliance with Occupational Safety and Health Administration (OSHA) as established in 29 CFR 1910.120.	Health and Safety Officer	Yearly at a minimum	All Employee upon initial employment and as refresher training every year	Weston Solutions, Inc., RST 3	Within Division	February 2018
Others	SCRIBE, ICS 100 and 200, and Air Monitoring Equipment Trainings provided to all employees	EPA ERT – all trainings	Upon initial employment and as needed				February 2018
	Dangerous Goods Shipping	JJ Keller Corporation	Every 3 years				February 2018

All team members are trained in the concepts and procedures in recognizing opportunities for continual improvement, and the approaches required to improve procedures while maintaining conformance with legal, technical, and contractual obligations.

¹If training records and/or certificates are on file elsewhere; document their location in this column. If training records and/or certificates do not exist or are not available, then this should be noted.

QAPP Worksheet #4, 7 & 8: Personnel Qualification and Sign-off Sheet (Concluded)

Organization: Weston Solutions, Inc., RST 3

Name	Project Title/Role	Education and Experience Qualifications	Specialized Training/ Certifications	Organizational Affiliation	Signature	Date
Michael Lang	SPM, RST 3	3+ years*	Implementing and executing the technical, QA and health and safety during sampling event, sample collection and sample management.	Weston Solutions, Inc.		2/4/2019
Smita Sumbaly	QAO, RST 3	30 years	Chemist QA/QC Specialist	Weston Solutions, Inc.		2/4/2019
Bernard Nwosu	HSO, RST 3	25 years	Health and Safety Officer	Weston Solutions, Inc.		2/4/2019

*All RST 3 members, including subcontractor's resumes are in possession of RST 3 Program Manager, EPA Project Officer, and Contracting officers.

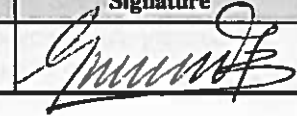
SPM – Site Project Manager

RST 3 – Removal Support Team 3

QAO – Quality Assurance Officer

HSO – Health & Safety Officer

Organization: EPA Region II, OSC

Name	Project Title/Role	Education and Experience Qualifications	Specialized Training/ Certifications	Organizational Affiliation	Signature	Date
Gez Bushra	EPA OSC	NA	All project coordination, direction and decision making.	EPA, Region II		2/6/19

EPA – U.S. Environmental Protection Agency

OSC – On-Scene Coordinator

NA – Not Applicable

QAPP Worksheet #6: Communication Pathways

Communication Drivers	Responsible Entity	Name	Phone Number	Procedure
Point of contact with EPA OSC	SPM, Weston Solutions, Inc., RST 3	Michael Lang	(908) 565-2974	All technical, QA and decision-making matters in regard to the project (verbal, written or electronic)
Adjustments to QAPP	SPM, Weston Solutions, Inc., RST 3	Michael Lang	(908) 565-2974	QAPP approval dialogue
Health and Safety On-Site Meeting	HSO, Weston Solutions, Inc., RST 3	Michael Lang	(908) 565-2974	Explain Site hazards, personnel protective equipment, hospital location, etc.
Lab Data Quality Issues (including sample receipt variances and laboratory quality control variances)	Laboratory Project Manager Test America	Tina Paulauskas	602-659-7617	Laboratory PM will report any issues with project samples to the WESTON Chemist QA/QC Specialist within 1 business day of notification. The WESTON Chemist QA/QC Specialist will contact the field sampler if necessary to resolve sample receiving discrepancies.
Data verification and data validation issues	RST 3 Data Validator	Smita Sumbaly	732-585-4410	The RST 3 Data Validator will review the data verification and validation.
Analytical Corrective Actions	WESTON Chemist QA/QC Specialist RST 3 Data Validator or Laboratory PM	Smita Sumbaly; Tina Paulauskas	732-585-4410 602-659-7617	If laboratory corrective actions are necessary, the WESTON Chemist QA/QC Specialist will communicate with laboratory project manager.
Data Tracking and Management, Release of Analytical Data	WESTON SPM, Operations Manager	Michael Lang	(908) 565-2974	The need for corrective actions will be determined by the SPM upon review of the data. No analytical data will be released prior to validation and all releases must be approved by the Chemist QA/QC Specialist, SPM and EPA OSC/TM.

EPA: U.S. Environmental Protection Agency
OSC: On-Scene Coordinator
SPM: Site Project Manager
RST 3: Removal Support Team 3
HSO: Health and Safety Officer
QA/QC: Quality Assurance/Quality Control
TBD: To Be Determined

QAPP Worksheet #9: Project Planning Session Summary

Date of Planning Session: 1/31/2019				
Location: Phone Call				
Purpose: Scoping meeting for UFP-QAPP for EPA Region II Removal Support Team 3				
Name	Title/Role	Organization	E-mail Address	Phone No.
Gez Bushra	EPA OSC	EPA	bushra.gezahegne@epa.gov	(732) 690-7055
Michael Lang	RST 3 SPM	WESTON	Michael.lang@westonsolutions.com	(908) 565-2974

Notes/Comments: Site-Specific Initial Scoping Meeting:

Weston Solutions, Inc., Removal Support Team 3 (RST 3) has been tasked by the U.S. Environmental Protection Agency (EPA) with providing one Core Response Team (CRT) member with Level D personal protective equipment (PPE) capability to provide Removal Action support at the Port Refinery Site (the Site). RST 3 will perform air sampling for mercury at seven indoor locations to be determined on-Site by the EPA On-Scene Coordinator (OSC). Air samples will be collected at a low flow rate over an 8-hour period of time. All Site activities will be noted in the Site logbook and documented with photographs.

Consensus Decisions Made:

The Removal Action is scheduled to begin on February 5, 2019 and will be completed in one day. The analytical results from this sampling event will enable EPA determine the air quality and any potential environmental impact to the residence.

QAPP Worksheet #9: Project Planning Session Summary (Concluded)

Action Items:

Action	Responsible Party	Due Date
Prepare CLP Analytical Request Form	SPM, RST 3	2/1/2019
Develop Health and Safety Plan	SPM, RST 3	2/1/2019
Develop QAPP	SPM, RST 3	2/4/2019
Develop Work Plan (driller, sampler, survey, etc.)	SPM, RST 3	Not Required
Develop Equipment List	SPM, RST 3	1/31/2019
Develop Site-Specific Data Management Plan	SPM, RST 3	2/1/2019

QAPP Worksheet #10: Conceptual Site Model

Background Information:

The Site is situated on a residential property located at 55 Hillandale Road in the Village of Rye Brook, Westchester County, New York. The geographic coordinates of the Site are 41° 1' 42" north latitude and 73° 40' 26" west longitude. The Site includes adjacent residential properties on both sides (51 and 57 Hillandale Road) and is situated in an affluent neighborhood. The Site also includes a common area of an adjacent condominium complex, The Arbors, as well as sections of a storm drainage system consisting of underground pipes, an open stream channel, and four small ponds situated on three additional residential properties down-gradient from Hillandale Road.

On February 20, 1991, the Westchester County Department of Health (WCDH) notified EPA of a release of elemental mercury at 55 Hillandale Road. On May 1, 1991, the New York State Department of Environmental Conservation (NYSDEC) requested that EPA take appropriate actions at the Site under Section 104 of Comprehensive Environmental Response, Compensation, and Liability Act (CERCLA), 42 U.S.C. § 9604. The Site was subsequently added to the Comprehensive Environmental Response, Compensation, and Liability Information System (CERCLIS) database under CERCLIS ID No. NYD986954048.

From September 1991 to mid-1996, EPA conducted a Removal Action at the Site. This action included excavation, removal, and off-site disposal of 5,300 tons of mercury-contaminated soil from the 55 Hillandale Road property. Included with the excavation and removal of mercury-contaminated soil was the demolition of an on-Site garage, swimming pool, and associated cabana along with removal of mercury-contaminated sediment from three small ponds that were associated with the storm drainage system located on the Site. This Removal Action also addressed contaminated sediment and debris from two drainage sumps and associated piping at the 55 Hillandale Road property which was removed and consolidated into 78 drums for off-Site disposal. On April 29, 2004, the Village of Rye Brook Police Department received a report of a release of elemental mercury that appeared to come from a rock pile along a paved walkway owned by The Arbors. The point of the release was adjacent to and northwest of the property located at 55 Hillandale Road. With notification and verification of the release of mercury by WCDH and EPA, a verbal authorization for \$250,000 was granted by the Acting Regional Administrator on April 29, 2004 to initiate an Emergency Response to address threats associated with the release of mercury.

On April 30, 2004, EPA mobilized its Emergency and Rapid Response Services (ERRS) Contractor to address the release of elemental mercury. Approximately 1 ton of soil containing elemental mercury was excavated and loaded into drums and temporarily staged inside the Village of Rye Brook Police Department storage shed pending disposal. Post-excavation soil samples collected at the base of the excavation identified mercury at concentrations greater than the previously established Site-Specific Action Level of 10 milligrams per kilogram (mg/kg).

From May thru July 2004, with support from the Removal Support Team (RST) Contractor, Weston Solutions, Inc., EPA collected surface soil samples near the area where mercury-contaminated soil was discovered in May 2004 to evaluate any additional areas of contamination. Analytical results of the soil samples indicated that mercury-contaminated soil, above the Site-

QAPP Worksheet #10: Conceptual Site Model (Concluded)

Specific Action Level of 10 mg/kg, was present in areas south and southeast of the paved walkway, following the surface drainage pathway toward the storm drainage system that flows to the southeast from the area. Sampling results confirmed that the mercury contamination extended beyond the rock pile area and onto adjoining properties. Based on these results, EPA expanded its soil investigation onto several properties adjacent to the original spill area. On August 31, 2004, a 6 foot temporary chain-link fence with warning signs was erected around the known contaminated areas on The Arbors' property to restrict access from the paved walkway that connects the neighborhood with Blind Brook High School.

On September 30, 2004, a verbal authorization for a ceiling increase of \$787,000 was granted by the Acting Regional Administrator to continue the Emergency Response action at the Site.

In an effort to fully delineate the horizontal and vertical extent of mercury contamination, a comprehensive investigation of surface and subsurface soils, storm water drainage sediment, groundwater, and indoor air was conducted at the Site between September 15, 2004 and December 17, 2004. The results of this comprehensive investigation determined that mercury contamination was prevalent throughout the 55 Hillandale Road property, adjacent properties, as well as residential properties hydrologically downgradient.

Water samples collected from a basement sump located at 55 Hillandale Road revealed elevated concentrations of mercury. This water, which was collecting from within the sump, was found to be a contributing source of mercury detected within the indoor air of the residence. The sump was very active due to the high water table. To mitigate the elevated mercury contamination entering the sump a Kinetic Degradation Fluxion filter was installed to treat the sump water prior to being discharged from the residence.

On June 20, 2005, an Action Memorandum for the Site was approved that documented the two verbal authorizations, as well as approving the requests for a change-in-scope, ceiling increase, and 12-month and \$2 million exemption. The additional funding was for the excavation and off-Site disposal of mercury-contaminated soil from The Arbors and adjoining properties, including the residence at 55 Hillandale Road. These costs also included compensation to the property owners of 55 Hillandale Road for the demolition of their home and to cover temporary relocation expenses until excavation activities were completed and a replacement structure could be built.

On June 24, 2005, EPA received approval from the Director of the Office of Emergency Management to offer compensation in the form of a financial settlement to the owner of the residence located at 55 Hillandale Road.

Due to the extensive mercury-contaminated soil identified during excavation and sampling activities, a mitigation ceiling increase was requested to continue mitigation efforts. On September 29, 2005, an Action Memorandum was approved by the Deputy Assistant Administrator for the Office of Solid Waste and Emergency Response (OSWER) for a ceiling increase of \$2,105,000 bringing EPA's total project ceiling to \$8,084,000.

A Removal Action was conducted at the Site at 55 Hillandale Road and the surrounding areas from May 2005 through September 2007.

QAPP Worksheet #11: Project/Data Quality Objectives

1. State the Problem:

EPA has completed the excavation and removal of mercury-contaminated soil at the Site, and has tasked RST 3 with the collection of up to 10 air samples, including quality assurance/quality control (QA/QC) samples to be submitted to the assigned laboratory for mercury analysis in order to verify that there is no concern of mercury vapor in the residence located in proximity to the Site.

2. Identify the Goals of the Study:

The analytical results from this sampling event will be compared with the EPA Regional Screening Level (RSL) for mercury in residential air in order to determine the air quality and any potential environmental impact to the residence sampled.

3. Identify Information Inputs:

Up to 10 air samples will be collected during the Removal Action, including QA/QC samples consisting of one field duplicate, one field blank and one lot blank for every new lot of sorbent tubes utilized.

4. Define the Boundaries of the Study:

Overall project objectives include: Air samples will be collected using air samplers consisting of sorbent tubes [6 x 70 millimeter (mm) size, 200 milligram (mg) sorbent] and GilAir pumps with flow rates set between 0.15 and 0.25 liters per minute (L/min) for a period of eight hours. Air samples collected will be submitted to the assigned laboratory for mercury analysis. Air sampling will be conducted at up to seven locations to be determined on-Site by the EPA OSC.

Who will use the data? Data will be used by the EPA, Region II OSC.

5. Develop the Analytic Approach:

Analytical Techniques: Off-site laboratory analysis

Type of Data: Definitive data

Matrix: Ambient Air

Parameters: Mercury via NIOSH Method 6009

Sampling Equipment: Air samplers consisting of sorbent tubes and GilAir Pumps

Access Agreement: Obtained by EPA, Region II OSC.

Sampling Locations: Sample locations will be identified by the EPA OSC.

How much data are needed? Up to 10 air samples, including QA/QC samples.

QAPP Worksheet #11: Project/Data Quality Objectives (Concluded)

6. Specify Performance or Acceptance Criteria:

How “good” does the data need to be in order to support the environmental decision?

Sampling/analytical measurement performance criteria (MPC) for Precision, Accuracy, Representativeness, Completeness, and Comparability (PARCC) parameters will be established. Refer to Worksheet #12, criteria for performance measurement for definitive data.

Where, when, and how should the data be collected/generated?

The location of the air samples will be determined on-Site by the EPA OSC. The sampling event is scheduled to begin on February 5, 2019 and will be completed in one day. All samples will be collected using methods outlined in the EPA ERT/SERAS contractor’s SOP Nos: 2001: *General Field Sampling Guidelines* and 2008: *General Air Sampling Guidelines*.

7. Develop the Detailed Plan for Obtaining Data:

Who will collect and generate the data? The air samples will be collected by RST 3 and submitted to a RST3-procured laboratory for mercury analysis. The analytical results will be generated by the assigned laboratory and will be validated by RST 3 data validation personnel.

How will the data be reported? All data will be reported by the RST 3-procured laboratory (Preliminary, Electronic, and Hard Copy format). The Site Project Manager will provide a Sampling Trip Report, Status Reports, Maps/Figures, Analytical Report, and Data Validation Report to the EPA OSC.

How will the data be archived? Electronic data deliverables will be archived in a Scribe database.

**Worksheet #12: Measurement Performance Criteria
Mercury**

Matrix	Ambient Air (GilAir Pumps)	
Analytical Group	Mercury	
Concentration Level	Low Level	
Data Quality Indicators (DQIs)	Measurement Performance Criteria	QC Sample and/or Activity Used to Assess Measurement Performance
Precision	RPD \pm 25%	Laboratory Duplicates
Accuracy/Bias	%R = Within control chart limits	Laboratory Control Sample (LCS)
Precision	\pm 50% RPD	Collocated Field Duplicates
Accuracy/Bias Contamination	No target compound \geq RL	Field Blank Lot Blank

QAPP Worksheet #13: Secondary Data Criteria and Limitations

Any data needed for project implementation or decision making that are obtained from non-direct measurement sources such as computer databases, background information, technologies and methods, environmental indicator data, publications, photographs, topographical maps, literature files and historical data bases will be compared to the data quality objectives (DQOs) for the project to determine the acceptability of the data. Thus, for example, analytical data from historical surveys will be evaluated to determine whether they satisfy the validation criteria for the project and to determine whether sufficient data was provided to allow an appropriate validation to be done. If not, then a decision to conduct additional sampling for the site may be necessary.

Data Type	Data Source (Originating Organization, Report Title, and Date)	Data Uses Relative to Current Project	Factors Affecting the Reliability of Data and Limitations on Data Use
Not Applicable	Not Applicable	Not Applicable	Not Applicable

QAPP Worksheet #14 & 16: Project Tasks and Schedules

Activity	Responsible Party	Planned Start Date	Planned Completion Date	Deliverable(s)	Deliverable Due Date
Develop Project-Specific Health and Safety Plan (HASP)	WESTON	2/4/2019	2/4/2019	HASP	2/4/2019
Develop Project-Specific QAPP	WESTON	2/4/2019	2/4/2019	QAPP	2/4/2019
Procure WESTON-subcontracted laboratory for analytical services	WESTON	1/31/2019	2/4/2019	NA	2/4/2019
Scoping meeting Operational Manager, SPM, HSO, and sampling team to discuss data collection activities, objectives, and logistics	WESTON	1/31/2019	1/31/2019	Meeting Notes	1/31/2019
Mobilization/Demobilization	WESTON	2/5/2019	2/5/2019	Field Notes	2/5/2019
Sample Collection Tasks	WESTON	2/5/2019	2/5/2019	Field Notes	2/5/2019
Analytical Tasks	WESTON	2/6/2019	2/7/2019	Field Notes/Laboratory Reports	2/6/2019
Quality Control Tasks	WESTON	2/6/2019	2/7/2019	Report of Analyses/Data Package	2/8/2019
Data Validation	WESTON	2/14/2019	2/21/2019	Validation Summary Report	2/21/2019
Summarize Data	WESTON	2/14/2019	2/19/2019	Project-Specific Summary Report/Table	2/19/2019
Develop Project-Specific Report	WESTON	2/19/2019	2/26/2019	Draft Project-Specific Report	2/26/2019
Address EPA comments on Draft Project-Specific Report	WESTON	2/26/2019	3/5/2019	Project-Specific Report	3/5/2019
Contract Closeout	WESTON	2/4/2019	6/30/2019	Contract Closeout Report	6/30/2019

QAPP Worksheet #14 & 16: Project Tasks and Schedules (Continued)

Sampling Tasks:

RST 3 has been tasked with collecting up to 10 air samples, including QA/QC samples from seven locations to be determined on-site by the EPA OSC. The air samples will be submitted to the assigned laboratory for mercury analysis.

Analysis Tasks:

Mercury via NIOSH Method 6009.

Decontamination:

Not Applicable.

Quality Control Tasks:

The air samples will be collected for a definitive QA objective. QA/QC samples will be collected consisting of one field duplicate, one field blank and one lot blank for every new lot of sorbent tubes utilized.

Data Management Tasks:

Activities under this project will be reported in status and trip reports and other deliverables (e.g., analytical reports, final reports) described herein. Activities will also be summarized in appropriate format for inclusion in monthly and annual reports. The following deliverables will be provided under this project:

Trip Report: A trip report will be prepared to provide a detailed accounting of what occurred during each sampling mobilization. The trip report will be prepared within two weeks of the last day of each sampling mobilization. Information will be provided on time of major events, dates, and personnel on-site (including affiliations).

Maps/Figures: Maps depicting site layout, contaminant source areas, and sample locations will be included in the trip report, as appropriate.

Analytical Report: An analytical report will be prepared for samples analyzed under this plan. Information regarding the analytical methods or procedures employed, sample results, QA/QC results, chain-of-custody documentation, laboratory correspondence, and raw data will be provided within this deliverable.

Data Review: A review of the data generated under this plan will be undertaken. The assessment of data acceptability or usability will be provided separately, or as part of the analytical report.

QAPP Worksheet #14 & 16: Project Tasks and Schedules (Continued)

Documentation and Records:

All sample documents will be completed legibly, in ink. Any corrections or revisions will be made by lining through the incorrect entry and by initialing the error.

Field Logbook: The field logbook is essentially a descriptive notebook detailing site activities and observations so that an accurate account of field procedures can be reconstructed in the writer's absence. Field logbook will be bound and paginated. All entries will be dated and signed by the individuals making the entries, and should include (at a minimum) the following

1. Site name and project number
2. Name(s) of personnel on-site
3. Dates and times of all entries (military time preferred)
4. Descriptions of all site activities, site entry and exit times
5. Noteworthy events and discussions
6. Weather conditions
7. Site observations
8. Sample and sample location identification and description*
9. Subcontractor information and names of on-site personnel
10. Date and time of sample collections, along with chain of custody information
11. Record of photographs
12. Site sketches

* The description of the sample location will be noted in such a manner as to allow the reader to reproduce the location in the field at a later date.

Sample Labels: Sample labels will clearly identify the particular sample, and should include the following:

1. RST 3 sample identification number
2. Sample collection date and time
3. Analytical parameters
4. Sample preservation

Sample labels will be written in indelible ink and securely affixed to the sample container. Tie-on labels can be used if properly secured.

Custody Seals: Custody seals demonstrate that a sample container has not been tampered with or opened. The individual in possession of the sample(s) will sign and date the seal, affixing it in such a manner that the container cannot be opened without breaking the seal. The name of this individual, along with a description of the sample packaging, will be noted in the field logbook.

Assessment/Audit Tasks: No performance audit of field operations is anticipated at this time. If conducted, performance and system audit will be in accordance with the project plan.

QAPP Worksheet #14 & 16: Project Tasks and Schedules (Concluded)

Data Review Tasks: Analytical data will be validated in accordance with NIOSH Method 6009 and will be reviewed by RST 3 data validation personnel.

QAPP Worksheet #15: Project Action Limits and Laboratory Specific Detection/Quantitation Limits

Analyte	CAS Number	Project Action Limit ¹ (µg/m ³)	Project Quantitation Limit (ppbv)	Achievable Laboratory Limits	
				*MDLs ² (µg/sample)	QLs (µg/sample)
Mercury	7439-97-6	0.31 µg/m ³	NS	0.0250	0.0250

µg/m³ = microgram per cubic meter

*Reporting limit based on sample volume will be collected.

NS- Not Specified

See Attachment D for EPA RSLs for Residential Air

QAPP Worksheet #17: Sampling Design and Rationale

All field sampling and air sampling activities will be performed in accordance with EPA's ERT/SERAS contractor's SOP No. 2001, *General Field Sampling Guidelines* and No. 2008: *General Air Sampling Guidelines*. Up to 10 air samples will be collected during the Removal Action, including QA/QC samples consisting of field duplicate, field blanks and lot blanks.

Air samples will be collected using air samplers consisting of sorbent tubes [6 x 70 (mm) size, 200 (mg) sorbent] and GilAir pumps with flow rates set between 0.15 and 0.25 (L/min) for a period of eight hours. Air sampling will be conducted at up to seven locations to be determined on-Site by the EPA OSC. All sample information will be transcribed into EPA's SCRIBE sample management database from which sample labels and chains of custody (COC) record will be generated. A sample label affixed to an identification tag will be attached to each sorbent tube sample.

The air samples will be submitted to a RST 3-procured laboratory for mercury analysis via NIOSH Method 6009. The air samples will be collected for a definitive data QA objective. One field duplicate, one field blank and one lot blank will be collected.

This sampling design is based on information currently available and may be modified on site in light of field-screening results and other acquired information.

Lab Name/Location/Contact	Matrix	Parameters
Test America Laboratories, Inc. 4625 E. Cotton Center Blvd Suite 189 Phoenix, AZ 85040 Phone: 602-437-3340 Lab Contact: Carlene McCutcheon	Ambient Air	Mercury, via NIOSH Method 6009

QAPP Worksheet #18: Sampling Locations and Methods/SOP Requirements Table

The following information is project-specific and will be included in the site-specific QAPP.

Sampling Location	Matrix	Concentration Units	No. of Samples (identify field duplicates)	Analyte/Analytical Group(s)	Sampling SOP Reference ¹	Comments
Up to 7	Ambient Air	µg/m ³	Up to 8 (1)	Mercury	SOP#s 2001 and 2008	To verify the air quality at the Site

The website for EPA-ERT SOPs is: https://response.epa.gov/site/site_profile.aspx?site_id=2107

QAPP Worksheet #19 & 30: Sample Containers, Preservation, and Hold Times

Matrix	Analytical Group	Analytical and Preparation Method/SOP Reference ¹	Containers (number, size, and type)	Sample Volume	Preservation Requirements (chemical, temperature, light protected)	Maximum Holding Time (preparation/analysis)	Data Package Turnaround Time
Ambient Air	Mercury	NIOSH Method 6009	Anasorb (300 tube or equivalent) & (10) GilAir Pumps	96 L	NR	30 days	7 days

NR- Not Required

QAPP Worksheet #20: Field Quality Control Sample Summary

Matrix	Analytical Group	No. of Field Samples	No. of Field Duplicates	No. of Extra Volume Laboratory QC (e.g., MS/MSD) Samples	No. of Field Blanks	No. of Field Blanks	No of Lot Blanks	Total No. of Samples to Lab
Ambient Air	Mercury	Up to 7	1	NR	1 / day	1	1 / new lot of cassettes	Up to 10

NR – Not Required

QAPP Worksheet #21: Project Sampling SOP References Table

Reference Number	Title, Revision Date and/or Number	Originating Organization	Equipment Type	Modified for Project Work? (Y/N)	Comments
SOP #2001	General Field Sampling Guidelines; Rev. 1.0, June 2013	EPA ERT/SERAS Contractor	Site Specific	N	None
SOP #2008	General Air Sampling Guidelines; Rev 0.0, November 1994	EPA ERT/SERAS Contractor	Summa Canister with Pressure Gauge, Wrench, Teflon Tubing	N	None

See attachment B for SOP Nos. 2001 and 2008

Note the website for EPA ERT/SERAS contractor SOPs is:

https://response.epa.gov/site/doc_list.aspx?site_id=2107&category=Field%20Activities

QAPP Worksheet #22: Field Equipment Calibration, Maintenance, Testing, and Inspection Table

Field Equipment	Calibration Activity	Maintenance Activity	Testing/ Inspection Activity	Frequency	Acceptance Criteria	Corrective Action	Responsible Person	SOP Reference
GilAir Pump	Prior to and after sampling	Clean, keep fully charged	Confirm battery is fully charged and pump is operating properly.	As needed	Per sampling method	Contact equipment manager/manufacture	Vendor/manufacture	NIOSH #6009

NA = Not applicable
RL = Reporting Limit

QAPP Worksheet #23: Analytical SOPs

Analytical Method/Lab SOP#	Title, Revision Date, and URL (if available)	Definitive or Screening Data	Matrix/Analytical Group	SOP option or Equipment Type	Modified for Project Work? (Y/N)*
NIOSH Method 6009	Mercury: Method 6009, Issue 2, dated 15 August 1994 https://www.cdc.gov/niosh/docs/2003-154/pdfs/6009.pdf	Definitive	CVAA	Test America SOP# PE-MET-013, Rev. 5	N

* If yes, explain the modification.

QAPP Worksheet #24: Analytical Instrument Calibration

Instrument	Calibration Procedure	Frequency of Calibration	Acceptance Criteria	Corrective Action (CA)	Person Responsible for CA	SOP Reference ¹
CVAA	Initial calibration (IC), minimum 6-points for all analytes	See NIOSH Method 6009 & SOP	See NIOSH Method 6009 & SOP	See NIOSH Method 6009 & SOP	Analyst	NIOSH 6009

¹ Refer to the Analytical SOPs table (Worksheet 23). A laboratory-specific QA Manual may be referenced on a project-specific basis and will be identified in the site specific QAPP.

GC = Gas Chromatograph

GC/MS = Gas Chromatograph/Mass Spectrometer

CVAA = Cold Vapor Atomic Absorption Spectrophotometer

QAPP Worksheet #25: Analytical Instrument and Equipment Maintenance, Testing, and Inspection Table

Instrument/ Equipment	Maintenance Activity	Testing/Inspection Activity	Frequency	Acceptance Criteria	Corrective Action	Responsible Person for CA	SOP Reference ¹
CVAA / Perkin Elmer FIMS-1001 Hg-CVAA Atomic Absorption Spectrophotometer	As Specified by method	Mercury	As Specified by method	As Specified by method	Perform maintenance, check standards, recalibrate	Laboratory analyst	NIOSH 6009

GC/MS = Gas Chromatograph/Mass Spectrometer

¹Specify the appropriate reference letter or number from Analytical SOP References table (Worksheet #23)

QAPP Worksheet #26 & 27: Sample Handling, Custody, and Disposal

Sampling Organization: Weston Solutions, Inc., RST 3

Laboratory: RST 3-Procured Laboratory

Method of sample delivery (shipper/carrier): FedEx or Hand Deliver

Number of days from reporting until sample disposal: 60 days

Activity	Organization and Title or Position of Person Responsible for the Activity	SOP Reference ¹
Sample Labeling	RST 3 Site Project Manager, RST 3 Sampling Team	EPA-540-R-014-013, October 2014
Chain-of-Custody Form Completion	RST 3 Site Project Manager, RST 3 Sampling Team	EPA-540-R-014-013, October 2014
Sample Packaging	RST 3 Site Project Manager, RST 3 Sampling Team	EPA-540-R-014-013, October 2014
Shipping Coordination	RST 3 Site Project Manager, RST 3 Sampling Team	EPA-540-R-014-013, October 2014
Sample Receipt, Inspection, & Log-in	Laboratory Sample Custodian	As per non-CLP Laboratory SOP
Sample Custody and Storage	Laboratory Sample Custodian /Laboratory Analytical Personnel	As per non-CLP Laboratory SOP
Sample Disposal	Field Personnel/Laboratory Sample Custodian /Laboratory Analytical Personnel	As per non-CLP Laboratory SOP

Sample Identification Procedures: Each sample collected by RST 3 will be designated by a code that will identify the sample in accordance with previous sampling (if applicable). Ambient air samples will be identified by the date the sample will be collected, location, unit, and sample number. Field Blank and Lot blank samples will be designated by FB and LB, respectively.

Example ambient air sample: 190205-ASL01-AS01-01

190205 = Date | ASL01 = Location | AS01 = Unit | 01 = Sample number.

QAPP Worksheet #26 & 27: Sample Handling, Custody, and Disposal (Concluded)

Field Sample Custody Procedures (sample collection, packaging, shipment, and delivery to laboratory): Each sample will be individually identified and labeled after collection, then sealed with custody seals and enclosed in a plastic cooler. The sample information will be on chain of custody (COC) forms, and the samples shipped to the appropriate laboratory via overnight delivery service or courier. COC records must be prepared in Scribe to accompany samples from the time of collection and throughout the shipping process. Each individual in possession of the samples must sign and date the sample COC Record. The chain-of-custody record will be considered completed upon receipt at the laboratory. A traffic report and chain-of-custody record will be maintained from the time the sample is taken to its final deposition. Every transfer of custody must be noted and signed for, and a copy of this record kept by each individual who has signed. When samples are not under direct control of the individual responsible for them, they must be stored in a locked container sealed with a custody seal. Specific information regarding custody of the samples projected to be collected on the weekend will be noted in the field logbook. The chain-of-custody record should include (at minimum) the following: 1) Sample identification number; 2) Sample information; 3) Sample location; 4) Sample date; 5) Sample Time; 6) Sample Type Matrix; 7) Sample Container Type; 8) Sample Analysis Requested; 9) Name(s) and signature(s) of sampler(s); and 10) Signature(s) of any individual(s) with custody of samples.

A separate COC form must accompany each cooler for each daily shipment. The COC form must address all samples in that cooler, but not address samples in any other cooler. This practice maintains the COC for all samples in case of mis-shipment.

Laboratory Sample Custody Procedures (receipt of samples, archiving, and disposal): A sample custodian at the laboratory will accept custody of the shipped samples, and check them for discrepancies, proper preservation, integrity, etc. If noted, issues will be forwarded to the laboratory manager for corrective action. The sample custodian will relinquish custody to the appropriate department for analysis. At this time, no samples will be archived at the laboratory. Disposal of the samples will occur only after analyses and QA/QC checks are completed. Refer to SERAS SOP #4005, *Chain of Custody Procedures*.

QAPP Worksheet #28: Analytical Quality Control and Corrective Action

Matrix	Ambient Air (GilAir Pumps)				
Analytical Group	Mercury				
Concentration Level	Low Level				
Sampling SOP	ERT SOP #2001 and 2008				
Analytical Method/ SOP Reference	NIOSH Method 6009				
Sampler's Name	Michael Lang				
Field Sampling Organization	Weston Solutions, Inc.				
Analytical Organization	RST 3-Procured Laboratory				
No. of Sample Locations	Up to 7				

QC Sample:	Frequency/Number	Method/SOP QC Acceptance Limits	Corrective Action	Person(s) Responsible for Corrective Action	Measurement Performance Criteria
Laboratory Method Blank	1 per < 20 samples	No analyte >CRQL	Suspend analysis unit source recertified	Non-RAS Laboratory Technician	No analyte > RL
Laboratory Control Sample/Laboratory Control Sample Duplicate (LCS/LCSD) - Second Source	1 per ≤ 20 samples	75-125%	Re-inject LCS/LCSD and/or qualify, initiate CAR and report 2) Perform maintenance, if needed.	Non-RAS Laboratory Technician	75-125%
6 Point Calibration Curve – Linear + Calibration Blank (3 points minimum)	Each day of analysis	r>0.995	1) Re-inject curve 2) Prepare new standards 3) Perform maintenance, if needed.	Non-RAS Laboratory Technician	r>0.995

QAPP Worksheet #28: Analytical Quality Control and Corrective Action (Continued)

Matrix	Ambient Air (GilAir Pumps)				
Analytical Group	Mercury				
Concentration Level	Low Level				
Sampling SOP	ERT SOP #2001 and 2008				
Analytical Method/ SOP Reference	NIOSH Method 6009				
Sampler's Name	Michael Lang				
Field Sampling Organization	Weston Solutions, Inc.				
Analytical Organization	RST 3-Procured Laboratory				
No. of Sample Locations	Up to 7				
QC Sample:	Frequency/Number	Method/SOP QC Acceptance Limits	Corrective Action	Person(s) Responsible for Corrective Action	Measurement Performance Criteria
Initial Calibration Verification (ICV), Low Standard at 0.025 μ g Hg or below; Second Source or Independently Prepared Neat Source	Beginning of each analytical batch (after the standard curve)	75 – 125% recovery	1) Re-inject ICV 2) Re-inject curve 3) Prepare new standards 4) Run new curve 5) Perform maintenance, if needed.	Non-RAS Laboratory Technician	75 – 125% recovery
Calibration Blank/Continuing Calibration Blank (CB/CCB)	Beginning, every 10 samples, and end of each analysis day, before or after ICV	< Report Limit is expected	1) Subtract, if necessary and/or qualify and report 2) Recalibrate and reanalyze all samples since last compliant calibration blank, if needed.	Non-RAS Laboratory Technician	< Report Limit is expected

QAPP Worksheet #28: Analytical Quality Control and Corrective Action (Concluded)

Matrix	Ambient Air (GilAir Pumps)				
Analytical Group	Mercury				
Concentration Level	Low Level				
Sampling SOP	ERT SOP #2001 and 2008				
Analytical Method/ SOP Reference	NIOSH Method 6009				
Sampler's Name	Michael Lang				
Field Sampling Organization	Weston Solutions, Inc.				
Analytical Organization	RST 3-Procured Laboratory				
No. of Sample Locations	Up to 7				

QC Sample:	Frequency/Number	Method/SOP QC Acceptance Limits	Corrective Action	Person(s) Responsible for Corrective Action	Measurement Performance Criteria
CCV at Highest Calibration Point; Primary Source	Every 10 samples and at end of analysis	Acceptable recovery is 90 – 110% of true value of the standard	1) Re-inject CCV 2) Re-inject curve 3) Prepare new standards 4) Run new curve 5) Perform maintenance, if needed.	Non-RAS Laboratory Technician	Acceptable recovery is 90 – 110% of true value of the standard
Reporting Limit Verification (RLV) – Digested QC sorbent sample at or below the Reporting Level	Initially, annually and/or if there is a significant change in the background or instrument response.	50 – 150%	1) Re-inject; Re-digest additional spike; raise RL and perform RLV at higher RL.	Non-RAS Laboratory Technician	50-150%

QAPP Worksheet #29: Project Documents and Records

Sample Collection and Field Records			
Record	Generation	Verification	Storage Location/Archival
Field Logbook or Data Collection Sheets	SPM/Field Personnel	Group Leader or Operational Manager	Project File
Chain-of-Custody Forms	SPM/Field Personnel	Group Leader or Operational Manager	Project File
Custody Seals	SPM/Field Personnel	Group Leader or Operational Manager	Project File
Air Bills	SPM/Field Personnel	Group Leader or Operational Manager	Project File
Daily QC Reports	SPM	Group Leader or Operational Manager	Project File
Deviations	SPM/Field Scientist	Group Leader or Operational Manager	Project File
Corrective Action Reports	Delegated QA Manager	Operational Manager or Program Manager or designee	Project File
Correspondence	SPM	Delegated QA Manager	Project File
Field Sample Results/Measurements	SPM/Field Scientist	Delegated QA Manager	Project File
Tailgate Safety Meeting Items	SPM/Field Safety Officer	Delegated QA Manager	Project File

Project Assessments			
Record	Generation	Verification	Storage Location/Archival
Data Verification Checklists	Data validator/Chemist QA/QC Specialist	Group Leader or Operational Manager	Project File
Data Validation Report	Data validator/Chemist QA/QC Specialist	Group Leader or Operational Manager	Project File
Data Usability Assessment Report	Site Project Manager	Group Leader or Operational Manager	Project File
Corrective Action Reports	Group Leader/HSO/Chemist QA/QC Specialist	Group Leader	Project File
Correspondence	Group Leader/HSO/Chemist QA/QC Specialist	Program Manager or designee	Project File

QAPP Worksheet #29: Project Documents and Records (Concluded)

Laboratory Records			
Record	Generation	Verification	Storage Location/Archival
Sample Receipt, Custody, and Checklist	Laboratory Sample Receiving	Laboratory PM/Delegated QA Manager	Laboratory Data Package and Project File
Equipment Calibration Logs	Laboratory Technician	Laboratory PM/Delegated QA Manager	Laboratory Data Package and Project File
Standard Traceability Logs	Laboratory Technician	Laboratory PM/Delegated QA Manager	Laboratory Data Package and Project File
Sample Prep Logs	Laboratory Technician	Laboratory PM/Delegated QA Manager	Laboratory Data Package and Project File
Run Logs	Laboratory Technician	Laboratory PM/Delegated QA Manager	Laboratory Data Package and Project File
Equipment Maintenance, Testing, and Inspection Logs	Laboratory Technician/ Laboratory QA Manager	Laboratory PM/Delegated QA Manager	Laboratory File
Corrective Action Reports	Laboratory QA Manager	Laboratory PM/Delegated QA Manager	Laboratory File and Project File
Laboratory Analytical Results	Laboratory Technician/ Laboratory QA Manager	Laboratory PM/Delegated QA Manager	Laboratory Data Package and Project File
Laboratory QC Samples, Standards, and Checks	Laboratory Technician/ Laboratory QA Manager	Laboratory PM/Delegated QA Manager	Laboratory Data Package and Project File
Instrument Results (raw data) for Primary Samples, Standards, QC Checks, and QC Samples	Laboratory Technician/ Laboratory QA Manager	Laboratory PM/Delegated QA Manager	Laboratory Data Package and Project File
Sample Disposal Records	Laboratory Technician	Laboratory PM/Delegated QA Manager	Laboratory File

Laboratory Data Deliverables ¹				
Record	Mercury			
Narrative	Y			
Chain of Custody	Y			
Summary Results	Y			
QC Results	Y			
Chromatograms or raw data	Y			
Tentatively Identified Compounds	Y			

¹ The blank Laboratory Data Deliverables table is designed to be a checklist for use in supporting data completeness. The records and analytical groups in this table are not all inclusive of those that may be used on a specific project and should be modified and utilized by the Delegated SPM as applicable

QAPP Worksheet #31, 32 & 33: Assessments and Corrective Action (Non-CLP Worksheet)

Information in this worksheet is program-specific and is incorporated by reference into the site-specific QAPP. All reports will be prepared by WESTON and distributed to the following, to include but not be limited to, the WESTON Operational Manager, Program Manager, and Chemist QA/QC Specialist; and the EPA OSC, PO, TM, and QA Manager as applicable.

Assessments:

Assessment Type	Responsible Party & Organization	Number/ Frequency	Estimated Dates	Assessment Deliverable	Deliverable Due Date
Field Sampling Technical Systems Audit (TSA) ¹	Chemist QA/QC Specialist (or designee) and Group Leader or Operational Manager WESTON	As needed, as determined by WESTON Chemist QA/QC Specialist (or designee) and Group Leader or Program Manager WESTON	To be completed near the beginning of field sample collection activities/TBD	TSA Memorandum and Checklist	1 day
Laboratory TSA ²	Laboratory QA Manager Regulatory Agency	CLP, DESA, and certified subcontract laboratories are routinely audited by accrediting authorities.	Every Year	Written Report	1 week
Data Validation	Data Validator WESTON	Each data package for which data validation was requested; varies by site	TBD	Data Validation Report	1 week
Management Review	Group Leader and/or Operational Manager WESTON	Varies; as determined by WESTON Program Manager	TBD	Quality Management Report (memo/e-mail to file)	1-2 weeks following assessment

¹ Field sampling TSAs may include, but are not limited to the following: sample collection records; sample handling, preservation, packaging, shipping, and custody records; equipment operation, maintenance, and calibration records.

² Laboratory TSAs may include, but are not limited to the following: sample log-in, identification, storage, tracking, and custody procedures; sample and standards preparation procedures; availability of analytical instruments; analytical instrument operation, maintenance, and calibration records; laboratory security procedures; qualifications of analysts; case file organization and data handling procedures.

Worksheet 31, 32 & 33 — Assessments and Corrective Action (Concluded)

Assessment Response and Corrective Action:

Assessment Type	Responsibility for Responding to Assessment Findings	Assessment Response Documentation	Timeframe for Response	Responsibility for Implementing Corrective Action	Responsible for Monitoring Corrective Action Implementation
Field Sampling Technical Systems Audit (TSA) ¹	SPM, WESTON	Findings of field audit.	24 hours of receipt of audit report	Operational Manager WESTON	SPM or Operational Manager, WESTON
Laboratory TSA ²	Laboratory QA Manager Chemist QA/QC Specialist (or designee) WESTON	Written response to EPA Region II to address deficiencies	1 week of receipt of request from EPA Region II (or RST on behalf of EPA)	Laboratory Manager	Quality Manager (or designee) and/or Chemist, WESTON
Project-Specific PT/PE Samples	Laboratory QA Manager	Written response to EPA Region II to address deficiencies for failing scores on PE samples	1 week of receipt of request from EPA Region II (or RST on behalf of EPA)	Laboratory Manager	Laboratory QA Manager
Data Validation	Chemist QA/QC Specialist (or designee) WESTON	Validation Report	Within 48 hours of receipt of validation inquiry	Laboratory QA Manager and/or Chemist	Chemist, WESTON
Management Review	Program Manager, WESTON	Quality Management Response	48 hours of receipt of Quality Management report	Program Manager WESTON	Chemist QA/QC Specialist (or designee) and Program Manager WESTON

¹ Field sampling TSAs may include, but are not limited to the following: sample collection records; sample handling, preservation, packaging, shipping, and custody records; equipment operation, maintenance, and calibration records.

² Laboratory TSAs may include, but are not limited to the following: sample log-in, identification, storage, tracking, and custody procedures; sample and standards preparation procedures; availability of analytical instruments; analytical instrument operation, maintenance, and calibration records; laboratory security procedures; qualifications of analysts; case file organization and data handling procedures.

QAPP Worksheet #34: Data Verification and Validation Inputs

Item	Description	Verification (completeness)	Validation (conformance to specifications)
Planning Documents/Records			
1	Approved QAPP	X	
2	Contract	X	
3	Field SOPs	X	
4	Laboratory SOPs	X	
5	Laboratory QA Manual	NA	
6	Laboratory Certifications	X	
Field Records			
7	Field Logbooks	X	X
8	Equipment Calibration Records	X	X
9	Chain of Custody Forms	X	X
10	Sampling Diagrams/Surveys	X	X
11	Drilling Logs	NA	NA
12	Geophysics Reports	NA	NA
13	Relevant Correspondence	X	X
14	Change Orders/Deviations	X	X
15	Field Audit Reports	X	X
16	Field Corrective Action Reports	X	X
17	Sample Location Verification (Worksheet 18)	X	X
Analytical Data Package and Other Laboratory Deliverables			
18	Cover Sheet (laboratory identifying information)	X	X
19	Case Narrative	X	X
20	Internal Laboratory Chain of Custody	X	X
21	Sample Receipt Records	X	X
22	Sample Chronology (i.e. dates and times of receipt, preparation, & analysis)	X	X
23	Communication Records	X	X
24	Project-specific PT Sample Results	NA	NA
25	RL/MDL Establishment and Verification	X	X
26	Standards Traceability	NA	NA
27	Instrument Calibration Records	X	X
28	Definition of Laboratory Qualifiers	X	X
29	Results Reporting Forms	X	X
30	QC Sample Results	X	X
31	Corrective Action Reports	X	X
32	Raw Data	X	X
33	Electronic Data Deliverable	X	X

QAPP Worksheet #35: Data Verification Procedures

The following information includes program-specific and project-specific documents which may be incorporated by reference in the site-specific QAPP. Inputs may include, but are not limited to, those identified in the table below.

Records Reviewed	Required Documents	Process Description	Responsible Person, Organization
Contract QAPP	Contract, EPA and UFP-QAPP Guidance documents	Verify completeness, correctness, and contractual compliance of all program QA/QC against the methods, SOPs, and contract requirements.	Timothy Benton WESTON Program Manager Smita Sumbaly, WESTON Chemist QA/QC Specialist
Site-specific QAPP	Contract QAPP, Work Scope in TDD	Verify sampling and analytical methods specified in site-specific QAPP are correct and all contract QAPP protocols are followed and required QC samples will be collected in the correct bottles and properly preserved.	Bernard Nwosu WESTON Operational Manager Smita Sumbaly, WESTON Chemist QA/QC Specialist
Field Logs and SOPs	Contract and site-specific QAPP, SOPs	Ensure that all field sampling SOPs specified in site-specific QAPP were followed.	WESTON SPM and Data Validation Personnel
Analytical SOPs	Analytical Method and Contract QAPP	Ensure that laboratory analytical SOPs comply with the published method.	Laboratory QA Manager, Smita Sumbaly, WESTON Chemist QA/QC Specialist /Data validation Personnel
Laboratory QA Manual	EPA Guidance Documents	Verify that best practices specified in EPA Guidance Documents are incorporated into the Laboratory QA Manual.	Laboratory QA Manager
Laboratory Certifications	Generic and site-specific QAPP	Ensure that laboratory performing analytical sample analyses has current State, National Environmental Laboratory Accreditation Program, National Voluntary Laboratory Accreditation Program, or American Industrial Hygiene Association certifications as required by the project.	Laboratory PM, Smita Sumbaly, WESTON Chemist QA/QC Specialist
Laboratory Deliverables	Contract and site-specific QAPP	Verify that the laboratory deliverable contains all records specified in the contract QAPP. Check sample receipt records to ensure sample condition upon receipt was noted, and any missing/broken sample containers were noted and reported. Compare the data package with Chains of custody to verify that results were provided for all collected samples. Review the narrative to ensure all QC exceptions are described. If Stage 2B or higher validation is required, verify that analytical instrumentation met calibration requirements. Check for evidence that any required notifications were provided to project personnel. Verify that necessary signatures and dates are present.	WESTON Data Validation Personnel: Smita Sumbaly, Chemist QA/QC Specialist

* Site-specific QAPP may contain additional data validation inputs as required by the project objectives.

QAPP Worksheet #35: Data Verification Procedures (Concluded)

Records Reviewed	Required Documents	Process Description	Responsible Person, Organization
WESTON Data Validation Deliverables	Laboratory Report, Analytical Method and Laboratory SOPs	Verify that the report consists of the following for all field samples submitted to the laboratory: 1) Data validation report (pdf), 2) Sample Summary Report with data validation qualifiers, and 3) Excel EDD file with data validation qualifiers	Weston Data Validator: Smita Sumbaly, Chemist QA/QC Specialist
Field Logbook, Field Sheets, Sample Diagrams/ Surveys	Contract and site-specific QAPP	Verify that records are present and complete for each day of field activities. Verify that all planned samples including field QC samples were collected and that sample collection locations are documented. Verify that meteorological data were provided for each day of field activities. Verify that changes/exceptions are documented and were reported in accordance with requirements. Verify that any required field monitoring was performed and results are documented.	WESTON SPM and Operational Manager
Field Equipment Calibration Records	Contract and site-specific QAPP, SOPs, field logbook	Ensure that all field analytical instrumentation SOPs for equipment calibration were followed.	WESTON SPM and Operational Manager
Chain of Custody Forms	Site-specific QAPP; Field Logbook; and other sampling records (e.g., boring logs, etc.)	Verify the completeness of Chain-of-Custody records. Examine entries for consistency with the field logbook. Check that appropriate methods were requested and sample preservation was recorded. Verify that the required volume of sample has been collected and that sufficient sample volume is available for Laboratory QC samples (e.g., MS/MSD and S/D). Verify that all required signatures and dates are present. Check for transcription errors.	WESTON SPM, WESTON Chemist QA/QC Specialist, and Laboratory PM
Relevant reports and correspondence	Contract and site-specific QAPP	Verify that reports are present and complete for each day of field activities. Verify that correspondence is documented and was reported in accordance with requirements.	WESTON Operational Manager and SPM
Audit Reports, Corrective Action Reports	Generic and site-specific QAPP	Verify that all planned audits were conducted. Examine audit reports. For any deficiencies noted, verify that corrective action was implemented according to plan.	Weston Data Validator: Smita Sumbaly, Chemist QA/QC Specialist

QAPP Worksheet #36: Data Validation Procedures

The following information is project-specific and will be identified in the site-specific or QAPP.

Data Validator: RST 3 Data Validation Personnel

Analytical Group/Method	Data Deliverable Requirements	Analytical Specifications	MPC	Percent of Data Packages to be Validated	Percent of Raw Data Reviewed	Percent of Results to be Recalculated	Validation Procedure	Validation Code	Electronic Validation Program/Version
Mercury / NIOSH 6009	Definitive Data	SEDD Stage IIa	Worksheets 12, 24, 28	100%	100%	10%	Analytical Method NIOSH 6009 & Lab SOP	NA	Excel EDD

NA – Not Applicable

QAPP Worksheet #37: Usability Assessment

Data usability assessments will be performed as directed by EPA. This worksheet documents procedures that will be used to perform the data usability assessment (DUA). The DUA is performed at the conclusion of data collection activities using the outputs from data verification and data validation (i.e., data of known and documented quality). It is the data interpretation phase, which involves a qualitative and quantitative evaluation of environmental data to determine whether the Site data are of the right type, quality, and quantity to support the decisions that need to be made. It involves a retrospective evaluation of the systematic planning process, and involves participation by key members of the project team. The DUA evaluates whether underlying assumptions used during systematic planning are supported, sources of uncertainty have been accounted for and are acceptable, data are representative of the population of interest, and the results can be used as intended, with the acceptable level of confidence.

Data, whether generated in the field or by the laboratory, are tabulated and reviewed for Precision, Accuracy, Representativeness, Completeness, and Comparability (PARCCS) by the SPM for field data or the data validator for laboratory data. The review of the PARCC Data Quality Indicators (DQI) will compare with the Data Quality Objectives (DQO) detailed in the site-specific QAPP, the analytical methods used and impact of any qualitative and quantitative trends will be examined to determine if bias exists. A hard copy of field data is maintained in a designated field or site logbook. Laboratory data packages are validated, and final data reports are generated. All documents and logbooks are assigned unique and specific control numbers to allow tracking and management.

Where applicable, the following documents will be followed to evaluate data for fitness in decision making: EPA QA/G-4, Guidance on Systematic Planning using the Data Quality Objectives Process, EPA/240/B-06/001, February 2006, and EPA QA/G-9R, Guidance for Data Quality Assessment, A reviewer's Guide EPA/240/B-06/002, February 2006.

Personnel (organization and position/title) responsible for participating in the data usability assessment may include, but not be limited to:

- RST 3 Operational Manager;
- RST 3 Quality Manager (or designee);
- RST 3 Risk Assessor;
- RST 3 SPM;
- RST 3 Chemist QA/QC Specialist;
- RST 3 Statistician.

Based on project-specific oversight responsibilities and analytical scopes, this DUA worksheet outlines the approach that will be taken as the analytical scope expands on a project-specific basis.

The following general steps will be followed to assure that the data usability assessment evaluates whether underlying assumptions used during systematic planning are supported, sources of uncertainty have been accounted for and are acceptable, data are representative of the population of interest, and the results can be used as intended, with the acceptable level of confidence:

QAPP Worksheet #37: Usability Assessment (Concluded)

Step 1 – Review the project’s objectives and sampling design: This includes reviewing the DQOs and MPC to make sure they are still applicable. The sampling design will be consistent with stated DQOs.

Step 2 – Review the data verification and data validation outputs: Graphs, maps, and tables can be prepared to summarize the data. Deviations from activities planned in the Project QAPP should be considered, including samples not collected (potential data gaps), holding time exceedances, damaged samples, impact of non-compliant PE sample results, and SOP deviations. The implications of unacceptable QC sample results will be assessed.

Step 3 – Verify the assumptions of the selected statistical method: The underlying assumptions for the selected statistical methods (if specified in the QAPP) will be verified for validity. Common assumptions include the distributional form of the data, independence of the data, dispersion characteristics, homogeneity, etc. Depending on the robustness of the statistical method, minor deviations from assumptions usually are not critical to statistical analysis and data interpretation. If serious deviations from assumptions are discovered, then another statistical method may be selected.

Step 4 - Implement the statistical method: If specified in the site-specific QAPP, statistical procedures will be implemented for analyzing the data and reviewing underlying assumptions. For a decision project that involves hypothesis testing (e.g., “concentrations of lead in groundwater are below the action level”) the consequences of selecting the incorrect alternative will be considered; for estimation projects (e.g., establishing a boundary for surface soil contamination), the tolerance for uncertainty in measurements will be considered.

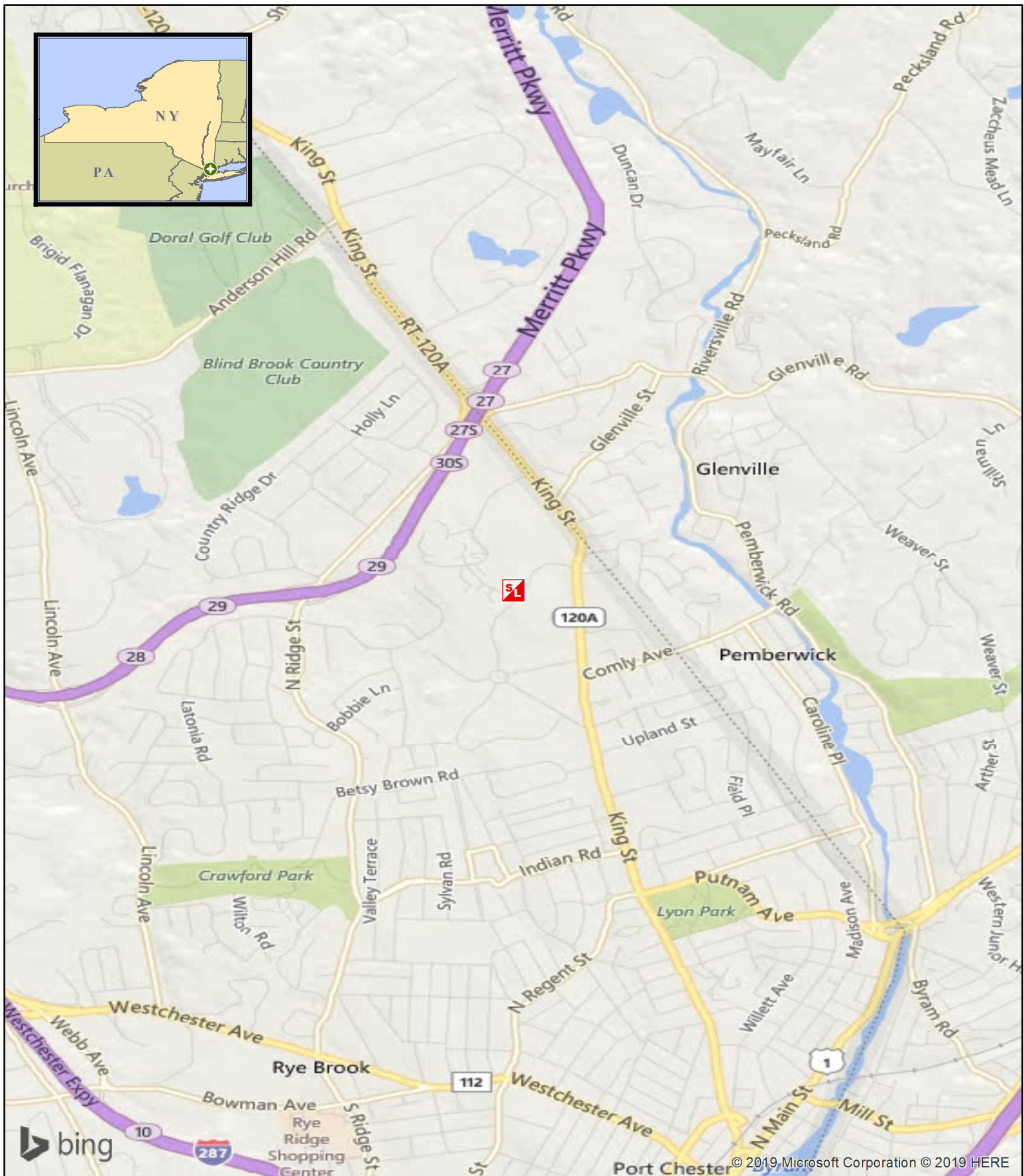
Step 5 – Document data usability and draw conclusions:

The DUA considered the final step in the data evaluation process. All data will be assessed for usability regardless of data evaluation/validation process implementation. Data usability goes beyond validation in that it evaluates the achievement of the DQOs based on the comparison of the project DQIs and site-specific QAPP with the obtained results. The results of the DUA, and particularly any changes to the DQOs necessitated by the data not meeting usability criteria, will be communicated in accordance with Worksheet 6.

The usability of the data as intended will be determined. Achievable DQOs, based on comparison with the Site DQIs, will be discussed. The performance of the sampling design will be assessed and limitations of the data use identified. The conceptual site model will be updated and conclusions documented. A DUA report (in the form of text/or table) will be prepared or a data usability summary will be included in the final report.

ATTACHMENT A

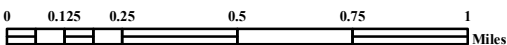
Figure 1: Site Location Map



Legend



Site Location



Weston Solutions, Inc.
East Division

In Association With
Scientific and Environmental Associates, Inc.,
Environmental Compliance Consultants, Inc.,
Avatar Environmental, LLC, On-Site Environmental,
Inc. and Sovereign Consulting, Inc

Figure 1:
Site Location Map

Port Refinery
Rye Brook, NY

U.S. ENVIRONMENTAL PROTECTION AGENCY
REMOVAL SUPPORT TEAM 3
CONTRACT # EP-S2-14-01

GIS ANALYST:	M. LANG
EPA OSC:	G. BUSHRA
RST SPM:	M. LANG
CHARGE #:	TO-0370-0117

DATE MODIFIED: 2/4/2019

ATTACHMENT B

Sampling SOPs

ERT SOP# 2001: *General Field Sampling Guidelines*

ERT SOP# 2008: *General Air Sampling Guidelines*



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GENERAL FIELD SAMPLING GUIDELINES

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Complete Rewrite: SOP #2001; Revision 1.0; 03/15/13; U.S. EPA Contract EP-W-09-031

SUPERCEDES: SOP #2001; Revision 0.0; 08/11/94; U.S. EPA Contract 68-C4-0022



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1.0 OBJECTIVE

The objective of this standard operating procedure (SOP) is to describe the general field sampling techniques and guidelines that will assist the Scientific Engineering Response and Analytical Services (SERAS) personnel in planning, choosing sampling strategies and sampling locations, and frequency of Quality Control (QC) samples for proper assessment of site characteristics. The ultimate goal is to ensure data quality during field collection activities.

2.0 APPLICABILITY

This SOP applies to the collection of aqueous and non-aqueous samples for subsequent laboratory analysis to determine the presence, type, and extent of contamination at a site.

3.0 DESCRIPTION

Representative sampling ensures that a sample or a group of samples accurately reflect the concentration of the contaminant at a given time and location. Depending on the contaminant of concern and matrix, several variables may affect the representativeness of the samples and subsequent measurements. Environmental variability due to non-uniform distribution of the pollutant due to topographic, meteorological and hydrogeological factors, changes in species, and dispersion of contaminants and flow rates contribute to uncertainties in sampling design.

Determining the sampling approach depends on what is known about the site from prior sampling (if any) and the site history, variation of the contaminant concentrations throughout a site, potential migration pathways, and human and environmental receptors. The objectives of an investigation determine the appropriate sampling design.

The frequency of sampling and the specific sample locations that are required must be defined in the site-specific Quality Assurance Project Plan (QAPP).

3.1 Planning Stage

The objectives of an investigation are established and documented in the site-specific QAPP. The technical approach including the media/matrix to be sampled, sampling equipment to be used, sampling design and rationale, and SOPs or descriptions of the procedure to be implemented are included in the QAPP. Refer to the matrix-specific SOPs for sampling techniques which include the equipment required for sampling.

During the planning stage, the data quality objectives (DQOs) will be determined. In turn, the project's DQOs will determine the need for screening data or definitive data. Screening data supports an intermediate or preliminary decision but eventually is supported by definitive data before the project is complete (i.e., placement of monitor wells, estimation of extent of contamination). Definitive data is suitable for final decision making, has defined precision and accuracy requirements and is legally defensible (i.e., risk assessments, site closures).

3.2. Sampling Design

Representative sampling approaches include judgmental, random, systematic grid, systematic simple random, stratified random and transect sampling. Sampling designs may be applied to soil,



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sediment and water; however, the random and systematic random approaches are not practical for sampling water systems, especially flowing water systems.

3.2.1 Judgmental Sampling

Judgmental sampling is the subjective selection of sampling locations based on the professional judgment of the field team. This method is useful to locate and to identify potential sources of contamination. It may not be representative of the full site and is used to document worst case scenarios. For example, groundwater sampling points are typically chosen based on professional judgment, whether permanently installed wells or temporary well points.

3.2.2 Systematic Sampling

Systematic grid sampling involves the collection of samples at fixed intervals when the contamination is assumed to be randomly distributed. A random point is chosen as the origin for the placement of the grid. A grid is constructed over a site and samples are collected from the nodes (where the grid lines intersect). Depending on the number of samples that are required to be collected, the distance between the sampling locations can be adjusted. The representativeness of the sampling may be improved by shortening the distance between sample locations.

Systematic random sampling is used for estimating contaminant concentrations within grid cells. Instead of sampling at each node, a random location is chosen within each grid cell. The systematic grid and random sampling approaches are useful for delineating the extent of contamination, documenting the attainment of clean-up goals, and evaluating and determining treatment and disposal options.

Transect sampling involves one or more transect lines established across the site. Samples are collected at systematic intervals along the transect lines. The number of samples to be collected and the length of the transect line determines the spacing between the sampling points. This type of sampling design is useful for delineating the extent of contamination at a particular site, for documenting the attainment of clean-up goals, and for evaluating and determining treatment and disposal options.

3.2.3 Simple and Stratified Random Sampling

Statistical random sampling includes simple, stratified and systematic sampling. Simple random sampling is appropriate for estimating means and total concentrations, if the site or population does not contain a major trend or pattern of contamination. A statistician will generate the sampling locations based on sound statistical methods. Stratified random sampling is a useful tool for estimating average contaminant concentrations and total amounts of contaminants within specified strata and across the entire site. It is useful when a heterogeneous population or area can be broken down into regions with less variability within the boundaries of a stratum than between the strata. Additionally, strata can be defined based on the decisions that will be made. This type of sampling design uses historical information, known ecological and human receptors, soil type, fate and transport mechanism and other ecological factors to divide the sampling area into smaller regions or strata. Sampling locations are selected from each stratum using random sampling.



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The simple random sampling approach is applied when there are many sample locations and the concentrations are assumed to be homogeneous across a site with respect to the parameter(s) that are going to be analyzed or monitored for. The stratified random sampling approach is useful for sampling drums, evaluating and determining treatment and disposal options, and locating and identifying sources of contamination.

3.3 Sampling Techniques

Sampling is the selection of a representative portion of a larger population or body. The primary objective of all sampling activities is to characterize a site accurately in a way that the impact on human health and the environment can be evaluated appropriately.

3.3.1 Sample Collection Techniques

Sample collection techniques may be either grab or composite. A grab sample is a discrete aliquot representative of a specific location at a given time and collected all at once from one location. The representativeness of such samples is defined by the nature of the materials that are sampled. Samples collected for volatile organic compounds (VOCs) are always grab samples and are never homogenized. Composite samples are non-discrete samples composed of more than one specific aliquot collected at selected sampling locations. Composite samples must be homogenized by mixing prior to putting the sample into containers. Composite samples can, in certain instances, be used as an alternative to analyzing a number of individual grab samples and calculating an average value. Incremental sampling conducted over a grid is a special case of composite sampling and is detailed in SOP #2019, *Incremental Soil Sampling*. Choice of collecting discrete or composite samples is based on project's DQOs.

3.3.2 Homogenization

Mixing of soil and sediment samples is critical to obtain a representative sample. An adequate volume/weight of sample is collected and placed in a stainless steel or Teflon[®] container, and is thoroughly mixed using a spatula or spoon made of an inert material. Once the sample is thoroughly mixed the sample is placed into sample containers specific for an analysis. Avoid the use of equipment made of plastic or polyvinyl chloride (PVC) when sampling for organic compounds when the reporting limit (RL) is in the parts per billion (ppb) or parts per trillion (ppt) ranges. Refer to SERAS SOP #2012, *Soil Sampling*, for more details on homogenization.

3.3.3 Filtration

In-line filters are used specifically for collecting groundwater samples for dissolved metals analysis and for filtering large volumes of turbid groundwater. Groundwater samples collected for VOCs are typically not filtered due to potential VOC losses. Filtering groundwater is performed to remove silt particulates from samples to prevent interference with the laboratory analysis. The filters used in groundwater sampling are either cartridge type filters inserted into a reusable housing, or are self-contained and disposable. Filter chambers are usually made of polypropylene housing an inert filtering material that removes particles larger than 0.45 micrometers (μm). Refer to SERAS SOP



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#2007, *Groundwater Well Sampling* and SERAS SOP #2013, *Surface Water Sampling*, for more details on filtration techniques.

3.4 Quality Assurance /Quality Control Samples

QA/QC samples provide an evaluation of both the laboratory's and the field sampling team's performance. Including QA/QC samples in a sampling design allows for identifying and measuring sources of error potentially introduced from the time of sample container preparation through analysis. The most common QA/QC samples collected in the field are collocated field duplicates, field replicates, equipment blanks, field blanks and trip blanks. Extra volume/mass is collected for a matrix spike/matrix spike duplicate (MS/MSD) at a frequency of 5% (one in 20 samples). Spiking is performed in the laboratory. For additional information or other QA/QC samples pertinent to sample analysis, refer to SERAS SOP #2005, *Quality Assurance/Quality Control Samples*.

Collocated field duplicates may be collected based on site objectives and used to measure variability associated with the sampling process including sample heterogeneity, sampling methodology, and analytical procedures. Field replicates are field samples obtained from one location, homogenized, and divided into separate containers. This is useful for determining whether the sample has been homogenized properly. Equipment blanks (also known as rinsate blanks) are typically collected at a rate of one per day. The equipment blank is used to evaluate the relative cleanliness of non-dedicated equipment.

3.5 Sample Containers, Preservation, Storage and Holding Times

The amount of sample to be collected, the proper sample container type (i.e., glass, plastic), chemical preservation, and storage requirements are dependent on the matrix sampled and the analyses to be conducted. This information is provided in SERAS SOP #2003, *Sample Storage, Preservation, and Handling*. Field personnel need to be cognizant of any short holding times that warrant immediate shipment/transfer to the laboratory.

3.6 Documentation

Field conditions and site activities must be documented. Scribe will be used to document sample locations and generate chain of custody records. Other field measurements not typically entered into Scribe will be documented in a site-specific logbook or in a personal logbook. All sample documentation will be maintained in accordance with SERAS SOP #2002, *Sample Documentation* and SERAS SOP #4005, *Chain of Custody Procedures*.

4.0 RESPONSIBILITIES

4.1 SERAS Task Leaders

Task Leaders (TLs) are responsible for the overall management of the project. Task Leader responsibilities include ensuring that field personnel are well informed of the sampling requirements for a specific project and that SOP and QA/QC procedures stated in the site-specific QAPP are adhered to, issuing a Field Change Form that documents any changes to sampling activities after the QAPP has been approved and maintaining sample documentation.



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4.2 SERAS Field Personnel

Field personnel are responsible for reading the QAPP prior to site activities and performing sample collection activities as written. They are responsible for notifying the TL of deviations from sample collection protocols which occurred during the execution of sampling activities. Field staff will collect samples and prepare documentation in accordance with SERAS SOP #2002, *Sample Documentation*. In addition, field personnel are responsible for reading and conforming to the approved site-specific Health and Safety Plan (HASP).

4.3 SERAS Program Manager

The SERAS Program Manager is responsible for the overall technical and financial management of the project.

4.4 SERAS QA/QC Officer

The QA/QC Officer is responsible for reviewing this SOP and ensuring that the information in this SOP is updated on a timely basis. Compliance to this SOP may be monitored by either conducting a field audit or reviewing deliverables prepared by the SERAS TL.

4.5 Health and Safety (H&S) Officer

The H&S Officer is responsible for ensuring that a HASP has been written in conformance with SOP # 3012, *SERAS Health and Safety Guidelines for Field Activities* and approved prior to field activities. Additionally, the H&S Officer is responsible for ensuring that SERAS site personnel's H&S training is current as per SOP # 3006, *SERAS Field Certification Program* and that their medical monitoring is current as per *SERAS SOP #3004, SERAS Medical Monitoring Program*.

the 1990s, the number of people with a mental health problem has increased by 50% (Mental Health Foundation 2000).

There is a growing awareness of the need to address the needs of people with mental health problems in the community. The Department of Health (2000) has set out a vision for the future of mental health services, which includes a focus on preventing mental health problems, supporting people with mental health problems in the community, and providing specialist services for people with severe mental health problems.

The Department of Health (2000) has also set out a number of key principles for the future of mental health services, which include: a focus on prevention, a focus on supporting people with mental health problems in the community, and a focus on providing specialist services for people with severe mental health problems.

The Department of Health (2000) has also set out a number of key objectives for the future of mental health services, which include: to reduce the incidence of mental health problems, to improve the quality of life for people with mental health problems, and to ensure that people with mental health problems have access to the services they need.

The Department of Health (2000) has also set out a number of key strategies for the future of mental health services, which include: to develop a culture of prevention, to develop a culture of support, and to develop a culture of specialist services.

The Department of Health (2000) has also set out a number of key actions for the future of mental health services, which include: to develop a culture of prevention, to develop a culture of support, and to develop a culture of specialist services.

The Department of Health (2000) has also set out a number of key outcomes for the future of mental health services, which include: to reduce the incidence of mental health problems, to improve the quality of life for people with mental health problems, and to ensure that people with mental health problems have access to the services they need.

The Department of Health (2000) has also set out a number of key indicators for the future of mental health services, which include: to reduce the incidence of mental health problems, to improve the quality of life for people with mental health problems, and to ensure that people with mental health problems have access to the services they need.

The Department of Health (2000) has also set out a number of key challenges for the future of mental health services, which include: to develop a culture of prevention, to develop a culture of support, and to develop a culture of specialist services.

The Department of Health (2000) has also set out a number of key opportunities for the future of mental health services, which include: to develop a culture of prevention, to develop a culture of support, and to develop a culture of specialist services.

The Department of Health (2000) has also set out a number of key risks for the future of mental health services, which include: to develop a culture of prevention, to develop a culture of support, and to develop a culture of specialist services.

The Department of Health (2000) has also set out a number of key strengths for the future of mental health services, which include: to develop a culture of prevention, to develop a culture of support, and to develop a culture of specialist services.

The Department of Health (2000) has also set out a number of key weaknesses for the future of mental health services, which include: to develop a culture of prevention, to develop a culture of support, and to develop a culture of specialist services.

The Department of Health (2000) has also set out a number of key threats for the future of mental health services, which include: to develop a culture of prevention, to develop a culture of support, and to develop a culture of specialist services.

The Department of Health (2000) has also set out a number of key opportunities for the future of mental health services, which include: to develop a culture of prevention, to develop a culture of support, and to develop a culture of specialist services.



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1.0 SCOPE AND APPLICATION

This Standard Operating Procedure (SOP) provides guidance in developing and implementing sampling plans to assess the impact of hazardous waste sites on ambient air. It presents the United States Environmental Protection Agency/Environmental Response Team's (U.S. EPA/ERT's) approach to air sampling and monitoring and identifies equipment requirements. It is not within the scope of this SOP to provide a generic air sampling plan. Experience, objectives, site characteristics, and chemical characteristics will dictate sampling strategy. This SOP does not address indoor air sampling.

Two basic approaches can be used to assess ambient air (also referred to as air pathway assessments): modeling and measurements. The modeling approach initially estimates or measures the overall site emission rate(s) and pattern(s). These data are input into an appropriate air dispersion model, which predicts either the maximum or average air concentrations at selected locations or distances during the time period of concern. This overall modeling strategy is presented in the first three volumes of the Air Superfund National Technical Guidance Series on Air Pathway Assessments^(1,2,3). Specific applications of this strategy are presented in several additional Air Superfund Technical Guidance documents⁽⁴⁾.

The measurement approach involves actually measuring the air impact at selected locations during specific time periods. These measurements can be used to document actual air impacts during specific time intervals (i.e., during cleanup operations) or to extrapolate the probable "worst case" concentrations at that and similar locations over a longer time period than was sampled.

This SOP addresses issues associated with this second assessment strategy. This SOP also discusses the U.S. EPA/ERT's monitoring instruments, air sampling kits, and approach to air sampling and monitoring at hazardous waste sites.

These are standard (i.e., typically applicable) operating procedures which may be varied or changed as required, depending on site conditions, equipment limitations, or limitations imposed by the procedure. In all instances, the ultimate procedures employed should be documented and associated with the final report.

Mention of trade names or commercial products does not constitute U.S. EPA endorsement or recommendation for use.

2.0 METHOD SUMMARY

Air monitoring is defined as the use of direct-reading instruments and other screening or monitoring equipment and techniques that provide instantaneous (real-time) data on the levels of airborne contaminants. The U.S. EPA/ERT maintains numerous monitors for real-time measurements. Examples of air monitoring equipment are hand-held photoionization detectors (PID), flame ionization detectors (FID), oxygen/combustible gas detectors, and remote optical sensors.

Air sampling is defined as those sampling and analytical techniques that require either off- or on-site laboratory analysis and therefore do not provide immediate results. Typically, air sampling occurs after use of real-time air monitoring equipment has narrowed the number of possible contaminants and has provided some qualitative measurement of contaminant concentration. Air sampling techniques are used to more accurately detect, identify and quantify specific chemical compounds relative to the majority of air monitoring technologies.



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In the Superfund Removal Program, On-Scene Coordinators (OSCs) may request the U.S. EPA/ERT to conduct air monitoring and sampling during the following situations: emergency responses, site assessments, and removal activities. Each of these activities has a related air monitoring/sampling objective that is used to determine the potential hazards to workers and/or the community.

- Emergency Response

Emergency responses are immediate responses to a release or threatened release of hazardous substances presenting an imminent danger to public health, welfare, or the environment (i.e., chemical spills, fires, or chemical process failures resulting in a controlled release of hazardous substances). Generally these situations require rapid on-site investigation and response. A major part of this investigation consists of assessing the air impact of these releases.

- Removal Site Assessment

Removal site assessments (referred to as site assessments) are defined as any of several activities undertaken to determine the extent of contamination at a site and which help to formulate the appropriate response to a release or threatened release of hazardous substances. These activities may include a site inspection, multimedia sampling, and other data collection.

- Removal Actions

Removal actions clean up or remove hazardous substances released into the environment. Removal actions include any activity conducted to abate, prevent, minimize, stabilize, or eliminate a threat to public health or welfare, or to the environment.

Personal risk from airborne contaminants can be determined by comparing the results of on-site monitoring and sampling to health-based action levels such as the American Conference of Governmental Industrial Hygienists (ACGIH) Threshold Limit Values (TLVs) and the Occupational Safety and Health Administration (OSHA) Permissible Exposure Limits (PELs). Residential risk can be determined by comparing the results of off-site monitoring or sampling to health-based action levels such as those developed by the Agency for Toxic Substance and Disease Registry (ATSDR).

The extent to which valid inferences can be drawn from air monitoring/sampling depends on the degree to which the monitoring/sampling effort conforms to the objectives of the event. Meeting the project's objectives requires thorough planning of the monitoring/sampling activities, and implementation of the most appropriate monitoring/sampling and analytical procedures. These issues will be discussed in this SOP.

3.0 SAMPLE PRESERVATION, CONTAINERS, HANDLING, AND STORAGE

Preservation, containers, handling and storage for air samples are discussed in the specific SOPs for the technique selected. In addition, the analytical method (i.e., U.S. EPA, National Institute for Occupational Safety and Health [NIOSH], and OSHA Methods) may be consulted for storage temperature, holding times and packaging requirements. After sample collection, the sampling media (i.e., cassettes or tubes) are immediately sealed. The samples are then placed into suitable containers (i.e., whirl bags, resealable bags or culture tubes) which are then placed into a shipping container.



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Use bubble wrap or Styrofoam peanuts when packing air samples for shipment. DO NOT USE VERMICULITE.

4.0 INTERFERENCES AND POTENTIAL PROBLEMS

Upwind sources can contribute to sample concentration. Natural sources, such as biological waste, can produce hydrogen sulfide and methane which may contribute to the overall contaminant level. Extraneous anthropogenic contaminants (i.e., burning of fossil fuels; emissions from vehicular traffic, especially diesel; and volatile compounds) from petrochemical facilities; effluvium from smoke stacks) may also contribute. Air sampling stations should be strategically placed to identify contributing sources.

Photoreactivity or reaction of the parameters of concern may occur with nonrelated compounds [i.e., nitrogen compounds and polyaromatic hydrocarbons (PAHs)]. Some sorbent media/samples should not be exposed to light during or after sampling due to photochemical effects (i.e., PAHs).

Various environmental factors, including humidity, temperature and pressure, also impact the air sampling methodology, collection efficiency and detection limit. Since the determination of air contaminants is specifically dependent on the collection parameters and efficiencies, the collection procedure is an integral part of the analytical method.

Detection limits depend on the contaminants being investigated and the particular site situation. It is important to know why the data are needed and how the data will be used. Care should be taken to ensure the detection limits are adequate for the intended use of the final results.

Some equipment may be sensitive to humidity and temperature extremes.

5.0 EQUIPMENT/APPARATUS

5.1 Direct Reading Instruments (Air Monitoring Instruments)

There are two general types of direct reading instruments: portable screening devices and specialized analytical instruments. Generally all these techniques involve acquiring, for a specific location or area, continuous or sequential direct air concentrations in either a real-time or semi-real-time mode. None of these instruments acquires true time-weighted average concentrations. In addition, these instruments are not capable of acquiring simultaneous concentration readings at multiple locations, although several are able to sequentially analyze samples taken remotely from different locations. The document, "Guide to Portable Instruments for Assessing Airborne Pollutants Arising from Hazardous Waste Sites⁽⁵⁾," provides additional information about air sampling and monitoring. The hazard levels for airborne contaminants vary. See the ACGIH TLVs and the OSHA PELs for safe working levels. Common screening devices and analytical instruments are described in Appendix A.

5.2 Air Sampling Equipment and Media/Devices

The U.S. EPA/ERT uses the following analytical methods for sampling: *NIOSH Manual of Analytical Methods*⁽⁶⁾, *American Society for Testing and Materials (ASTM) Methods*⁽⁷⁾, *U.S. EPA Compendium of Methods for the Determination of Toxic Organic Compounds in Ambient Air*^(8,9), and *OSHA Methods*⁽¹⁰⁾.



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Additional air sampling references include *Industrial Hygiene and Toxicology* (3rd Ed.)⁽¹¹⁾ and *Air Sampling Instruments for Evaluation of Atmospheric Contaminants*⁽¹²⁾. These methods typically specify equipment requirements for sampling. Since air sampling is such a diverse technology, no single method or reference is best for all applications. Common sampling equipment and media/devices are described in Appendix B.

5.3 Tools/Material and Equipment List

In addition to equipment and materials identified in Appendices A and B, the following equipment and materials may be required to conduct air sampling and monitoring at hazardous waste sites:

- Camera
- Site logbook
- Clipboard
- Chain of custody records
- Custody seals
- Air sampling worksheets
- Sample labels
- Small screwdriver set
- Aluminum foil
- Extension cords
- Glass cracker
- Multiple plug outlet
- Whirl bags or culture tubes
- Teflon tape
- Calibration devices
- Tygon and/or Teflon tubing
- Surgical gloves
- Lint-free gloves
- Ice
- Sample container

Use the following additional equipment when decontaminating glassware on site:

- Protective equipment (i.e., gloves, splash goggles, etc.)
 - Appropriate solvent(s)
 - Spray bottles
 - Liquinox (soap)
 - Paper towels
 - Distilled/deionized water
 - Five-gallon buckets
 - Scrub brushes and bottle brushes

6.0 REAGENTS



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Impinger sampling involves using reagents contained in a glass vial to absorb contaminants of concern (for example, NIOSH Method 3500 for formaldehyde uses 1% sodium bisulfite solution). Impinger solutions vary and are method-dependent.

Reagents such as acetone and hexane are required to decontaminate glassware and some air sampling equipment. Decontamination solutions are specified in ERT/REAC SOP #2006, Sampling Equipment Decontamination.

7.0 PROCEDURES

7.1 Air Monitoring Design

7.1.1 Initial Surveys

In general, the initial survey is considered to be a relatively rapid screening process for collecting preliminary data at hazardous waste sites. However, initial surveys may require many hours to complete and may consist of more than one entry.

Some information is generally known about the site; therefore, real-time instrumentation for specific compounds (i.e., detector tubes and electrochemical sensors) can be used to identify hot spots. Sufficient data should be obtained with real-time instruments during the initial entry to screen the site for various contaminants. When warranted, intrinsically safe or explosion-proof instruments should be used. An organic vapor analyzer (OVA) is typically used during this survey. These gross measurements may be used on a preliminary basis to (1) determine levels

of personal protection, (2) establish site work zones, and (3) map candidate areas for more thorough qualitative and quantitative studies involving air sampling.

In some situations, the information obtained may be sufficient to preclude additional monitoring. Materials detected during the initial survey may call for a more comprehensive evaluation of hazards and analyses for specific compounds. Since site activities and weather conditions change, a continuous program to monitor the ambient atmosphere must be established.

7.1.2 Off-Site Monitoring

Typically, perimeter monitoring with the same instruments employed for on-site monitoring is utilized to determine site boundaries. Because air is a dynamic matrix, physical boundaries like property lines and fences do not necessarily delineate the site boundary or area influenced by a release. Whenever possible, atmospheric hazards in the areas adjacent to the on-site zone should be monitored with direct-reading instruments. Monitoring at the fenceline or at varying locations off site provides useful information regarding pollutant migration. Three to four locations downwind of the source (i.e., plume) at breathing-zone height provide a basic fingerprint of the plume. Negative instrument readings off site should not be interpreted as the complete absence of airborne toxic substances; rather, they should be considered another piece of information to assist in the preliminary evaluation. The interpretation of negative readings is instrument-



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dependent. The lack of instrument readings off site should not be interpreted as the complete absence of all airborne toxic substances; rather, it is possible that the particular compound or class of compounds to which the monitoring instrument responds is not present or that the concentration of the compound(s) is below the instrument's detection limit.

7.2 Air Sampling Design

7.2.1 Sampling Plan Design

The goal of air sampling is to accurately assess the impact of a contaminant source(s) on ambient air quality. This impact is expressed in terms of overall average and/or maximum air concentrations for the time period of concern and may be affected by the transport and release of pollutants from both on- and off-site sources. The location of these sources must be taken into account as they impact the selection of sampling locations. Unlike soil and groundwater concentrations, air concentrations at points of interest can easily vary by orders of magnitude over the period of concern. This variability plays a major role in designing an air sampling plan.

Downwind air concentration is determined by the amount of material being released from the site into the air (the emission rate) and by the degree that the contamination is diluted as it is transported. Local meteorology and topography govern downwind dilution. Contaminant emission rates can also be heavily influenced by on-site meteorology and on-site activities. All of these concerns must be incorporated into an air sampling plan.

A sampling strategy can be simple or complex, depending on the sampling program objectives. Programs involving characterization of the pollutant contribution from a single point source tend to be simple, whereas sampling programs investigating fate and transport characteristics of components from diverse sources require a more complex sampling strategy. In addition, resource constraints may affect the complexity of the sampling design.

An optimal sampling strategy accounts for the following site parameters:

- Location of stationary as well as mobile sources
- Analytes of concern
- Analytical detection limit to be achieved
- Rate of release and transport of pollutants from sources
- Availability of space and utilities for operating sampling equipment
- Meteorological monitoring data
- Meteorological conditions in which sampling is to be conducted

The sampling strategy typically requires that the concentration of contaminants at the source or area of concern as well as background contributions be quantified. It is important to establish background levels of contaminants in order to develop a reference point from which to evaluate the source data. Field blanks and lot blanks, as well as various other types of QA/QC samples, can be utilized to determine other sources. The impact of extraneous sources on sampling results can frequently be accounted for by



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placing samplers upwind, downwind and crosswind from the subject source. The analytical data from these different sampling locations may be compared to determine statistical differences.

7.2.2 Sampling Objectives

The objectives of the sampling must be determined prior to developing the sampling plan. Does the sampling plan verify adequate levels of protection for on-site personnel, or address potential off-site impacts associated with the site or with site activities? In addition, the assumptions associated with the sampling program must be defined. These assumptions include whether the sampling is to take place under "typical," "worst case", or "one-time" conditions. If the conditions present at the time of sampling are different from those assumed during the development of the sampling plan, then quality of the data collected may be affected. The following definitions have been established:

- Typical: routine daily sampling or routine scheduled sampling at pre-established locations.
- Worst case: sampling conducted under the worst meteorological and/or site conditions which would result in elevated ambient concentrations.
- One-time: only one chance is given to collect a sample without regard to time or conditions. Qualitative data acquired under these conditions are usually applicable only to the time period during which the data were collected and may not provide accurate information to be used in estimating the magnitude of an air impact during other periods or over a long time interval.

The sampling objectives also dictate the detection limits. Sampling methods for airborne contaminants will depend upon the nature and state (solid, liquid or gas) of the contaminant. Gases and vapors may be collected in aqueous media or adsorbents, in molecular sieves, or in suitable containers. Particulates are collected by filters or impactors. The volume of sample to be collected is dependent upon an estimate of the contaminant concentration in the air, the sensitivity of the analytical method, and the standard or desired detection limit. A sufficient amount of sample must be collected to achieve the desired detection limit without interference from other contaminants. In addition, the selected method must be able to detect the target compound(s).

7.2.3 Location and Number of Individual Sampling Points

Choose the number and location of sampling points according to the variability, or sensitivity, of the sampling and analytical methods being utilized, the variability of contaminant concentration over time at the site, the level of precision required and cost limitations. In addition, determine the number of locations and placement of samplers by considering the nature of the response, local terrain, meteorological conditions, location of the site (with respect to other conflicting background sources), size of the site, and the number, size, and relative proximity of separate on-site emission sources and upwind sources. The following are several considerations for sampler placement:



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- Location of potential on-site emission sources, as identified from the review of site background information or from preliminary on-site inspections.
- Location of potential off-site emission sources upwind of the sampling location(s). Review local wind patterns to determine the location of off-site sources relative to wind direction.
- Topographic features that affect the dispersion and transport of airborne toxic constituents. Avoid natural obstructions when choosing air sampling station locations, and account for channelization around those obstructions.
- Large water bodies, which affect atmospheric stability and the dispersion of air contaminants.
- Roadways (dirt or paved), which may generate dust that could mask site contaminants.
- Vegetation, such as trees and shrubs, which stabilizes soil and retards subsurface contaminants from becoming airborne. It also affects air flow and scrubs some contaminants from the air. Sometimes thick vegetation can make an otherwise ideal air monitoring location inaccessible.

Consider the duration of sampling activities when choosing the location and number of samples to be collected. For example, if the sampling period is limited to a few hours, one or two upwind and several downwind samples would typically be adequate, especially around major emission sources.

A short-term monitoring program ranges from several days to a few weeks and generally includes gathering data for site assessments, removal actions, and source determination data (for further modeling). Activities involved in a short-term sampling strategy must make the most of the limited possibilities for data collection. Consider moving upwind/downwind locations daily based on National Oceanic and Atmospheric Administration (NOAA) weather forecasts. Weather monitoring becomes critical where complex terrain and local meteorological effects frequently change wind direction. Often, a number of alternatives can fulfill the same objective.

Prevailing winds running the length of a valley usually require a minimum number of sampler locations; however, a complex valley may require more sampler locations to account for the wide variety of winds. Ocean/lake effects may require a radical plan to collect enough samples to reach a low detection limit. Two sets of samplers may be placed next to each other: one set would be activated during the sea breeze while the other set is turned off, and vice versa when there is no sea breeze. After the sampling event, the respective upwind and downwind samples would be combined. Another alternative for sampling near a large body of water may be to use automatic, wind-vector-operated samplers, which turn the sampler on only when the wind comes from a specified vector. At sites located on hillsides, wind will move down a valley and produce an upward fetch at the same time. Sampling locations may have to ring the site to measure



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the wind's impact.

Off-site sources may affect on-site monitoring. In this case, on-site meteorological data, concurrent with sampling data, is essential to interpreting the acquired data. Also, additional upwind sampling sites may be needed to fully characterize ambient background contaminant levels. Multiple off-site sources may require several monitoring locations, but if the sources are at a sufficient distance, only one monitoring location is needed.

Topography and weather are not the only factors in sampler location; the sampling sites must be secure from vandals and mishap. Secure all sampling locations to maintain chain of custody, and to prevent tampering with samples or loss of sampling units. High-volume sampling methods often require the use of 110 VAC electric power. When portable generators are used, the power quality may affect sampler operation. Also, be aware that the generators themselves could be a potential pollution source if their placement is not carefully considered.

Air quality dispersion models can be used to place samplers. The models incorporate source information, surrounding topography, and meteorological data to predict the general distance and directions of maximum ambient concentrations. Modeling results should be used to select sampling locations in areas of maximum pollutant concentrations.

7.2.4 Time, Duration and Frequency of Sampling Events

After choosing appropriate sampling or monitoring locations, determine the sampling frequency and the number of samples to be collected. The time of day, duration and frequency of sampling events is governed by:

- The effects of site activities and meteorology on emission rates
- The diurnal effect of the meteorology on downwind dispersion
- The time period(s) of concern as defined by the objective
- The variability in the impact from other non-site-related sources
- If defined, the degree of confidence needed for either the mean or maximum downwind concentrations observed
- The precision requirements for single measurements
- Cost and other logistical considerations

The duration of the removal action and the number of hours per day that site work is conducted determine the time, duration, and frequency of samples. Short-term sampling programs may require daily sampling, while long-term programs may require 24-hour sampling every sixth or twelfth day. If the site will be undergoing removal activities 24 hours a day, continuous air sampling may be warranted. However, if the site activities will be conducted for only eight hours a day, and there are no emissions likely to occur during the remaining 16 hours, then sampling would be appropriate prior to the start of daily activities, would continue during operations, and end at the conclusion of the daily activities. An off-peak sample collection can ensure that emissions are not persisting



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after the conclusion of daily cleanup activities. For some sites, emissions are still a factor several hours after daily site activities have been completed. Because of the typically decreased downwind dispersion in the evening, higher downwind concentrations than were present during daytime site activities may be detected. For sites where this is possible, the sampling duration needs to be lengthened accordingly.

Sampling duration and flow rate dictate the volume of air collected, and to a major degree, the detection limit. The analytical method selected will provide a reference to flow rate and volume. Flow rates are limited to the capacity of the pumps being employed and the contact time required by the collection media.

The duration or period of air sampling is commonly divided into two categories (1) samples collected over a brief time period are referred to as "instantaneous" or "grab" samples and are usually collected in less than five minutes and (2) average or integrated samples are collected over a significantly longer period of time. Integrated samples provide an average concentration over the entire sampling period. Integrated samples are not suited to determining cyclical releases of contaminants because periodic or cyclical events are averaged out by the proportionally long sampling duration.

Air quality dispersion models can predict the maximum air contaminant concentration expected from a source. The meteorological and site conditions expected to cause the highest concentration are known as worst-case conditions and can be identified by analyzing the modeling results. Depending upon the objective, one may sample when the model predicts worst-case conditions will exist.

7.2.5 Meteorological and Physical/Chemical Considerations

A meteorological monitoring program is an integral part of site monitoring activities. Meteorological data, which define local terrain impacts on air flow paths, are needed to interpret air concentration data. Meteorological data may be available from an existing station located near the site (i.e., at a local airport), otherwise a station should be set up at the site. This data will document the degree that samples actually were downwind and verify whether other worst-case assumptions were met. Meteorological parameters to be monitored are, at a minimum, wind speed, wind direction, and sigma theta (which is the horizontal wind direction standard deviation and an indicator of atmospheric stability). The remaining parameters primarily affect the amount of a contaminant available in the air.

- Wind Speed

When the contaminant of concern is a particulate, wind speed is critical in determining whether the particulate will become airborne, the quantity of the particulate that becomes airborne, and the distance the particulate will travel from the source. Wind speed also contributes to the volatilization of contaminants from liquid sources.

- Wind Direction



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Wind direction highly influences the path of airborne contaminants. In addition, variations in wind direction increase the dispersion of pollutants from a given source.

- Atmospheric Stability

Atmospheric stability refers to the degree to which the atmosphere tends to dampen vertical and horizontal motion. Stable atmospheric conditions (i.e., evenings) result in low dispersion, and unstable atmospheric conditions (i.e., hot sunny days) result in higher dispersion.

- Temperature

Higher temperatures increase the rate of volatilization of organic and some inorganic compounds and affect the initial rise of gaseous or vapor contaminants. Therefore, worst-case emission of volatiles and semivolatiles occurs at the hottest time of day, or on the hottest day.

- Humidity

High humidity affects water-soluble chemicals and particulates. Humid conditions may dictate the sampling media used to collect the air sample, or limit the volume of air sampled and thereby increase the detection limit.

- Atmospheric Pressure

Migration of landfill gases through the landfill surface and through surrounding soils are governed by changes in atmospheric pressure. Atmospheric pressure will influence upward migration of gaseous contaminants from shallow aquifers into the basements of overlying structures.

In many cases, the transport and dispersion of air pollutants is complicated by local meteorology. Normal diurnal variations (i.e., temperature inversions) affect dispersion of airborne contaminants. Terrain features can enhance or create air inversions and can also influence the path and speed of air flow, complicating transport and dispersion patterns.

The chemical characteristics of a contaminant (i.e., molecular weight, physical state, vapor pressure, aerodynamic size, temperature, reactive compounds, and photodegradation) affects its behavior and can influence the method used to sample and analyze it.

8.0 CALCULATIONS

Volume is obtained by multiplying the sample time in minutes by the flow rate. Sample volume should be indicated on the chain of custody record. Adjustments for temperature and pressure differences may be required.

Results are usually provided in parts per million (ppm), parts per billion (ppb), milligrams per cubic meter



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(mg/m³) or micrograms per cubic meter (µg/m³).

Refer to the analytical method or regulatory guidelines for other applicable calculations.

9.0 QUALITY ASSURANCE/QUALITY CONTROL

The manufacturer's instructions should be reviewed prior to instrument use. Instruments must be utilized in accordance with manufacturer's instructions. Equipment checkout and calibration activities must occur prior to and after monitoring and sampling and must be documented.

9.1 QA/QC Samples

QA/QC samples provide information on the variability and usability of environmental sample results. Various QA/QC samples may be collected to detect error. QA/QC samples are submitted with the field samples for analysis to aid in identifying the origin of analytical discrepancies; then a determination can be made as to how the analytical results should be used. Collocated samples, background samples, field blanks, and lot blanks are the most commonly collected QA/QC field samples. Performance evaluation (PE) samples and matrix spikes provide additional measures of data QA/QC control. QA/QC results may suggest the need for modifying sample collection, preparation, handling, or analytical procedures if the resultant data do not meet site-specific QA or data quality objectives. Refer to ERT/REAC SOP #2005, Quality Assurance/Quality Control Samples, for further details, and suggested frequencies for submittal of QA/QC samples.

9.2 Sample Documentation

All sample and monitoring activities should be documented legibly, in ink. Any corrections or revisions should be made by lining through the incorrect entry and by initialing the error. All samples must be recorded on an Air Sampling Worksheet. A chain of custody record must be maintained from the time a sample is taken to the final deposition of the sample. Custody seals demonstrate that a sample container has not been opened or tampered with during transport or storage of samples. Refer to ERT/REAC SOP #2002, Sample Documentation, for further information.

10.0 DATA VALIDATION

Results for QA/QC samples should be evaluated for contamination. This information should be utilized to qualify the environmental sample results accordingly with data quality objectives.

11.0 HEALTH AND SAFETY

Personal protection equipment (PPE) requirements identified in federal and/or state regulations and 29 Code of Federal Regulations (CFR) 1910.120 for hazardous waste site work must be followed.

The majority of physical precautions involved in air sampling are related to the contaminant sampled. Attention should be given when sampling in potentially explosive, flammable or acidic atmospheres. On rare occasions, the collection media may be hazardous; for example, in the instance where an acidic or basic solution is utilized in an impinger.



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When working with potentially hazardous materials, follow U.S. EPA, OSHA and corporate health and safety procedures.

12.0 REFERENCES

- (1) U.S. EPA. *Air Superfund National Technical Guidance Series. Volume I. Application of Air Pathway Analyses for Superfund Activities.* EPA/450/1-89/001.
- (2) U.S. EPA. *Air Superfund National Technical Guidance Series. Volume II. Estimation of Baseline Air Emissions at Superfund Sites.* EPA/450/1-89/002.
- (3) U.S. EPA. *Air Superfund National Technical Guidance Series. Volume III. Estimations of Air Emissions from Cleanup Activities at Superfund Sites.* EPA/450/1-89/003.
- (4) U.S. EPA. *Air Superfund National Technical Guidance Series. Volume IV. Procedures for Dispersion Air Modeling and Air Monitoring for Superfund Air Pathway Analysis.* EPA/450/1-89/004.
- (5) *Guide to Portable Instruments for Assessing Airborne Pollutants Arising from Hazardous Wastes*, International Organization of Legal Metrology (OIML) U.S. National Working Group (NWG) for OIML, American Conference of Governmental Industrial Hygienists, Cincinnati, OH
- (6) NIOSH. *Manual of Analytical Methods, Second Edition. Volumes 1-7.* U.S. Department of Health and Human Services Publication No. 84-100.

NIOSH. *Manual of Analytical Methods*, February 1984. U.S. Department of Health and Human Services Publication No. 84-100.
- (7) ASTM. 1990. *Annual Book of Standards, Volume 11.03.*
- (8) Riggin, R.M. *Compendium of Methods for the Determination of Toxic Organic Compounds in Ambient Air.* EPA/600/4-84/041.
- (9) Winberry, W.T. *Supplement to U.S. EPA/600/4-84/041: Compendium of Methods for the Determination of Toxic Organic Compounds in Ambient Air.* EPA/600/4-87/006.
- (10) OSHA. *Analytical Methods Manual, Second Edition. Part 1, Organic Substances*, January 1990. *Part 2, Inorganic Substances* August 1991.
- (11) Patty, F.A., *Industrial Hygiene and Toxicology, Third Edition*, John Wiley and Sons, Inc., New York, NY.
- (12) *Air Sampling Instruments for Evaluation of Atmospheric Contaminants, Seventh Edition*, 1989, American Conference of Governmental Industrial Hygienists, Cincinnati, OH

BIBLIOGRAPHY

Removal Program Representative Sampling Guidance, Volume 2: Air, Environmental Response Branch,



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Emergency Response Division, Office of Emergency and Remedial Response, Office of Solid Waste and Emergency Response, U.S. Environmental Protection Agency, April 1992, Interim Final.



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APPENDIX A
Portable Screening Devices and Specialized Analytical Instruments
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GENERAL AIR SAMPLING GUIDELINES

PORTABLE SCREENING DEVICES

Where possible, a datalogger should be used to minimize the length of time required for site personnel to be in a potentially contaminated area. Datalogger cable is available from manufacturers for linear output instruments and some nonlinear output instruments. U.S. EPA ERT/REAC has output cables for organic vapor analyzers (i.e., HNU and OVA), toxic gas analyzers (i.e., monitox) and real-time aerosol monitors (i.e., RAM and miniram).

- Total Hydrocarbon Analyzers

Total hydrocarbon analyzers used to detect a variety of volatile organic compounds (VOCs) at hazardous waste sites principally employ either a photoionization detector (PID) or a flame ionization detector (FID). Compounds are ionized by a flame or an ultraviolet lamp. PIDs depend on the ionization potential of the compounds. PIDs are sensitive to aromatic and olefinic (unsaturated) compounds such as benzene, toluene, styrene, xylenes, and acetylene. Greater selectivity is possible by using low-voltage lamps. The ionization potential of individual compounds can be found in the NIOSH Pocket Guide to Chemical Hazards. These instruments are not compound-specific and are typically used as screening instruments. FIDs are sensitive to volatile organic vapor compounds such as methane, propanol, benzene and toluene. They respond poorly to organic compounds lacking hydrocarbon characteristics.

- Oxygen and Combustible Gas Indicators

Combustible Gas Indicators (CGIs) provide efficient and reliable methods to test for potentially explosive atmospheres. CGI meters measure the concentration of a flammable vapor or gas in air and present these measurements as a percentage of the lower explosive limit (LEL). The measurements are temperature-dependent. The property of the calibration gas determines sensitivity. LELs for individual compounds can be found in the NIOSH Pocket Guide to Chemical Hazards. If readings approach or exceed 10% of the LEL, extreme caution should be exercised in continuing the investigation. If readings approach or exceed 25% LEL, personnel should be withdrawn immediately.

CGIs typically house an electrochemical sensor to determine the oxygen concentration in ambient air. Normally, air contains approximately 20.9% oxygen by volume. Oxygen measurements are of particular importance for work in enclosed spaces, low-lying areas, or in the vicinity of accidents that have produced heavier-than-air vapors which could displace ambient air. The meters are calibrated for sea level and may indicate a false negative (i.e., O₂ content) at higher altitudes. Since the air has been displaced by other substances, these oxygen-deficient areas are also prime locations for taking additional organic vapor and combustible gas measurements. Oxygen-enriched atmospheres increase the potential for fires by their ability to contribute to combustion or to chemically react with flammable compounds and promote auto-ignition.

- Toxic Atmosphere Analyzers

The toxic atmosphere analyzer is a compound-specific instrument, designed and calibrated to identify and quantify a specific compound or class of compounds in either gaseous or vapor form. Cross-sensitivity to air pollutants not of interest may lead to erroneous results.

U.S. EPA/ERT has the following toxic atmosphere analyzers: carbon monoxide, phosgene, nitrous oxide, hydrogen cyanide, sulfur dioxide, hydrogen sulfide, and chlorine gas.



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- Aerosol/Particulate Monitors

A Real-Time Aerosol/Particulate Monitor (RAM) displays readings for total particulates. The instrument employs a pulse light emitting diode which generates a narrow band emission in conjunction with a photovoltaic cell to detect light scattered from particulates.

The U.S. EPA/ERT uses the RAM when the contaminant of concern is associated with particulates, and when responding to fires involving hazardous materials, to identify plume levels. The instrument is very useful in determining the presence of a plume when it is not visible. The U.S. EPA/ERT typically uses RAMs on tripods to obtain particulate concentrations at the breathing zone level. Personal dataloggers are used with the RAMs to document minimum, average and maximum concentrations. This provides real-time data without requiring those in personal protective equipment to be constantly present in the plume.

- Chemical Detector Tubes (Colorimetric Tubes)

A chemical detector tube is a hollow, tube-shaped, glass body containing one or more layers of chemically impregnated inert material. To use, the fused ends are broken off and a manufacturer-specified volume of air is drawn through the tube with a pump to achieve a given detection limit. The chemicals contained within the packing material undergo a chemical reaction with the airborne pollutant present, producing a color change during the intake of each pump stroke. The concentration of a pollutant is indicated by the length of discoloration on a calibrated scale printed on the detector tube.

- Radiation Meters

Radiation meters determine the presence and level of radiation. The meters use a gas or solid ion detection media which becomes ionized when radiation is present. The meters are normally calibrated to one probe. Meters that detect alpha, beta, and gamma radiation are available.

- Gold Film (Hydrogen Sulfide and Mercury Vapor) Monitors

Hydrogen sulfide (H₂S) and Mercury (Hg) monitors operate on the principle that electric resistivity increases across a gold film as a function of H₂S and Hg concentration. The monitors provide rapid and relatively low detection limits for H₂S and Hg in air. After extensive sampling periods or high concentrations of H₂S and Hg, the gold film must be heated to remove contamination and return the monitor to its original sensitivity.

- Infrared Detectors

Infrared detectors such as the Miniature Infrared Analyzer (MIRAN) use infrared (IR) absorption as a function of specific compounds. MIRAN instruments apply to situations where the contaminants are identified but concentrations are not. MIRAN instruments generally require AC power.



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GENERAL AIR SAMPLING GUIDELINES

SPECIALIZED ANALYTICAL INSTRUMENTS

The continuous monitors described above provide qualitative measurement of air contaminants. Quantitative measurements in the field can be obtained using more sophisticated instruments, such as portable Gas Chromatographs, to analyze grab samples.

- Direct Air Sampling Portable Gas Chromatographs (GCs)

Portable GCs use gas chromatography to identify and quantify compounds. The time it takes for a compound to move through a chromatographic column is a function of that specific compound or group of compounds. A trained technician with knowledge of the range of expected concentrations of compounds can utilize a portable GC in the field to analyze grab samples. GCs generally require AC power and shelter to operate. This method is limited by its reliance on a short-term grab sample to be representative of the air quality at a site.

- Remote Optical Sensing

This technique, also referred to as long-path or open-path monitoring, involves transmitting either an infrared or ultraviolet light beam across a long open path and measuring the absorbance at specific wavelengths. The technique is capable of analyzing any preselected organic or inorganic volatile compound that can be resolved from compounds naturally occurring in ambient air. Current projected removal applications include perimeter monitoring during site cleanups and measurement of emission source strengths during site assessments.

- TAGA Direct Air Sampling Mass Spectrometer/Mass Spectrometer

The Trace Atmospheric Gas Analyzer (TAGA), which is operated by the U.S. EPA/ERT, is capable of real-time detection of preselected organic compounds at low parts-per-billion concentrations. The instrument has been successfully used by the U.S. EPA/ERT for isolating individual emission plumes and tracking those plumes back to their sources.



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GENERAL AIR SAMPLING GUIDELINES

APPENDIX B
Air Sampling Equipment and Media/Devices
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AIR SAMPLING EQUIPMENT

- High-Volume, Total Suspended Particulate (TSP) Samplers

High-volume TSP samplers collect all suspended particles by drawing air across an 8- by 10-inch glass-quartz filter. The sample rate is adjusted to 40 cubic feet per minute (CFM), or 1134 liters per minute (L/min), and it is held constant by a flow controller over the sample period. The mass of TSPs can be determined by weighing the filter before and after sampling. The composition of the filter varies according to the analytical method and the detection limit required.

- PM-10 Samplers

PM-10 samplers collect particulates with a diameter of 10 microns or less from ambient air. Particulates of this size represent the respirable fraction, and thus are of special significance. PM-10 samplers can be high-volume or low-volume. The high-volume sampler operates in the same manner as the TSP sampler at a constant flow rate of 40 CFM; it draws the sample through a special impactor head which collects particulates of 10 microns or less. The particulate is collected on an 8- by 10-inch filter. The low-volume sampler operates at a rate of approximately 17 L/min. The flow must remain constant through the impactor head to maintain the 10-micron cut-off point. The low-volume PM-10 collects the sample on 37-mm Teflon® filters.

- High-Volume PS-1 Samplers

High-volume PS-1 samplers draw a sample through polyurethane foam (PUF) or a combination foam and XAD-2 resin plug, and a glass quartz filter at a rate of 5-10 CFM (144 to 282 L/min). This system is excellent for measuring low concentrations of semivolatiles, PCBs, pesticides, or chlorinated dioxins in ambient air.

- Area Sampling Pumps

These pumps provide flow-rate ranges of 2-20 L/min and have a telescopic sampling mast with the sampling train. Because of the higher volume, this pump is suitable for sampling low concentrations of airborne contaminants (i.e., asbestos sampling). These pumps are also used for metals, pesticides and PAH sampling which require large sample volumes.

- Personal Sampling Pumps

Personal sampling pumps are reliable portable sampling devices that draw air samples through a number of sampling media including resin tubes, impingers, and filters. Flow rates are usually adjustable from 0.1 to 4 L/min (or 0.01 to .75 L/min with a restrictive orifice) and can remain constant for up to 8 hours on one battery charge or continuously with an AC charger/converter.

- Canister Samplers

Evacuated canister sampling systems use the pressure differential between the evacuated canister and ambient pressure to bleed air into the canister. The sample is bled into the canister at a constant rate over



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the sampling period using a critical orifice, a mechanically compensated regulator, or a mass flow control device until the canister is near atmospheric pressure.

Pressure canister sampling systems use a pump to push air into the canister. To maintain a higher, more controlled flow, the pump typically controls the pressure differential across a critical orifice at the inlet of the canister, resulting in a pressurized canister at the completion of sampling.



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GENERAL AIR SAMPLING GUIDELINES

AIR SAMPLING MEDIA/DEVICES

If possible, before employing a specific sampling method, consult the laboratory that will conduct the analyses. Many of the methods can be modified to provide better results or a wider range of results.

- Summa Canisters

Summa canisters are highly polished passivated stainless steel cylinders. The Summa polishing process brings chrome and nickel to the surface of the canisters, which results in an inert surface. This surface restricts adsorption or reactions that occur on the canister's inner surface after collection. At the site, the canister is either placed in a sampler to control sample collection rate, or opened to collect a grab sample. Samples can be collected by allowing air to bleed into or be pumped into the canister. U.S. EPA/ERT uses 6-liter Summa canisters for VOC and permanent gas analysis.

- Passive Dosimeters

Passive dosimeters are clip-on vapor monitors (samplers) in which the diffused contaminants are absorbed on specially prepared active surfaces. Industrial hygienists commonly use dosimeters to obtain time-weighted averages or concentrations of chemical vapors, as they can trap over 130 organic compounds. Selective dosimeters have also been developed for a number of chemicals including formaldehyde, ethylene oxide, hydrogen sulfide, mercury vapor, nitrogen dioxide, sulfur dioxide, and ozone. Dosimeters must be sent to a laboratory for analysis.

- Polyurethane Foam (PUF)

PUF is a sorbent used with a glass filter for the collection of semivolatile organic compounds such as pesticides, PCBs, chlorinated dioxins and furans, and PAHs. Fewer artifacts (chemical changes that occur to collected compounds) are produced than with some other solid sorbents. PUF is used with the PS-1 sampler and U.S. EPA Method TO13. PUF can also be used with personal sampling pumps when sampling for PAHs using the Lewis/McCloud method. Breakthrough of the more volatile PCBs and PAHs may occur when using PUF.

- Sampling Bags (Tedlar)

Sampling bags, like canisters, transport air samples to the laboratory for analysis. Samples are generally pumped into the bags, but sometimes a lung system is used, in which a pump creates a vacuum around the bag in a vacuum box. Then the sample flows from a source into the bag. This method is used for VOCs, fixed gases (CO_2 , O_2 and N_2) and methane.

- Impingers

An impinger allows an air sample to be bubbled through a solution, which collects a specific contaminant by either chemical reaction or absorption. For long sampling periods, the impinger may need to be kept in an ice bath to prevent the solution from evaporating during sampling. The sample is drawn through the impinger by using a sampling pump or more elaborate sampling trains with multiple impingers.

- Sorbent Tubes/Cartridges



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A variety of sampling media are available in sorbent tubes, which are used primarily for industrial hygiene. A few examples are carbon cartridges, carbon molecular sieves, Tenax tubes and tube containing the XAD-2 polymer. Depending upon the sorbent material, tubes can be analyzed using either a solvent extraction or thermal desorption. The former technique uses standard laboratory equipment and allows for multiple analyses of the same sample. The latter technique requires special, but readily available, laboratory equipment and allows only one analysis per sample. In addition, thermal desorption typically allows for lower detection limits by two or more orders of magnitude. Whenever sorbent tubes are being used for thermal desorption, they should be certified as "clean" by the laboratory doing the analysis.

Thermally Desorbed Media

During thermal desorption, high-temperature gas streams are used to remove the compounds collected on a sorbent medium. The gas stream is injected and often cryofocused into an analytical instrument, such as a GC, for compound analysis:

- Tenax Tubes

Tenax tubes are made from commercially available polymer (p-phenylene oxide) packed in glass or stainless steel tubes through which air samples are drawn or sometimes pumped. These tubes are used in U.S. EPA Method TO1 and VOST for volatile nonpolar organic, some polar organic, and some of the more volatile semivolatile organics. Tenax is not appropriate for many of the highly volatile organics (with vapor pressure greater than approximately 200 mm Hg).

- Carbonized Polymers

The carbonized molecular sieve (CMS), a carbonized polymer, is a commercially available, carbon sorbent packed in stainless-steel sampling tubes through which air samples are drawn or sometimes pumped. These are used in U.S. EPA Method TO2 for highly volatile nonpolar compounds which have low-breakthrough volumes on other sorbents. When high-thermal desorption temperatures are used with CMS, more variability in analysis may occur than with other sorbents.

- Mixed Sorbent Tubes

Sorbent tubes can contain two type of sorbents. Combining the advantages of each sorbent into one tube increases the possible types of compounds to be sampled. The combination of two sorbents can also reduce the chance that highly volatile compounds will break through the sorbent media. An example of a mixed sorbent tube is the combination of Tenax and charcoal with a carbonized molecular sieve. A potential problem with mixed sorbent tubes is the breakthrough of a compound from an earlier sorbent to a later sorbent from which it cannot be desorbed.

Solvent-Extracted Media

Solvent-extracted media use the principle of chemical extraction to remove compounds collected on a sorbent media. The chemical solvent is injected into an instrument, such as a GC, for analysis of compounds. Examples of solvent-extracted media follow:

- Chemically Treated Silica Gel



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Silica gel is a sorbent which can be treated with various chemicals. The chemically treated silica gel can then be used to sample for specific compounds in air. Examples include the DNPH-coated silica gel cartridge used with U.S. EPA Method TO11.

- XAD-2 Polymers

XAD-2 polymers usually are placed in tubes, custom-packed sandwich-style with polyurethane foam, and prepared for use with U.S. EPA Method TO13 or the semi-VOST method. The polymers are used for the collection of semivolatile polar and nonpolar organic compounds. The compounds collected on the XAD-2 polymer are chemically extracted for analysis.

- Charcoal Cartridges

Charcoal cartridges, consisting of primary and backup sections, trap compounds by adsorption. Ambient air is drawn through them so that the backup section verifies that breakthrough of the analytes on the first section did not occur, and the sample collection was therefore quantitative. Quantitative sample collection is evident by the presence of target chemicals on the first charcoal section and the absence on the second section. Next, the adsorbed compounds must be eluted, usually with a solvent extraction, and analyzed by GC with a detector, such as a Mass Spectrometer (MS).

- Tenax Tubes

Cartridges are used in OSHA and NIOSH methods in a manner similar to charcoal cartridges but typically for less volatile compounds.

- Particulate Filters

Particulate filters are used by having a sampling pump pass air through them. The filter collects the particulates present in the air and is then analyzed for particulate mass or chemical or radiological composition. Particulate filters are made from different materials which are described below.

- Mixed Cellulose Ester (MCE)

MCE is manufactured from mixed esters of cellulose which are a blend of nitro-cellulose and cellulose acetate. MCE filters are used often for particulate sampling.

- Glass Fiber

Glass fiber is manufactured from glass fibers without a binder. Particulate filters with glass fiber provide high flow rates, wet strength, and high, solid holding capacity. Generally, the filters are used for gravimetric analysis of particulates.

- Polyvinyl Chloride

Particulate filters with polyvinyl chloride are resistant to concentrated acids and alkalis. Their low moisture pickup and light tare weight make them ideal for gravimetric analysis.



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- Teflon

Teflon is manufactured from polytetrafluorethylene (PTFE). Particulate filters with Teflon are easy to handle and exceptionally durable. Teflon filters are used for metal collection.

- Silver

Particulate filters manufactured from pure silver have high collection efficiency and uniform pore size. These filters are used for mercury collection and analysis.

- Cellulose

Particulate filters with cellulose contain less than 0.01% ash. These filters are used to collect particulates.

ATTACHMENT C

NIOSH Method 6009



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ANALYSIS OF MERCURY IN AIR WITH A MODIFIED NIOSH 6009 METHOD

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ANALYSIS OF MERCURY IN AIR WITH A MODIFIED NIOSH 6009 METHOD

1.0 SCOPE AND APPLICATION

This method is a modification of NIOSH Method 6009 for mercury (Appendix A). It is applicable to the analysis of indoor air samples of volatilized elemental mercury (Hg) collected on solid sorbent material (typically Hopcalite) contained in glass collection tubes. The sorbent sample is digested and the Hg concentration is determined by the cold-vapor Atomic Absorption (AA) spectroscopy technique. The method is simple, rapid, and relatively free of matrix interferences.

Detection limits, sensitivity, and optimum ranges for Hg analysis will vary with the sorbent material, volume of air sampled, and models of atomic absorption spectrophotometers used.

These are standard operating procedures which may be varied or changed as required, depending upon site conditions, equipment limitations or limitations imposed by the procedure. In all instances, the ultimate procedures employed will be documented and associated with the final report.

Mention of trade names or commercial products does not constitute U.S. Environmental Protection Agency (U.S. EPA) endorsement or recommendation for use.

2.0 METHOD SUMMARY

Indoor air samples of elemental Hg are collected on solid sorbent material contained in glass tubes according to NIOSH method 6009. The sorbent material from the collection tube (typically 200 mg in a single section) is quantitatively transferred to a 100-milliliter (mL) volumetric flask. The sample is digested by first adding 2.5 mL of concentrated nitric acid followed by 2.5 mL of concentrated hydrochloric acid. After digestion is complete, the sample is diluted to volume with deionized water.

The sample is analyzed by the cold-vapor AA spectroscopy technique with no additional dilutions. The principle is essentially the same as direct aspiration AA, except a cold-vapor generator system, rather than a flame, is used to atomize the sample. Radiation from a given excited element is passed through the vapor containing ground-state atoms of that element. The intensity of the transmitted radiation decreases in proportion to the amount of the ground-state element in the vapor. A monochromator isolates the characteristic radiation from the hollow cathode lamp and a photosensitive device measures the attenuated transmitted radiation.

3.0 SAMPLE PRESERVATION, CONTAINERS, HANDLING, AND STORAGE

Sample holding times, suggested collection volume, preservative, and type of containers are as follows:

Measurement	Volume Req. (L)	Collection Type of Containers	Holding Preservative	Time
<hr/>				
<u>Mercury in air:</u>				
Solid sorbent	10 - 200 ⁽¹⁾	glass tube	sorbent, 25° C	21 days

⁽¹⁾ The volume of air collected is directly related to detection limit; the larger the volume, the lower the detection limit.



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4.0 INTERFERENCES AND POTENTIAL PROBLEMS

Although the method minimizes dilution and sample matrix effects, the technique is not completely interference free. Inorganic and organic Hg compounds may cause a positive interference

Cross-contamination and contamination of the sample can be major sources of error because of the sensitivities achieved with the cold-vapor AA spectroscopy technique. The sample preparation work area should be kept scrupulously clean. All glassware should be cleaned as directed in Section 5.3.

5.0 EQUIPMENT/APPARATUS

5.1 Atomic Absorption Spectrophotometer

A single- or dual-channel, single- or double-beam instrument having a grating monochromator, Hg hollow cathode lamp, photomultiplier detector, adjustable slits, a wavelength range of 190 to 800 nanometers (nm), and provisions for interfacing with a strip-chart recorder or computer, printer, autosampler, and Hg cold-vapor generation system.

5.2 Strip-Chart Recorder, Integrator, or Printer

A recorder is useful to provide a permanent record and for easy recognition of any problems with the analysis.

5.3 Glassware and Containers

All glassware, polypropylene, or Teflon containers, including sample bottles, should be washed in the following sequence: detergent, tap water, 1:1 nitric acid, tap water, 1:1 hydrochloric acid, tap water, and Type I water. If it can be documented through an active analytical quality control program, using spiked samples and reagent blanks that certain steps in the cleaning procedure are not required for routine samples, these steps may be eliminated from the procedure.

6.0 REAGENTS

All standard solutions are prepared and documented in accordance with ERT/SERAS SOP #1012, *Preparation of Standard Solutions*.

6.1 Type I Water (ASTM D1193)

Use Type I water for the preparation of all reagents and calibration standards, and as dilution water.

6.2 Concentrated Nitric Acid (HNO₃)

Use commercially available high-purity, spectrograde acid certified for AA use.

6.3 Concentrated Hydrochloric Acid (HCl)

Use commercially available high-purity, spectrograde acid certified for AA use.



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6.4 Stock Mercury Solutions

Use a commercially available Hg standard solution, accompanied by a certificate of analysis, or prepare a 1000 micrograms per milliliter ($\mu\text{g/mL}$) stock standard solution from high purity mercuric oxide (HgO) using Type I water and redistilled HCl. Dissolve 1.0798 grams (g) of dry HgO in 50 mL of 1:1 HCl, then dilute to one liter (1 L) with Type I water.

6.5 Stannous Chloride, 10% in 1:1 HCl

Dissolve 20 grams (g) reagent grade stannous chloride in 100 mL concentrated HCl. Slowly add this solution to 100 mL Type I water and mix well. Prepare fresh daily or each time calibration standards are prepared.

6.6 Calibration Standards

For those instruments that do not read out directly in concentration, a calibration curve is prepared to cover the appropriate concentration range. For best results, intermediate and working standards should be prepared fresh each time a batch of samples is analyzed. A blank and a minimum of five working standards must be used to calibrate the AA instrument.

Prepare all calibration and check standards using the procedures outlined in Section 7.0. Ideally, all QC standards are prepared by spiking blank sorbent media. This matches the sample matrix and, thereby, minimizes sample matrix effects. QC standards may also be prepared by spiking reagent blanks if sufficient blank sorbent media tubes are not available, or variability exists within media blanks.

7.0 PROCEDURES

7.1 Sample Preparation

1. Quantitatively transfer the sorbent and the front glass wool plug from each sampler tube into a 100-mL volumetric flask.
2. Add 2.5 mL of concentrated HNO_3 followed by 2.5 mL concentrated HCl.

NOTE: The Hg must be in the oxidized state to avoid loss. For this reason, the nitric acid must be added first.

3. Allow the sample to stand for 1 hour or until the black sorbent is dissolved. The solution will turn dark brown and may contain undissolved material.
4. Carefully dilute to 100 mL with Type I water.
5. The final sample solution (blue to blue-green in color) contains 2.5 percent HNO_3 , 2.5 percent HCl, and is analyzed for Hg with no additional dilution (except for samples containing high concentrations of Hg).

7.2 Calibration Standards



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Prepare a calibration blank, and a minimum of five working standards in graduated amounts in the linear part of the calibration range (0.2 to 10.0 µg/L) by spiking blank sorbent media (from unused sorbent tubes) with known amounts of Hg. Dissolve the blank sorbent media, using steps 1 - 3 of the procedure outlined in Section 7.1. Spike each standard solution with the appropriate amount of Hg, and dilute to volume per step 4 of the procedure outlined in Section 7.1.

NOTE: The calibration blank will also be used for the Initial Calibration Blank (ICB) and Continuing Calibration Blank (CCB) during sample analysis.

Calibration standards may also be prepared by spiking reagent blanks with known amounts of Hg to generate a calibration curve if sufficient blank sorbent media are not submitted, or variability exists within media blanks.

7.3 Laboratory Control Standard

Prepare the laboratory control standard (LCS) by spiking blank sorbent media (same lot and type of media) with a known amount of Hg (at or near midrange of the calibration curve). Use an independent source of Hg (different than that used to prepare calibration standards) for the LCS sample. Prepare the LCS sample using the procedure outlined in Section 7.1 at the same time the samples are prepared. A LCS will be analyzed with the frequency of one per batch.

If calibration standards are prepared by spiking reagent blanks, the LCS sample is also prepared by spiking a reagent blank with a known amount of Hg.

7.4 Calibration Verification Standards

Prepare the initial calibration verification (ICV) and the continuing calibration verification (CCV) standards by spiking blank sorbent media with known amounts of Hg (at or near midrange of the calibration curve). Prepare ICV and CCV standards using the procedure outlined in Section 7.1.

If calibration standards are prepared by spiking a reagent blank, the ICV and CCV standards are also prepared by spiking reagent blanks.

7.5 Method Detection Limit Standard

Prepare the method detection limit (MDL) standard by spiking blank sorbent media with a known amount of Hg at the expected MDL (typically half the concentration of the lowest calibration standard) using the same source used for calibration. Prepare the MDL standard using the procedure outlined in Section 7.1.

If the calibration standards are prepared by spiking the reagent blank, the MDL standard is also prepared by spiking a reagent blank.

7.6 Blank Spike/Blank Spike Duplicate

Prepare the media blank spike (BS) and blank spike duplicate (BSD) samples by spiking blank sorbent media with known amounts of Hg (5 to 10 times the detection limit) at a frequency of one in twenty samples or per batch. Spike blank sorbent media with appropriate amounts of Hg (near the midrange of the calibration) and prepare BS/BSD samples using the procedure outlined in Section 7.1. BS/BSD



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samples are prepared in the laboratory to monitor precision and accuracy of the method.

Use the BS/BSD samples delivered to the laboratory with unknown samples, or blank tubes from the same lot if none are provided with the samples.

7.7 Lot Blanks

This section applies only if calibration and QC standards are prepared in reagent blank. If this is the case, a minimum of three (3) lot blanks must be prepared and analyzed with the samples.

7.8 AA Calibration and Measurement

Differences between the various makes and models of satisfactory AA instruments and cold-vapor generators prevent the formulation of detailed instructions applicable to each system. The analyst should follow the manufacturer's operating instructions for a particular instrument and cold-vapor generator system.

Analyze the working standards together with the samples and blanks. Analyze the full set of working standards at the beginning of the run to establish the initial calibration curve. Analyze additional standards during sample analysis to confirm instrument response (see Section 9.0).

Samples with concentrations of Hg over the high standard must be diluted into the linear calibration range (see Section 9.5).

8.0 CALCULATIONS

For determination of concentration, read the Hg value (B) in µg/L from the calibration curve or directly from the read-out system of the instrument. Calculate the concentration of Hg in the sample (A) as follows:

$$A = \mu\text{g/L Hg sample} = 3 \times \frac{B + C}{D}$$

where:

- B = Concentration of Hg from the calibration curve (µg/L)
- C = Amount of acid blank matrix used for dilution (mL)
- D = Sample aliquot used for dilution (mL)

Note: If no dilution was required, C = 0, and A = B.

Calculate the amount of Hg (W) for each sorbent tube:

$$W = \mu\text{g/tube Hg} = A (0.1)$$

where:

- A = Concentration of Hg in the sample (µg/L)
- 0.1 = Final solution volume (L/tube)



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Calculate the concentration ($\mu\text{g}/\text{m}^3$) of Hg in the air volume sampled:

$$\mu / \text{m}^3 \text{ Hg} = 1000 \times \frac{W - 3LK}{V}$$

where:

- W = Amount of Hg in each sorbent tube ($\mu\text{g}/\text{tube}$)
- V = Volume of air sampled (L/tube)
- BLK = amount of Hg in the blank tube ($\mu\text{g}/\text{tube}$)
 - = zero (0) when standards are prepared by spiking blank sorbent media; or
 - = average of lot blank results when standards are prepared in reagent blanks (Section 7.7)

(For any blank value $< \text{MDL}$, substitute the value of 0.0 for the raw data prior to calculating the averages. The calculated average is subtracted from each sample even if this average is $< \text{MDL}$.)

NOTE: Report $\mu\text{g}/\text{tube}$ results for lot blank, trip blank, field blank(s), and BS/BSD samples because no air volume is collected for these samples.

9.0 QUALITY ASSURANCE/QUALITY CONTROL

All quality control data should be documented and available for reference or inspection.

9.1 Initial Calibration/Calibration Check

A calibration curve must be prepared each day as described in Section 7.2. The correlation coefficient (r) must be greater than or equal to 0.995 for an acceptable calibration. The initial calibration must be verified by analysis of the ICV standard (Section 7.4), the ICB standard (Section 7.2), and the LCS sample (Section 7.3). The ICV result must be within ± 10 percent of the true value. Results for the ICB and method blank must be less than the MDL. The LCS result must be within the Performance Acceptance Limits (PALs) supplied by the vendor.

9.2 Method Detection Limit

The MDL standard (Section 7.5) must be analyzed at the beginning of sample analysis to verify the Hg MDL. The results for the MDL standard must be within ± 20 percent of the true value. If the MDL standard is not within ± 20 percent, the MDL must be elevated to the concentration of the lowest calibration standard.

9.3 Continuing Calibration Verification

The working standard curve must be verified by analyzing the CCV (Section 7.4) and the CCB standard after every 10 samples. CCV results must be within ± 20 percent of the true value and CCB results must be less than the MDL.

9.4 BS/BSD Samples



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At least one BS and one BSD sample (Section 7.6) must be analyzed with each batch of samples (not to exceed 20 samples) to verify precision and accuracy of the method.

BS/BSD percent recovery (%R) should be within the advisory limit of 75-125 percent and calculated as:

$$\%R = \frac{SSR}{SA} \times 100$$

where:

SSR = Spiked (BS or BSD) sample result (µg/tube)

SA = Spike added (µg/tube)

The Relative Percent Difference (RPD) for the BS/BSD samples should be within the advisory limit of ±20 percent and calculated as:

$$RPD = \frac{S - D}{(S + D)/2} \times 100$$

where:

S = %R for BS sample result

D = %R for BSD sample result

9.5 Dilution Analysis

If the Hg concentration of any sample exceeds the initial calibration range, that sample must be diluted and reanalyzed. Use the results of the original analysis to determine the approximate dilution factor required to get a Hg concentration within the initial linear calibration range. Dilute the sample (less than 10 mL sample required for analysis) and analyze the diluted sample aliquot. Report results for the diluted aliquot with the lowest dilution factor which produces a Hg concentration in the linear calibration range (see Section 8.0).

10.0 DATA VALIDATION

Data will be assessed by the Data Validation & Report Writing Group using the most current revision of the ERT/SERAS SOP #1017, Data Validation Procedure for Routine Inorganic Analysis. However, data is considered satisfactory for submission purposes when all of the requirements listed in the method are met.

11.0 HEALTH AND SAFETY

The toxicity or carcinogenicity of each reagent used in this method has not been precisely defined. However, each chemical compound should be treated as a potential health hazard. The laboratory is responsible for following the chemical hygiene plan and laboratory safety program regarding the safe handling of the chemicals specified in this method.

When working with potentially hazardous materials, refer to U.S. EPA, Occupational Safety and Health Administration (OSHA) and corporate health and safety practices. More specifically, refer to ERT/SERAS SOP



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#3013, *SERAS Laboratory Safety Program*.

The analyst should consult all appropriate MSDS information prior to running an analysis for the first time.

12.0 REFERENCES

U.S. Environmental Protection Agency, Office of Solid Waste and Emergency Response. 1992. *Test Methods for Evaluating Solid Waste*, SW-846. 3rd ed.

National Institute for Occupational Safety and Health. 1996. *NIOSH Manual of Analytical Methods*. Method 6009. 4th ed. Suppl.

U.S. EPA ERT/SERAS SOP #1818, *Determination of Metals by Atomic Absorption (AA) Methods*.



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APPENDIX A
NIOSH Method 6009 for Mercury
SOP #1827
February 2001



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ANALYSIS OF MERCURY IN AIR WITH A MODIFIED NIOSH 6009 METHOD

MERCURY

6009

Hg MW: 200.59 CAS: 7439-97-6 RTECS: OV4550000

METHOD: 6009, Issue 2

EVALUATION: PARTIAL

Issue 1: 15 May 1989
Issue 2: 15 August 1994

OSHA : C 0.1 mg/m³ (skin)
NIOSH: 0.05 mg/m³ (skin)
ACGIH: 0.025 mg/m³ (skin)

PROPERTIES: liquid; d 13.55 g/mL @ 20 °C; BP 356 °C;
HP -39 °C; VP 0.16 Pa (0.0012 mm Hg;
13.2 mg/m³) @ 20 °C; Vapor Density
(air=1) 7.0

SYNONYMS: quicksilver

SAMPLING		MEASUREMENT	
SAMPLER:	SOLID SORBENT TUBE (Hopcalite in single section, 200 mg)	TECHNIQUE:	ATOMIC ABSORPTION, COLD VAPOR
FLOW RATE:	0.15 to 0.25 L/min	ANALYTE:	elemental mercury
VOL-MIN:	2 L @ 0.5 mg/m ³	DESORPTION:	conc. HNO ₃ /HCl @ 25 °C, dilute to 50 mL
-MAX:	100 L	WAVELENGTH:	253.7 nm
SHIPMENT:	routine	CALIBRATION:	standard solutions of Hg ²⁺ in 1% HNO ₃
SAMPLE STABILITY:	30 days @ 25 °C [1]	RANGE:	0.1 to 1.2 µg per sample
FIELD BLANKS:	2 to 10 field blanks per set	ESTIMATED LOD:	0.03 µg per sample
MEDIA BLANKS:	at least 3 per set	PRECISION (S _p):	0.042 @ 0.9 to 3 µg per sample [4]
ACCURACY			
RANGE STUDIED:	0.002 to 0.8 mg/m ³ [2] (10-L samples)		
BIAS:	not significant		
OVERALL PRECISION (S _p):	not determined		
ACCURACY:	not determined		

APPLICABILITY: The working range is 0.01 to 0.5 mg/m³ for a 10-L air sample. The sorbent material irreversibly collects elemental mercury. A prefilter can be used to exclude particulate mercury species from the sample. The prefilter can be analyzed by similar methodology. The method has been used in numerous field surveys [3].

INTERFERENCES: Inorganic and organic mercury compounds may cause a positive interference. Oxidizing gases, including chlorine, do not interfere.

OTHER METHODS: This replaces method 6000 and its predecessors, which required a specialized desorption apparatus [4,5,6]. This method is based on the method of Rathje and Marcero [7] and is similar to the OSHA method ID 145H [2].



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REAGENTS:

1. Water, organics-free, deionized.
2. Hydrochloric acid (HCl), conc.
3. Nitric acid (HNO_3), conc.
4. Mercuric oxide, reagent grade, dry.
5. Calibration stock solution, Hg^{2+} , 1000 $\mu\text{g/mL}$. Commercially available or dissolve 1.0798 g of dry mercuric oxide (HgO) in 50 mL of 1:1 hydrochloric acid, then dilute to 1 L with deionized water.
6. Intermediate mercury standard, 1 $\mu\text{g/mL}$. Place 0.1 mL 1000 $\mu\text{g/mL}$ stock into a 100 mL volumetric containing 10 mL deionized water and 1 mL hydrochloric acid. Dilute to volume with deionized water. Prepare fresh daily.
7. Stannous chloride, reagent grade, 10% in 1:1 HCl. Dissolve 20 g stannous chloride in 100 mL conc. HCl. Slowly add this solution to 100 mL deionized water and mix well. Prepare fresh daily.
8. Nitric acid, 1% (w/v). Dilute 14 mL conc. HNO_3 to 1 L with deionized water.

EQUIPMENT:

1. Sampler: glass tube, 7 cm long, 6-mm OD, 4-mm ID, flame sealed ends with plastic caps, containing one section of 200 mg Hopcalite held in place by glass wool plugs (SKC, Inc., Cat. #226-17-1A, or equivalent).
NOTE: A 37-mm, cellulose ester membrane filter in a cassette preceding the sorbent may be used if particulate mercury is to be determined separately.
2. Personal sampling pump, 0.15 to 0.25 L/min, with flexible connecting tubing.
3. Atomic absorption spectrophotometer with cold vapor generation system (see Appendix) or cold vapor mercury analysis system.*
4. Strip chart recorder, or integrator.
5. Flasks, volumetric, 50-mL, and 100-mL.
6. Pipet, 5-mL, 20-mL, others as needed.
7. Micropipet, 10- to 1000- μL .
8. Bottles, biological oxygen demand (BOD), 300-mL.

* See SPECIAL PRECAUTIONS

SPECIAL PRECAUTIONS: Mercury is readily absorbed by inhalation and contact with the skin. Operate the mercury system in a hood, or bubble vented mercury through a mercury scrubber.

SAMPLING:

1. Calibrate each personal sampling pump with a representative sampler in line.
2. Break ends of sampler immediately prior to sampling. Attach sampler to pump with flexible tubing.
3. Sample at an accurately known rate of 0.15 to 0.25 L/min for a total sample size between 2 and 100 L.
NOTE: Include a minimum of three unopened sampling tubes from the same lot as the samples for use as media blanks.
4. Cap sampler and pack securely for shipment.

SAMPLE PREPARATION:

5. Place the Hopcalite sorbent and the front glass wool plug from each sampler in separate 50-mL volumetric flasks.
6. Add 2.5 mL conc. HNO_3 followed by 2.5 mL conc. HCl.
NOTE: The mercury must be in the oxidized state to avoid loss. For this reason, the nitric acid must be added first.
7. Allow the sample to stand for 1 h or until the black Hopcalite sorbent is dissolved. The solution will turn dark brown and may contain undissolved material.
8. Carefully dilute to 50 mL with deionized water. (Final solution is blue to blue-green).
9. Using a volumetric pipet, transfer 20 mL of the sample to a BOD bottle containing 80 mL of deionized water. If the amount of mercury in the sample is expected to exceed the standards, a smaller aliquot may be taken, and the volume of acid adjusted accordingly. The final volume in



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REAGENTS:

1. Water, organics-free, deionized.
2. Hydrochloric acid (HCl), conc.
3. Nitric acid (HNO₃), conc.
4. Mercuric oxide, reagent grade, dry.
5. Calibration stock solution, Hg²⁺, 1000 µg/mL. Commercially available or dissolve 1.0798 g of dry mercuric oxide (HgO) in 50 mL of 1:1 hydrochloric acid, then dilute to 1 L with deionized water.
6. Intermediate mercury standard, 1 µg/mL. Place 0.1 mL 1000 µg/mL stock into a 100 mL volumetric containing 10 mL deionized water and 1 mL hydrochloric acid. Dilute to volume with deionized water. Prepare fresh daily.
7. Stannous chloride, reagent grade, 10% in 1:1 HCl. Dissolve 20 g stannous chloride in 100 mL conc. HCl. Slowly add this solution to 100 mL deionized water and mix well. Prepare fresh daily.
8. Nitric acid, 1% (w/v). Dilute 14 mL conc. HNO₃ to 1 L with deionized water.

EQUIPMENT:

1. Sampler: glass tube, 7 cm long, 6-mm OD, 4-mm ID, flame sealed ends with plastic caps, containing one section of 200 mg Hopcalite held in place by glass wool plugs (SKC, Inc., Cat. #226-17-1A, or equivalent).
NOTE: A 37-mm, cellulose ester membrane filter in a cassette preceding the sorbent may be used if particulate mercury is to be determined separately.
2. Personal sampling pump, 0.15 to 0.25 L/min, with flexible connecting tubing.
3. Atomic absorption spectrophotometer with cold vapor generation system (see Appendix) or cold vapor mercury analysis system.*
4. Strip chart recorder, or integrator.
5. Flasks, volumetric, 50-mL, and 100-mL.
6. Pipet, 5-mL, 20-mL, others as needed.
7. Micropipet, 10- to 1000-µL.
8. Bottles, biological oxygen demand (BOD), 300-mL.

* See SPECIAL PRECAUTIONS

SPECIAL PRECAUTIONS: Mercury is readily absorbed by inhalation and contact with the skin. Operate the mercury system in a hood, or bubble vented mercury through a mercury scrubber.

SAMPLING:

1. Calibrate each personal sampling pump with a representative sampler in line.
2. Break ends of sampler immediately prior to sampling. Attach sampler to pump with flexible tubing.
3. Sample at an accurately known rate of 0.15 to 0.25 L/min for a total sample size between 2 and 100 L.
NOTE: Include a minimum of three unopened sampling tubes from the same lot as the samples for use as media blanks.
4. Cap sampler and pack securely for shipment.

SAMPLE PREPARATION:

5. Place the Hopcalite sorbent and the front glass wool plug from each sampler in separate 50-mL volumetric flasks.
6. Add 2.5 mL conc. HNO₃ followed by 2.5 mL conc. HCl.
NOTE: The mercury must be in the oxidized state to avoid loss. For this reason, the nitric acid must be added first.
7. Allow the sample to stand for 1 h or until the black Hopcalite sorbent is dissolved. The solution will turn dark brown and may contain undissolved material.
8. Carefully dilute to 50 mL with deionized water. (Final solution is blue to blue-green).
9. Using a volumetric pipet, transfer 20 mL of the sample to a BOD bottle containing 80 mL of deionized water. If the amount of mercury in the sample is expected to exceed the standards, a smaller aliquot may be taken, and the volume of acid adjusted accordingly. The final volume in



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EVALUATION OF METHOD:

Rathje and Marcero originally used Hopcalite (MSA, Inc.) as the sorbent material [7]. Later, Hopcalite was shown superior to other methods for the determination of mercury vapor [8]. Atmospheres of mercury vapor for the study were dynamically generated in the range 0.05 to 0.2 mg/m³ and an adsorbent tube loading of 1 to 7 µg was used. The Hydrar material sometimes used is similar to Hopcalite. No significant difference in the laboratory analysis of mercury collected on the two sorbent materials was observed [9]. OSHA also validated a method for mercury using Hydrar [2]. An average 99% recovery, with $\bar{S}_r = 0.042$, was seen for 18 samples with known amounts (0.9 to 3 µg) of mercury added (as Hg(NO₃)₂) [10]. No change in recovery was seen for samples stored up to 3 weeks at room temperature or up to 3 months at -15 °C; longer storage times were not investigated [10].

REFERENCES:

- [1] Evaluation of Mercury Solid Sorbent Passive Dosimeter. Backup Data Report. Inorganic Section, OSHA Analytical Laboratory, Salt Lake City, Utah, 1985.
- [2] Mercury in Workplace Atmospheres (Hydrar Tubes). Method ID 145H, Inorganic Section, OSHA Analytical Laboratory, Salt Lake City, UT, 1987.
- [3] NIOSH/MRSB. Reports for analytical Sequence Nos. 5854, 5900, 6219, and 6311, NIOSH (Unpublished, 1987-1988).
- [4] NIOSH Manual of Analytical Methods, 3rd. ed., Method 6000. (1984).
- [5] NIOSH Manual of Analytical Methods. 2nd. ed., V. 4, S199, U.S. Dept. of Health, Education, and Welfare Publ. (NIOSH) 79-141 (1979).
- [6] Ibid., V. 5, P&CAM 175, Publ. (NIOSH) 79-141 (1979).
- [7] Rathje, A.O., Marcero, D.H. Improved hopcalite procedure for the determination of mercury in air by flameless atomic absorption. *Am. Ind. Hyg. Assoc. J.* **37**, 311-314 (1976).
- [8] McCammon, C.S., Edwards, S.L., Hull, R.D., Woodfin, W.J., A comparison of four personal sampling methods for the determination of mercury vapor, *Am. Ind. Hyg. Assoc. J.*, **41**, 528-531 (1980).
- [9] Internal Methods Development Research, DataChem Laboratories, Inc., Salt Lake City, UT (1982).
- [10] Eller, P.M., NIOSH, unpublished data (1987-88).

METHOD WRITTEN BY:

Keith R. Nicholson and Michael R. Steele, DataChem Laboratories, Inc., Salt Lake City, Utah, under NIOSH contract No. 200-87-2533.

APPENDIX: COLD VAPOR MERCURY ANALYSIS SYSTEM



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1. The valve should direct the vented vapors to a hood or to a mercury scrubber system.
2. When the valve is opened to "Vent" the peristaltic pump should draw room air. Place a Hopcalite tube in the air intake to eliminate any mercury that may be present.
3. Adjust the peristaltic pump to a flow that will create a steady stream of bubbles in the BOD bottle, but not so great that solution droplets enter the tubing to the quartz cell.
4. If water vapor condenses in the quartz cell, heat the cell slightly above room temperature by wrapping it with a heating coil and attaching a variable transformer.
5. The bubbler consists of a glass tube with a bulb at the bottom, slightly above the bottom of the BOD bottle. The bulb contains several perforations to allow air to escape into the solution (in a stream of small bubbles). A second tube is provided to allow the exit of the vapor. The open end of the second tube is well above the surface of the liquid in the bottle. The two tubes are fixed into a stoppering device (preferably ground glass) which fits into the top of the bottle. A coarse glass frit can be used in place of the bulb on the first tube. However, it is more difficult to prevent contamination when a frit is used.
6. Replace the flexible tubing (Tygon or equivalent) used to connect the bubbler, cell, and pump periodically to prevent contamination from adsorbed mercury.

ATTACHMENT D

EPA RSLs

Key: I = IRIS; P = PPRTV; D = DWSHA; O = OPP; A = ATSDR; C = Cal EPA; X = APPENDIX PPRTV SCREEN (See FAQ #29); H = HEAST; F = See FAQ; E = see user guide Section 2.3.5; W = see user guide Section 2.3.6; L = see user guide on lead; M = mutagen; S = see user guide Section 5; V = volatile; R = RBA applied (See User Guide for Arsenic notice); c = cancer; n = noncancer; * = where n SL < 100X c SL; ** = where n SL < 10X c SL; SSL values are based on DAF=1; m = Concentration may exceed ceiling limit (See User Guide); s = Concentration may exceed Csat (See User Guide)

Toxicity and Chemical-specific Information										Contaminant		Screening Levels										SSLs				
SFO (mg/kg-day) ⁻¹	k _e IUR (ug/m ³) ⁻¹	k _e RfD ₀ (mg/kg-day)	k _e RfC ₁ (mg/m ³) ⁻¹	k _e V _o gen	muta- gen	GIAB S	ABS	C _{sat} (mg/kg)	Analyte	CAS No.	Resident Soil (mg/kg)	key	Industrial Soil (mg/kg)	key	Resident Air (ug/m ³)	key	Industrial Air (ug/m ³)	key	Tapwater (ug/L)	key	MCL (ug/L)	Risk-based SSL (mg/kg)	key	MCL-based SSL (mg/kg)		
2.2E-06	I	1.2E-03	O				1	0.1	Acetate	30560-19-1	7.6E+01	n	9.8E+02	n									5.3E-03	n		
		2.0E-02	I				1	0.1	Acetaldehyde	75-07-0	1.1E+01	c**	4.9E+01	c**	1.3E+00	c**	5.6E+00	c**	2.4E+01	n			5.2E-04	c**		
							1	0.1	Acetochlor	34256-82-1	1.3E+03	n	1.6E+04	n					3.5E+02	n			2.8E-01	n		
		9.0E-01	I	3.1E+01 A	V		1		Acetone	67-64-1	6.1E+04	n	6.7E+05	s	3.2E+04	n	1.4E+05	n	1.4E+04	n			2.9E+00	n		
		2.0E-03	X				1	0.1	Acetone Cyanohydrin	75-86-5	2.8E+06	nm	1.2E+07	nm	2.1E+00	n	8.8E+00	n								
		6.0E-02	I	V			1		Acetonitrile	75-05-8	8.1E+02	n	3.4E+03	n	6.3E+01	n	2.6E+02	n	1.3E+02	n			2.6E-02	n		
3.8E+00	C	1.3E-03	C	1.0E-01	I		V	1	0.1	Acetophenone	98-86-2	7.8E+03	ns	1.2E+05	s					1.9E+03	n		5.8E-01	n		
		5.0E-04	I	2.0E-05	I	V		1	0.1	Acetylaminofluorene, 2-Acrolein	53-96-3	1.4E-01	c	6.0E-01	c	2.2E-03	c	9.4E-03	c	1.6E-02	c		7.2E-05	c		
5.0E-01	I	1.0E-04	I	2.0E-03	I	6.0E-03	I	M	1	0.1	Acrylamide	79-06-1	2.4E-01	c	4.6E+00	c	1.0E-02	c	1.2E-01	c	5.0E-02	c		1.1E-05	c	
		5.0E-01	I	1.0E-03	I	V		1	0.1	Acrylic Acid	79-10-7	9.9E+01	n	4.2E+02	n	1.0E+00	n	4.4E+00	n	2.1E+00	n		4.2E-04	n		
5.4E-01	I	6.8E-05	I	4.0E-02	A	2.0E-03	I	V	1	0.1	Acrylonitrile	107-13-1	2.5E-01	c*	1.1E+00	c*	4.1E-02	c*	1.8E-01	c*	5.2E-02	c*		1.1E-05	c*	
				6.0E-03	P		1	0.1	Adiponitrile	111-69-3	8.5E+06	nm	3.6E+07	nm	6.3E+00	n	2.6E+01	n								
5.6E-02	C	1.0E-02	I				1	0.1	Alachlor	15972-60-8	9.7E+00	c*	4.1E+01	c					1.1E+00	c	2.0E+00	8.7E-04	c	1.6E-03		
		1.0E-03	I				1	0.1	Aldicarb	116-06-3	6.3E+01	n	8.2E+02	n					2.0E+01	n	3.0E+00	4.9E-03	n	7.5E-04		
		1.0E-03	I				1	0.1	Aldicarb Sulfone	1646-88-4	6.3E+01	n	8.2E+02	n					2.0E+01	n	2.0E+00	4.4E-03	n	4.4E-04		
1.7E+01	I	4.9E-03	I	3.0E-05	I		V	1		Aldicarb sulfoxide	1646-87-3											4.0E+00				
							1		Aldrin	309-00-2	3.9E-02	c*	1.8E-01	c	5.7E-04	c	2.5E-03	c	9.2E-04	c		1.5E-04	c			
2.1E-02	C	6.0E-06	C	5.0E-03	I	1.0E-04	X	V	1	0.1	Allyl Alcohol	107-18-6	3.5E+00	n	1.5E+01	n	1.0E-01	n	4.4E-01	n	2.1E-01	n		4.2E-05	n	
		1.0E+00	P	5.0E-03	P		1		Allyl Chloride	107-05-1	7.2E-01	c**	3.2E+00	c**	4.7E-01	c**	2.0E+00	c**	7.3E-01	c**		2.3E-04	c**			
		4.0E-04	I				1		Aluminum	7429-90-5	7.7E+04	n	1.1E+06	nm	5.2E+00	n	2.2E+01	n	2.0E+04	n		3.0E+04	n			
2.1E+01	C	6.0E-03	C	9.0E-03	I		1	0.1	Aluminum Phosphide	20859-73-8	3.1E+01	n	4.7E+02	n					8.0E+00	n						
		8.0E-02	P				1	0.1	Ametryn	834-12-8	5.7E+02	n	7.4E+03	n					1.5E+02	n		1.6E-01	n			
		4.0E-03	X				1	0.1	Aminobiphenyl, 4-	92-67-1	2.6E-02	c	1.1E-01	c	4.7E-04	c	2.0E-03	c	3.0E-03	c		1.5E-05	c			
		2.0E-02	P				1	0.1	Aminophenol, m-	591-27-5	5.1E+03	n	6.6E+04	n					1.6E+03	n		6.1E-01	n			
		2.5E-03	I				1	0.1	Aminophenol, o-	95-55-6	2.5E+02	n	3.3E+03	n					7.9E+01	n		3.0E-02	n			
				5.0E-01	I	V	1		Aminophenol, p-	123-30-8	1.3E+03	n	1.6E+04	n					4.0E+02	n		1.5E-01	n			
		2.0E-01	I				1		Amtraz	33085-61-1	1.6E+02	n	2.1E+03	n					8.2E+00	n		4.2E+00	n			
							1		Ammonia	7664-41-7					5.2E+02	n	2.2E+03	n								
							1		Ammonium Sulfamate	7773-06-0	1.6E+04	n	2.3E+05	nm					4.0E+03	n						
5.7E-03	I	1.6E-06	C	7.0E-03	P	1.0E-03	I	1	0.1	Amyl Alcohol, tert	75-85-4	8.2E+01	n	3.4E+02	n	3.1E+00	n	1.3E+01	n	6.3E+00	n		1.3E-03	n		
4.0E-02	P	2.0E-03	X	2.0E-03	X		1	0.1	Aniline	62-53-3	9.5E+01	c**	4.0E+02	c*	1.0E+00	n	4.4E+00	n	1.3E+01	c*		4.6E-03	c*			
							0.15		Anthraquinone, 9,10-	84-65-1	1.4E+01	c**	5.7E+01	c*					1.4E+00	c*		1.4E-02	c*			
		4.0E-04	I				0.15		Antimony (metallic)	7440-36-0	3.1E+01	n	4.7E+02	n					7.8E+00	n	6.0E+00	3.5E-01	n	2.7E-01		
		5.0E-04	H				0.15		Antimony Pentoxide	1314-60-9	3.9E+01	n	5.8E+02	n					9.7E+00	n						
		4.0E-04	H				0.15		Antimony Tetroxide	1332-81-6	3.1E+01	n	4.7E+02	n					7.8E+00	n						
1.5E+00	I	4.3E-03	I	3.0E-04	I	1.5E-05	C	0.03	Antimony trioxide	1309-64-4	2.8E+05	nm	1.2E+06	nm	2.1E-01	n	8.8E-01	n								
		3.6E-02	O	3.5E-06	C	5.0E-05	I	1	0.1	Arsenic, inorganic	7440-38-2	6.8E-01	c**	3.0E+00	c**	6.5E-04	c*	2.9E-03	c*	5.2E-02	c	1.0E+01	1.5E-03	c	2.9E-01	
		3.5E-06	C	5.0E-05	I		1		Arsine	7784-42-1	2.7E-01	n	4.1E+00	n	5.2E-02	n	2.2E-01	n	7.0E-02	n						
2.3E-01	C	3.5E-02	I				1	0.1	Asulam	3337-71-1	2.3E+03	n	3.0E+04	n					7.2E+02	n		1.8E-01	n			
8.8E-01	C	2.5E-04	C	3.5E-02	I		1	0.1	Atrazine	1912-24-9	4.6E+00	c	1.0E+01	c					3.0E-01	c	3.0E+00	2.0E-04	c	1.9E-03		
							1	0.1	Auramine	492-80-8	6.2E-01	c	2.6E+00	c	1.1E-02	c	4.9E-02	c	6.7E-02	c		6.1E-04	c			
1.1E-01	I	3.1E-05	I	4.0E-04	I	1.0E-02	A	1	0.1	Avermectin B1	65195-55-3	2.5E+01	n	3.3E+02	n					8.0E+00	n		1.4E+01	n		
							1	0.1	Azinphos-methyl	86-50-0	1.9E+02	n	2.5E+03	n	1.0E+01	n	4.4E+01	n	5.6E+01	n		1.7E-02	n			
							1	0.1	Azobenzene	103-33-3	5.6E+00	c	2.6E+01	c	9.1E-02	c	4.0E-01	c	1.2E-01	c		9.3E-04	c			
		1.0E+00	P	7.0E-06	P		1	0.1	Azodicarbonamide	123-77-3	8.6E+03	n	4.0E+04	n	7.3E-03	n	3.1E-02	n	2.0E+04	n		6.8E+00	n			
		2.0E-01	I	5.0E-04	H		0.07		Barium	7440-39-3	1.5E+04	n	2.2E+05	nm	5.2E-01	n	2.2E+00	n	3.8E+03	n	2.0E+03	1.6E+02	n	8.2E+01		
		5.0E-03	O				1		Benfluralin	1861-40-1	3.9E+02	n	5.8E+03	n					2.8E+01	n		9.4E-01	n			
		5.0E-02	I				1	0.1	Benomyl	17804-35-2	3.2E+03	n	4.1E+04	n					9.7E+02	n		8.5E-01	n			
		2.0E-01	I				1	0.1	Bensulfuron-methyl	83055-99-6	1.3E+04	n	1.6E+05	nm					3.9E+03	n		1.0E+00	n			
		3.0E-02	I				1	0.1	Bentazon	25057-89-0	1.9E+03	n	2.5E+04	n					5.7E+02	n		1.2E-01	n			
4.0E-03	P	1.0E-01	I				1		Benzaldehyde	100-52-7	1.7E+02	c*	8.2E+02	c					1.9E+01	c*	5.0E+00	4.1E-03	c	2.6E-03		
5.5E-02	I	7.8E-06	I	4.0E-03	I	3.0E-02	I	V	Benzene	71-43-2	1.2E+00	c*	5.1E+00	c*	3.6E-01	c*	1.6E+00	c*	4.6E-01	c*		2.3E-04	c*			
1.0E-01	X	3.0E-04	X				1	0.1	Benzenediamine-2-methyl sulfate, 1,4-	6369-59-1	5.4E+00	c**	2.3E+01	c*					7.8E-01	c**		2.2E-04	c**			

Key: I = IRIS; P = PPRTV; D = DWSHA; O = OPP; A = ATSDR; C = Cal EPA; X = APPENDIX PPRTV SCREEN (See FAQ #29); H = HEAST; F = See FAQ; E = see user guide Section 2.3.5; W = see user guide Section 2.3.6; L = see user guide on lead; M = mutagen; S = see user guide Section 5; V = volatile; R = RBA applied (See User Guide for Arsenic notice); c = cancer; n = noncancer; * = where n SL < 100X c SL; ** = where n SL < 10X c SL; SSL values are based on DAF=1; m = Concentration may exceed ceiling limit (See User Guide); s = Concentration may exceed Csat (See User Guide)

Toxicity and Chemical-specific Information										Contaminant		Screening Levels										SSLs		
SFO (mg/kg-day) ⁻¹	k _e IUR (ug/m ³) ⁻¹	k _e RfD _o (mg/kg-day)	k _e RfC _i (mg/m ³)	k _e v _o I	mutagen	GIAB	ABS	C _{sat} (mg/kg)	Analyte	CAS No.	Resident Soil (mg/kg)	key	Industrial Soil (mg/kg)	key	Resident Air (ug/m ³)	key	Industrial Air (ug/m ³)	key	Tapwater r (ug/L)	key	MCL (ug/L)	Risk-based SSL (mg/kg)	key	MCL-based SSL (mg/kg)
6.2E-02	I 3.7E-05 C	2.0E-02	I	V		1		#####	Bromodichloromethane	75-27-4	2.9E-01	c	1.3E+00	c	7.6E-02	c	3.3E-01	c	1.3E-01	c	8.0E+01(F)	3.6E-05	c	2.2E-02
7.9E-03	I 1.1E-06 I	2.0E-02	I	V		1		#####	Bromoform	75-25-2	1.9E+01	c*	8.6E+01	c	2.6E+00	c	1.1E+01	c	3.3E+00	c	8.0E+01(F)	8.7E-04	c	2.1E-02
		1.4E-03	I 5.0E-03	I V		1		#####	Bromomethane	74-83-9	6.8E+00	n	3.0E+01	n	5.2E+00	n	2.2E+01	n	7.5E+00	n		1.9E-03	n	
		5.0E-03	H	V		1		#####	Bromophos	2104-96-3	3.9E+02	n	5.8E+03	n					3.5E+01	n		1.5E-01	n	
1.0E-01	O	1.5E-02	O	1.0E-01 A V		1	0.1	#####	Bromopropane, 1-	106-94-5	2.2E+02	n	9.4E+02	n	1.0E+02	n	4.4E+02	n	2.1E+02	n		6.4E-02	n	
1.0E-01	O	1.5E-02	O	V		1		#####	Bromoxynil	1689-84-5	5.3E+00	c	2.2E+01	c					6.1E-01	c		5.2E-04	c	
3.4E+00	C 3.0E-05 I	3.0E-02	O	2.0E-03 I V		1	0.1	#####	Bromoxynil Octanoate	1689-99-2	6.7E+00	c	3.2E+01	c					2.4E-01	c		2.1E-03	c	
						1		#####	Butadiene, 1,3-	106-99-0	5.8E-02	c*	2.6E-01	c*	9.4E-02	c*	4.1E-01	c*	1.8E-02	c		9.9E-06	c	
						1		#####	Butanoic acid, 4-(2,4-dichlorophenoxy)-	94-82-6	1.9E+03	n	2.5E+04	n					4.5E+02	n		4.2E-01	n	
		1.0E-01	I	V		1		#####	Butanol, N-	71-36-3	7.8E+03	ns	1.2E+05	ns					2.0E+03	n		4.1E-01	n	
		2.0E+00	P 3.0E+01	P V		1		#####	Butyl alcohol, sec-	78-92-2	1.3E+05	s	1.5E+06	s	3.1E+04	n	1.3E+05	n	2.4E+04	n		5.0E+00	n	
		5.0E-02	I	V		1		#####	Butylate	2008-41-5	3.9E+03	n	5.8E+04	n					4.6E+02	n		4.5E-01	n	
2.0E-04	C 5.7E-08 C	3.0E-01	P			1	0.1	#####	Butylated hydroxyanisole	25013-16-5	2.7E+03	c	1.1E+04	c	4.9E+01	c	2.2E+02	c	1.5E+02	c		2.9E-01	c	
3.6E-03	P	5.0E-02	P	V		1	0.1	#####	Butylated hydroxytoluene	128-37-0	1.5E+02	c	6.4E+02	c					3.4E+00	c		1.0E-01	c	
						1		#####	Butylbenzene, n-	104-51-8	3.9E+03	ns	5.8E+04	ns					1.0E+03	n		3.2E+00	n	
		1.0E-01	X	V		1		#####	Butylbenzene, sec-	135-98-8	7.8E+03	ns	1.2E+05	ns					2.0E+03	n		5.9E+00	n	
		1.0E-01	X	V		1	0.1	#####	Butylbenzene, tert-	98-06-6	7.8E+03	ns	1.2E+05	s					6.9E+02	n		1.6E+00	n	
		2.0E-02	A			1	0.1	#####	Cacodylic Acid	75-60-5	1.3E+03	n	1.6E+04	n					4.0E+02	n		1.1E-01	n	
1.8E-03	I	1.0E-03	I 1.0E-05	A		0.025	0.001	#####	Cadmium (Diet)	7440-43-9	7.1E+01	n	9.8E+02	n										
1.8E-03	I	5.0E-04	I 1.0E-05	A		0.05	0.001	#####	Cadmium (Water)	7440-43-9					1.6E-03	c**	6.8E-03	c**	9.2E+00	n	5.0E+00	6.9E-01	n	3.8E-01
		5.0E-01	I 2.2E-03	C		1	0.1	#####	Caprolactam	105-60-2	3.1E+04	n	4.0E+05	nm	2.3E+00	n	9.6E+00	n	9.9E+03	n		2.5E+00	n	
1.5E-01	C 4.3E-05 C	2.0E-03	I			1	0.1	#####	Captafol	2425-06-1	3.6E+00	c*	1.5E+01	c	6.5E-02	c	2.9E-01	c	4.0E-01	c*		7.1E-04	c*	
2.3E-03	C 6.6E-07 C	1.3E-01	I			1	0.1	#####	Captan	133-06-2	2.4E+02	c*	1.0E+03	c	4.3E+00	c	1.9E+01	c	3.1E+01	c*		2.2E-02	c*	
		1.0E-01	I			1	0.1	#####	Carbaryl	63-25-2	6.3E+03	n	8.2E+04	n					1.8E+03	n		1.7E+00	n	
		5.0E-03	I			1	0.1	#####	Carbofuran	1563-66-2	3.2E+02	n	4.1E+03	n					9.4E+01	n	4.0E+01	3.7E-02	n	1.6E-02
7.0E-02	I 6.0E-06 I	1.0E-01	I 7.0E-01	I V		1		#####	Carbon Disulfide	75-15-0	7.7E+02	ns	3.5E+03	ns	7.3E+02	n	3.1E+03	n	8.1E+02	n		2.4E-01	n	
		4.0E-03	I 1.0E-01	I V		1		#####	Carbon Tetrachloride	56-23-5	6.5E-01	c	2.9E+00	c	4.7E-01	c	2.0E+00	c	4.6E-01	c	5.0E+00	1.8E-04	c	1.9E-03
						1		#####	Carbonyl Sulfide	463-58-1	6.7E+01	n	2.8E+02	n	1.0E+02	n	4.4E+02	n	2.1E+02	n		5.1E-01	n	
		1.0E-02	I			1	0.1	#####	Carbosulfan	55285-14-8	6.3E+02	n	8.2E+03	n					5.1E+01	n		1.2E+00	n	
		1.0E-01	I			1	0.1	#####	Carboxin	5234-68-4	6.3E+03	n	8.2E+04	n					1.9E+03	n		1.0E+00	n	
		1.0E-01	I	9.0E-04	I			#####	Ceric oxide	1306-38-3	1.3E+06	nm	5.4E+06	nm	9.4E-01	n	3.9E+00	n				4.0E-01	n	
		1.5E-02	I	V		1	0.1	#####	Chloral Hydrate	302-17-0	7.8E+03	n	1.2E+05	nm					2.9E+02	n		7.0E-02	n	
4.0E-01	H					1	0.1	#####	Chloramben	133-90-4	9.5E+02	n	1.2E+04	n					1.8E-01	c		1.5E-04	c	
3.5E-01	I 1.0E-04 I	5.0E-04	I 7.0E-04	I V		1	0.04	#####	Chloranil	118-75-2	1.3E+00	c	5.7E+00	c										
1.0E+01	I 4.6E-03 C	3.0E-04	I			1	0.1	#####	Chlordane	12789-03-6	1.7E+00	c*	7.7E+00	c*	2.8E-02	c*	1.2E-01	c*	2.0E-02	c*	2.0E+00	2.7E-03	c*	2.7E-01
						1	0.1	#####	Chlordecone (Kepone)	143-50-0	5.4E-02	c	2.3E-01	c	6.1E-04	c	2.7E-03	c	3.5E-03	c		1.2E-04	c	
		7.0E-04	A			1	0.1	#####	Chlorfenvinphos	470-90-6	4.4E+01	n	5.7E+02	n					1.1E+01	n		3.1E-02	n	
		9.0E-02	O			1	0.1	#####	Chlorimuron, Ethyl-	90982-32-4	5.7E+03	n	7.4E+04	n					1.8E+03	n		6.0E-01	n	
		1.0E-01	I 1.5E-04	A V		1		#####	Chlorine	7782-50-5	1.8E-01	n	7.8E-01	n	1.5E-01	n	6.4E-01	n	3.0E-01	n		1.4E-04	n	
		3.0E-02	I 2.0E-04	I V		1		#####	Chlorine Dioxide	10049-04-4	2.3E+03	n	3.4E+04	n	2.1E-01	n	8.8E-01	n	4.2E-01	n				
		3.0E-02	I			1		#####	Chlorite (Sodium Salt)	7758-19-2	2.3E+03	n	3.5E+04	n					6.0E+02	n	1.0E+03			
		5.0E+01	I V			1		#####	Chloro-1,1-difluoroethane, 1-	75-68-3	5.4E+04	ns	2.3E+05	s	5.2E+04	n	2.2E+05	n	1.0E+05	n		5.2E+01	n	
		3.0E-04	I	2.0E-02	H 2.0E-02	I V		#####	Chloro-1,3-butadiene, 2-	126-99-8	1.0E-02	c	4.4E-02	c	9.4E-03	c	4.1E-02	c	1.9E-02	c		9.8E-06	c	
4.6E-01	H	2.0E-02	H 2.0E-02	I V		1	0.1	#####	Chloro-2-methylaniline HCl, 4-	3165-93-3	1.2E+00	c	5.0E+00	c					1.7E-01	c		1.5E-04	c	
1.0E-01	P 7.7E-05 C	3.0E-03	X			1	0.1	#####	Chloro-2-methylaniline, 4-	95-69-2	5.4E+00	c*	2.3E+01	c	3.6E-02	c	1.6E-01	c	7.0E-01	c*		4.0E-04	c*	
2.7E-01	X			V		1		#####	Chloroacetaldehyde, 2-	107-20-0	2.6E+00	c	1.2E+01	c					2.9E-01	c	6.0E+01	5.8E-05	c	1.2E-02
		3.0E-05	I			1	0.1	#####	Chloroacetic Acid	79-11-8														
						1	0.1	#####	Chloroacetophenone, 2-	532-27-4	4.3E+04	n	1.8E+05	nm	3.1E-02	n	1.3E-01	n				1.6E-04	c	
2.0E-01	P	4.0E-03	I			1	0.1	#####	Chloroaniline, p-	106-47-8	2.7E+00	c*	1.1E+01	c					3.7E-01	c	1.0E+02	5.3E-02	n	6.8E-02
		2.0E-02	I 5.0E-02	P V		1		#####	Chlorobenzene	108-90-7	2.8E+02	n	1.3E+03	ns	5.2E+01	n	2.2E+02	n	7.8E+01	n		4.7E-01	n	
		1.0E-01	X			1	0.1	#####	Chlorobenzene sulfonic acid, p-	98-66-8	6.3E+03	n	8.2E+04	n					2.0E+03	n				
1.1E-01	C 3.1E-05 C	2.0E-02	I			1	0.1	#####	Chlorobenzilate	510-15-6	4.9E+00	c	2.1E+01	c	9.1E-02	c	4.0E-01	c	3.1E-01	c		1.0E-03	c	
		3.0E-02	X			1	0.1																	

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Toxicity and Chemical-specific Information													Contaminant		Screening Levels										SSLs				
SFO (mg/kg-day) ⁻¹	k _e (y)	IUR (ug/m ³) ⁻¹	k _e (y)	RfD ₀ (mg/kg-day)	k _e (y)	RfC ₁ (mg/m ³) ⁻¹	k _e (y)	muta- gen	GIAB S	ABS	C _{sat} (mg/kg)	Analyte	CAS No.	Resident Soil (mg/kg)	key	Industrial Soil (mg/kg)	key	Resident Air (ug/m ³)	key	Industrial Air (ug/m ³)	key	Tapwater (u/L)	key	MCL (ug/L)	Risk-based SSL (mg/kg)	key	MCL-based SSL (mg/kg)		
5.0E-01	C	8.4E-02	S	3.0E-03	I	1.0E-04	I	M	0.013			Chromium(III), Insoluble Salts	16065-83-1	1.2E+05	n	1.8E+06	nm	3.0E-01	c	6.3E+00	c	1.2E-05	c	1.5E-04	c	3.5E-02	4.0E+07	n	1.8E+05
				1.3E-02	I				0.013	0.1		Chromium(VI)	18540-29-9	2.3E+01	n	3.5E+02	n	3.1E-04	c*	1.4E-03	c*	6.0E+00	n		2.7E-01	n			
				9.0E-03	P	3.0E-04	P	6.0E-06	P			Clofentazine	7440-47-3	3.1E+03	n	4.7E+04	n	1.6E-03	c	2.0E-02	c	8.0E+02	n	1.3E+03	2.8E+01	n	4.6E+01		
				6.2E-04	I			V	M			Cobalt	7440-48-4	3.2E+03	n	4.1E+04	n	6.3E+02	n	2.6E+03	n	9.3E+02	n		7.4E-01	n			
				4.0E-02	H							Coke Oven Emissions	8007-45-2	3.2E+03	n	4.1E+04	n	6.3E+02	n	2.6E+03	n	9.3E+02	n		7.5E-01	n			
				5.0E-02	I	6.0E-01	C			0.1		Copper	7440-50-8	6.3E+03	n	8.2E+04	n	6.3E+02	n	2.6E+03	n	1.9E+03	n		1.5E+00	n			
				5.0E-02	I	6.0E-01	C			0.1		Cresol, m-	108-39-4	3.2E+03	n	4.1E+04	n	6.3E+02	n	2.6E+03	n	9.3E+02	n		1.7E+00	n			
				1.0E-01	A	6.0E-01	C			0.1		Cresol, o-	95-48-7	3.2E+03	n	4.1E+04	n	6.3E+02	n	2.6E+03	n	9.3E+02	n		1.3E+00	n			
				1.0E-01	A	6.0E-01	C			0.1		Cresol, p-	106-44-5	6.3E+03	n	8.2E+04	n	6.3E+02	n	2.6E+03	n	1.9E+03	n		1.5E+00	n			
				1.0E-01	A					0.1		Cresol, p-chloro-m-	59-50-7	6.3E+03	n	8.2E+04	n	6.3E+02	n	2.6E+03	n	1.4E+03	n		1.7E+00	n			
				1.0E-01	A	6.0E-01	C			0.1		Cresols	1319-77-3	6.3E+03	n	8.2E+04	n	6.3E+02	n	2.6E+03	n	1.5E+03	n		1.3E+00	n			
1.9E+00	H			1.0E-03	P			V				Crotonaldehyde, trans-	123-73-9	3.7E-01	c	1.7E+00	c					4.0E-02	c		8.2E-06	c			
				1.0E-01	I	4.0E-01	I	V				Cumene	98-82-8	1.9E+03	ns	9.9E+03	ns	4.2E+02	n	1.8E+03	n	4.5E+02	n		7.4E-01	n			
2.2E-01	C	6.3E-05	C							0.1		Cupferron	135-20-6	2.5E+00	c	1.0E+01	c	4.5E-02	c	1.9E-01	c	3.5E-01	c		6.1E-04	c			
8.4E-01	H			2.0E-03	H					0.1		Cyanazine	21725-46-2	6.5E-01	c	2.7E+00	c					8.8E-02	c		4.1E-05	c			
				1.0E-03	I							Cyanides	592-01-8	7.8E+01	n	1.2E+03	n					2.0E+01	n		n				
				5.0E-03	I							~Calcium Cyanide	544-92-3	3.9E+02	n	5.8E+03	n					1.0E+02	n		n				
				6.0E-04	I	8.0E-04	S	V				~Copper Cyanide	57-12-5	2.3E+01	n	1.5E+02	n	8.3E-01	n	3.5E+00	n	1.5E+00	n	2.0E+02	1.5E-02	n	2.0E+00		
				1.0E-03	I			V				~Cyanide (CN-)	460-19-5	7.8E+01	n	1.2E+03	n					2.0E+01	n		n				
				9.0E-02	I			V				~Cyanogen	506-68-3	7.0E+03	n	1.1E+05	nm					1.8E+03	n		n				
				5.0E-02	I			V				~Cyanogen Bromide	506-77-4	3.9E+03	n	5.8E+04	n					1.0E+03	n		n				
				6.0E-04	I	8.0E-04	I	V				~Hydrogen Cyanide	74-90-8	2.3E+01	n	1.5E+02	n	8.3E-01	n	3.5E+00	n	1.5E+00	n		1.5E-02	n			
				2.0E-03	I							~Potassium Cyanide	161-50-8	1.6E+02	n	2.3E+03	n					4.0E+01	n		n				
				5.0E-03	I					0.04		~Potassium Silver Cyanide	506-61-6	3.9E+02	n	5.8E+03	n					8.2E+01	n		n				
				1.0E-01	I					0.04		~Silver Cyanide	506-64-9	7.8E+03	n	1.2E+05	nm					1.8E+03	n		n				
				1.0E-03	I							~Sodium Cyanide	143-33-9	7.8E+01	n	1.2E+03	n					2.0E+01	n	2.0E+02	n				
				2.0E-04	P							~Thiocyanates	E1790664	1.6E+01	n	2.3E+02	n					4.0E+00	n		n				
				2.0E-04	X			V				~Thiocyanic Acid	463-56-9	1.6E+01	n	2.3E+02	n					4.0E+00	n		n				
				5.0E-02	I							~Zinc Cyanide	557-21-1	3.9E+03	n	5.8E+04	n					1.0E+03	n		n				
2.0E-02	X			2.0E-02	X	6.0E+00	I	V		0.1		Cyclohexane	110-82-7	6.5E+03	ns	2.7E+04	ns	6.3E+03	n	2.6E+04	n	1.3E+04	n		1.3E+01	n			
												Cyclohexane, 1,2,3,4,5-pentabromo-6-chloro-	87-84-3	2.7E+01	c*	1.1E+02	c					2.8E+00	c		1.6E-02	c			
				5.0E+00	I	7.0E-01	P	V				Cyclohexanone	108-94-1	2.8E+04	ns	1.3E+05	s	7.3E+02	n	3.1E+03	n	1.4E+03	n		3.4E-01	n			
				5.0E-03	P	1.0E+00	X	V				Cyclohexene	110-83-8	3.1E+02	ns	3.1E+03	ns	1.0E+03	n	4.4E+03	n	7.0E+01	n		4.6E-02	n			
				2.0E-01	I			V				Cyclohexylamine	108-91-8	1.6E+04	n	2.3E+05	nm					3.8E+03	n		1.0E+00	n			
				2.5E-02	I					0.1		Cyfluthrin	68359-37-5	1.6E+03	n	2.1E+04	n					1.2E+02	n		3.1E+01	n			
				1.0E-03	O					0.1		Cyhalothrin	68085-85-8	6.3E+01	n	8.2E+02	n					2.0E+01	n		1.4E+01	n			
				5.0E-01	O					0.1		Cyromazine	66215-27-8	3.2E+04	n	4.1E+05	nm					9.9E+03	n		2.5E+00	n			
2.4E-01	I	6.9E-05	C	3.0E-05	X					0.1		DDD, p,p'- (DDD)	72-54-8	1.9E+00	n	9.6E+00	c**	4.1E-02	c	1.8E-01	c	3.2E-02	c**		7.5E-03	c**			
3.4E-01	I	9.7E-05	C	3.0E-04	X			V				DDE, p,p'-	72-55-9	2.0E+00	c*	9.3E+00	c*	2.9E-02	c	1.3E-01	c	4.6E-02	c		1.1E-02	c			
3.4E-01	I	9.7E-05	I	5.0E-04	I					0.03		DDT	50-29-3	1.9E+00	c*	8.5E+00	c*	2.9E-02	c	1.3E-01	c	2.3E-01	c*		7.7E-02	c*			
				3.0E-02	I					0.1		Dalapon	75-99-0	1.9E+03	n	2.5E+04	n					6.0E+02	n	2.0E+02	1.2E-01	n	4.1E-02		
1.8E-02	C	5.1E-06	C	1.5E-01	I					0.1		Daminozide	1596-84-5	3.0E+01	c	1.3E+02	c	5.5E-01	c	2.4E+00	c	4.3E+00	c		9.5E-04	n			
7.0E-04	I			7.0E-03	I					0.1		Decabromodiphenyl ether, 2,2',3,3',4,4',5,5',6,6'- (BDE-209)	1163-19-5	4.4E+02	n	3.3E+03	c**					1.1E+02	c**		6.2E+01	c**			
				4.0E-05	I					0.1		Demeton	8065-48-3	2.5E+00	n	3.3E+01	n					4.2E-01	n		n				
1.2E-03	I			6.0E-01	I					0.1		Di(2-ethylhexyl)adipate	103-23-1	4.5E+02	c*	1.9E+03	c					6.5E+01	c	4.0E+02	4.7E+00	c	2.9E+01		
6.1E-02	H									0.1		Diallate	2303-16-4	8.9E+00	c	3.8E+01	c					5.4E-01	c		8.0E-04	c			
				7.0E-04	A					0.1		Diazinon	333-41-5	4.4E+01	n	5.7E+02	n					1.0E+01	n		6.5E-02	n			
				1.0E-02	X			V				Dibenzothiophene	132-65-0	7.8E+02	n	1.2E+04	n					6.5E+01	n		1.2E+00	n			
8.0E-01	P	6.0E-03	P	2.0E-04	P	2.0E-04	I	V	M			Dibromo-3-chloropropane, 1,2-	96-12-8	5.3E-03	c	6.4E-02	c	1.7E-04	c	2.0E-03	c	3.3E-04	c	2.0E-01	1.4E-07	c	8.6E-05		
				4.0E-04	X			V				Dibromobenzene, 1,3-	108-36-1	3.1E+01	n	4.7E+02	ns					5.3E+00	n		5.1E-03	n			
				1.0E-02	I			V				Dibromobenzene, 1,4-	106-37-6	7.8E															

Toxicity and Chemical-specific Information														Contaminant										Screening Levels										SSLs	
SFO (mg/kg-day) ⁻¹	k _e (ug/m ³ -y) ⁻¹	IUR (ug/m ³ -y) ⁻¹	k _e (mg/kg-day) ⁻¹	RfD _o (mg/kg-day) ⁻¹	k _e (mg/m ³ -y) ⁻¹	RfC _i (mg/m ³ -y) ⁻¹	k _e (mg/m ³ -y) ⁻¹	v _o (mg/m ³ -y) ⁻¹	muta- gen	GIAB	S	ABS	C _{sat} (mg/kg)	Analyte	CAS No.	Resident Soil (mg/kg)	key	Industrial Soil (mg/kg)	key	Resident Air (ug/m ³)	key	Industrial Air (ug/m ³)	key	Tapwater (ug/L)	key	MCL (ug/L)	Risk-based SSL (mg/kg)	key	MCL-based SSL (mg/kg)						
1.6E+01	I	4.6E-03	I	3.0E-05	I	5.0E-03	I	3.0E-04	X	V	1	0.1	#####	Dicortophos	141-66-2	1.9E+00	n	2.5E+01	n			3.1E-01	n	1.3E+00	n	6.0E-01	n		1.4E-04	n	1.4E-04				
				8.0E-02	O	3.0E-04	X	V			1	0.1	#####	Dicyclopentadiene	77-73-6	1.3E+00	n	5.4E+00	n			3.1E-01	n	1.3E+00	n	6.3E-01	n		2.2E-03	n	2.2E-03				
				5.0E-05	I						1	0.1	#####	Dieldrin	60-57-1	3.4E-02	c*	1.4E-01	c			6.1E-04	c	2.7E-03	c	1.8E-03	c		7.1E-05	c	7.1E-05				
				3.0E-04	C						1	0.1	#####	Diesel Engine Exhaust	E17136615	1.3E+02	n	1.6E+03	n			9.4E-03	c	4.1E-02	c	4.0E+01	n		8.1E-03	n	8.1E-03				
				2.0E-03	P	2.0E-04	P				1	0.1	#####	Diethanolamine	111-42-2	1.9E+03	n	2.4E+04	n			2.1E-01	n	8.8E-01	n	6.0E+02	n		1.3E-01	n	1.3E-01				
				3.0E-02	P	1.0E-04	P				1	0.1	#####	Diethylene Glycol Monobutyl Ether	112-34-5	3.8E+03	n	2.4E+04	n			1.0E-01	n	4.4E-01	n	6.0E+02	n		2.4E-01	n	2.4E-01				
				6.0E-02	P	3.0E-04	X	V			1	0.1	#####	Diethylene Glycol Monoethyl Ether	111-90-0	7.8E+01	n	1.2E+03	c			3.1E-01	n	1.3E+00	n	2.0E+01	n		4.1E-03	n	4.1E-03				
				1.0E-03	P			V			1	0.1	#####	Diethylformamide	617-84-5	1.6E+03	c	6.8E-03	c			2.8E-05	c	1.2E-04	c	5.1E-05	c		2.8E-05	c	2.8E-05				
				8.3E-02	O						1	0.1	#####	Diethylstilbestrol	56-53-1	5.2E+03	n	6.8E+04	n							1.7E+03	n		2.6E+02	n	2.6E+02				
				2.0E-02	I						1	0.1	#####	Difluorobenzuron	43222-48-6	1.3E+03	n	1.6E+04	n							2.9E+02	n		3.3E-01	n	3.3E-01				
						4.0E+01	I	V			1	0.1	#####	Diffuoroethane, 1,1-	75-37-6	4.8E+04	ns	2.0E+05	ns			4.2E+04	n	1.8E+05	n	8.3E+04	n		2.8E+01	n	2.8E+01				
						3.0E+01	X	V			1	0.1	#####	Diffuoropropane, 2,2-	420-45-1	2.4E+04	ns	1.0E+05	ns			3.1E+04	n	1.3E+05	n	6.3E+04	n		1.4E+02	n	1.4E+02				
								V			1	0.1	#####	Dihydrosafrole	94-58-6	9.9E+00	c	4.5E+01	c			2.2E-01	c	9.4E-01	c	3.0E-01	c		1.9E-04	c	1.9E-04				
						7.0E-01	P	V			1	0.1	#####	Diisopropyl Ether	108-20-3	2.2E+03	n	9.4E+03	n			7.3E+02	n	3.1E+03	n	1.5E+03	n		3.7E-01	n	3.7E-01				
				8.0E-02	I			V			1	0.1	#####	Diisopropyl Methylphosphonate	1445-75-6	6.3E+03	ns	9.3E+04	ns							1.6E+03	n		4.5E-01	n	4.5E-01				
				2.2E-02	O						1	0.1	#####	Dimethipin	55290-64-7	1.4E+03	n	1.8E+04	n							4.4E+02	n		9.6E-02	n	9.6E-02				
				2.2E-03	O						1	0.1	#####	Dimethoate	60-51-5	1.4E+02	n	1.8E+03	n							4.4E+01	n		9.9E-03	n	9.9E-03				
1.6E+00	P										1	0.1	#####	Dimethoxybenzidine, 3,3'-	119-60-4	3.4E-01	c	1.4E+00	c			4.7E-02	c	4.7E-02	c		4.6E-05	c	4.6E-05	c	4.6E-05				
1.7E-03	P										1	0.1	#####	Dimethyl methylphosphonate	758-79-6	3.2E-02	c	1.4E+03	c*			2.2E-03	c	9.4E-03	c	5.0E-03	c		1.2E-05	c	1.2E-05				
4.6E+00	C	1.3E-03	C								1	0.1	#####	Dimethylamino azobenzene [p-]	60-11-7	1.2E-01	c	5.0E-01	c							5.0E-03	c		2.1E-05	c	2.1E-05				
5.8E-01	H										1	0.1	#####	Dimethylaniline HCl, 2,4-	21436-96-4	9.4E-01	c	4.0E+00	c						1.3E-01	c		1.2E-04	c	1.2E-04					
2.0E-01	P										1	0.1	#####	Dimethylaniline, 2,4-	95-68-1	2.7E+00	c*	1.1E+01	c						3.7E-01	c		2.1E-04	c	2.1E-04					
2.7E-02	P										1	0.1	#####	Dimethylaniline, N,N-	121-69-7	2.6E+01	c**	1.2E+02	c*						2.5E+00	c*		9.0E-04	c*	9.0E-04					
1.1E+01	P										1	0.1	#####	Dimethylbenzidine, 3,3'-	119-93-7	4.9E-02	c	2.1E-01	c						6.5E-03	c		4.3E-05	c	4.3E-05					
				1.0E-01	P	3.0E-02	I	V			1	0.1	#####	Dimethylformamide	68-12-2	2.6E+03	n	1.5E+04	n			3.1E+01	n	1.3E+02	n	6.1E+01	n		1.2E-02	n	1.2E-02				
				1.0E-04	X	2.0E-06	X	V			1	0.1	#####	Dimethylhydrazine, 1,1-	57-14-7	5.7E-02	n	2.4E-01	n			2.1E-03	n	8.8E-03	n	4.2E-03	n		9.3E-07	n	9.3E-07				
5.5E+02	C	1.6E-01	C								1	0.1	#####	Dimethylhydrazine, 1,2-	540-73-8	8.8E-04	c	4.1E-03	c			1.8E-05	c	7.7E-05	c	2.8E-05	c		6.5E-09	c	6.5E-09				
				2.0E-02	I						1	0.1	#####	Dimethylphenol, 2,4-	105-67-9	1.3E+03	n	1.6E+04	n							3.6E+02	n		4.2E-01	n	4.2E-01				
				6.0E-04	I						1	0.1	#####	Dimethylphenol, 2,6-	576-26-1	3.8E+01	n	4.9E+02	n							1.1E+01	n		1.3E-02	n	1.3E-02				
				1.0E-03	I						1	0.1	#####	Dimethylphenol, 3,4-	95-65-8	6.3E+01	n	8.2E+02	n							1.8E+01	n		2.1E-02	n	2.1E-02				
4.5E-02	C	1.3E-05	C								1	0.1	#####	Dimethylvinylchloride	513-37-1	1.1E+00	c	4.8E+00	c			2.2E-01	c	9.4E-01	c	3.3E-01	c		1.1E-04	c	1.1E-04				
				8.0E-05	X						1	0.1	#####	Dinitro-o-cresol, 4,6-	534-52-1	5.1E+00	n	6.8E+01	n							1.5E+00	n		2.6E-03	n	2.6E-03				
				2.0E-03	I						1	0.1	#####	Dinitro-o-cyclohexyl Phenol, 4,6-	131-89-5	1.3E+02	n	1.6E+03	n							2.3E+01	n		7.7E-01	n	7.7E-01				
				1.0E-04	P						1	0.1	#####	Dinitrobenzene, 1,2-	528-29-0	6.3E+00	n	8.2E+01	n							1.9E+00	n		1.8E-03	n	1.8E-03				
				1.0E-04	I						1	0.1	#####	Dinitrobenzene, 1,3-	99-65-0	6.3E+00	n	8.2E+01	n							2.0E+00	n		1.8E-03	n	1.8E-03				
				1.0E-04	P						1	0.1	#####	Dinitrobenzene, 1,4-	100-25-4	6.3E+00	n	8.2E+01	n							2.0E+00	n		1.8E-03	n	1.8E-03				
6.8E-01	I										1	0.1	#####	Dinitrophenol, 2,4-	51-28-5	1.3E+02	n	1.6E+03	n						3.9E+01	n		4.4E-02	n	4.4E-02					
				2.0E-03	I						1	0.1	#####	Dinitrotoluene Mixture, 2,4/2,6-	E1615210	8.0E-01	c	3.4E+00	c							1.1E-01	c		1.5E-04	c	1.5E-04				
3.1E-01	C	8.9E-05	C								1	0.102	#####	Dinitrotoluene, 2,4-	121-14-2	1.7E+00	c*	7.4E+00	c			3.2E-02	c	1.4E-01	c	2.4E-01	c		3.2E-04	c	3.2E-04				
1.5E+00	P										1	0.099	#####	Dinitrotoluene, 2,6-	606-20-2	3.6E-01	c*	1.5E+00	c						4.9E-02	c		6.7E-05	c	6.7E-05					
				2.0E-03	S						1	0.006	#####	Dinitrotoluene, 2-Amino-4,6-	35572-78-2	1.5E+02	n	2.3E+03	n							3.9E+01	n		3.0E-02	n	3.0E-02				
4.5E-01	X										1	0.009	#####	Dinitrotoluene, 4-Amino-2,6-	19406-51-0	1.5E+02	n	2.3E+03	n							3.9E+01	n		3.0E-02	n	3.0E-02				
				9.0E-04	X						1	0.1	#####	Dinitrotoluene, Technical grade	25321-14-6	1.2E+00	c*	5.1E+00	c							1.0E-01	c		1.4E-04	c	1.4E-04				
				1.0E-03	I						1	0.1	#####	Dinoseb	88-85-7	6.3E+01	n	8.2E+02	n							1.5E+01	n		1.3E-01	n	1.3E-01				
1.0E-01	I	5.0E-06	I								1	0.1	#####	Dioxane, 1,4-	123-91-1	5.3E+00	c	2.4E+01	c			5.6E-01	c*	2.5E+00	c*	4.6E-01	c		9.4E-05	c	9.4E-05				
6.2E+03	I	1.3E+00	I								1	0.03	#####	-Hexachlorodibenzo-p-dioxin, Mixture		1.0E-04	c	4.7E-04	c			2.2E-06	c	9.4E-06	c	1.3E-05	c		1.7E-05	c	1.7E-05				
1.3E+05	C	3.8E+01	C								1	0.03	#####	-TCDD, 2,3,7,8-	1746-01-6	4.8E-06	c*	2.2E-05	c*			7.4E-08	c	3.2E-07	c	1.2E-07	c		5.9E-08	c	5.9E-08				
				3.0E-02	I	4.0E-08	C	V			1	0.1	#####	Diphenamid	957-51-7	1.9E+03	n	2.5E+04	n							5.3E+02	n		5.2E+00	n	5.2E+00				
						4.0E-04	X	V			1	0.1	#####	Diphenyl Ether	101-84-8	3.4E+01	n	1.4E+02	n			4.2E-01	n	1.8E+00	n	8.3E-01	n		3.4E-03	n	3.4E-03				
				8.0E-04	X						1	0.1	#####	Diphenyl Sulfone	127-63-9	5.1E+01	n	6.6E+02	n							1.5E+01	n		3.6E-02	n	3.6E-02				
				1.0E-01	O						1	0.1	#####	Diphenylamine	122-39-4	6.3E+03	n	8.2E+04	n							1.3E+03	n		2.3E+00	n	2.3E+00				
8.0E-01	I	2.2E-04	I								1	0.1	#####	Diphenylhydrazine, 1,2-	122-66-7	6.8E-01	c	2.9E+00	c			1.3E-02	c	5.6E-02	c	7.8E-02	c		2.5E-04	c	2.5E-04				
				2.2E-03	I						1	0.1	#####	Diquat	85-00-7	1.4E+02	n	1.8E+03	n							4.4E+01	n		8.3E-01	n	8.3E-01				
7.1E+00	C	1.4E-01	C								1	0.1	#####	Direct Black 38																					

Key: I = IRIS; P = PPRTV; D = DWSHA; O = OPP; A = ATSDR; C = Cal EPA; X = APPENDIX PPRTV SCREEN (See FAQ #29); H = HEAST; F = See FAQ; E = see user guide Section 2.3.5; W = see user guide Section 2.3.6; L = see user guide on lead; M = mutagen; S = see user guide Section 5; V = volatile; R = RBA applied (See User Guide for Arsenic notice); c = cancer; n = noncancer; * = where: n SL < 100X c SL; ** = where n SL < 10X c SL; SSL values are based on DAF=1; m = Concentration may exceed ceiling limit (See User Guide); s = Concentration may exceed Csat (See User Guide)

Toxicity and Chemical-specific Information													Contaminant		Screening Levels										SSLs			
SFO (mg/kg-day) ⁻¹	k _e (ug/m ³) ⁻¹	IUR (ug/m ³) ⁻¹	k _e (mg/kg-day)	RfD ₀ (mg/kg-day)	k _e (mg/m ³) ⁻¹	RfC ₁ (mg/m ³) ⁻¹	k _e (mg/m ³) ⁻¹	v _o 	muta- gen	GIAB S	ABS	C _{sat} (mg/kg)	Analyte	CAS No.	Resident Soil (mg/kg)	key	Industrial Soil (mg/kg)	key	Resident Air (ug/m ³)	key	Industrial Air (ug/m ³)	key	Tapwater (ug/L)	key	MCL (ug/L)	Risk-based SSL (mg/kg)	key	MCL-based SSL (mg/kg)
1.1E-02	C	2.5E-06	C	1.0E-05	I	1.0E+00	I	V	1		0.1	#####	Ethyl-p-nitrophenyl Phosphonate	2104-84-5	6.3E-01	n	8.2E+00	n					8.9E-02	n				
				1.0E-01	I							#####	Ethylbenzene	100-41-4	5.8E+00	c	2.5E+01	c	1.1E+00	c	4.9E+00	c	1.5E+00	c	7.0E+02	2.8E-03	n	
				7.0E-02	P						0.1	#####	Ethylene Cyanohydrin	109-78-4	4.4E+03	n	5.7E+04	n					1.4E+03	n		1.7E-03	c	7.8E-01
				9.0E-02	P			V	1			#####	Ethylene Diamine	107-15-3	7.0E+03	n	1.1E+05	nm					1.8E+03	n		4.1E-01	n	
				2.0E+00	I	4.0E-01	C		1		0.1	#####	Ethylene Glycol	107-21-1	1.3E+05	nm	1.6E+06	nm	4.2E+02	n	1.8E+03	n	4.0E+04	n		8.1E+00	n	
				1.0E-01	I	1.6E+00	I		1		0.1	#####	Ethylene Glycol Monobutyl Ether	111-76-2	6.3E+03	n	8.2E+04	n	1.7E+03	n	7.0E+03	n	2.0E+03	n		4.1E-01	n	
3.1E-01	C	3.0E-03	I							1		#####	Ethylene Oxide	75-21-8	2.0E+03	c	2.5E-02	c	3.4E-04	c	4.1E-03	c	6.7E-04	c		1.4E-07	c	
4.5E-02	C	1.3E-05	C	8.0E-05	I					1	0.1	#####	Ethylene Thiourea	96-45-7	5.1E+00	n	5.1E+01	c**	2.2E-01	c	9.4E-01	c	1.6E+00	n		3.6E-04	n	
6.5E+01	C	1.9E-02	C					V	1			#####	Ethyleneimine	151-56-4	2.7E+03	c	1.2E-02	c	1.5E-04	c	6.5E-04	c	2.4E-04	c		5.2E-08	c	
				3.0E+00	I					1	0.1	#####	Ethylphthalyl Ethyl Glycolate	84-72-0	1.9E+05	nm	2.5E+06	nm					5.8E+04	n		1.3E+02	n	
				2.5E-04	I					1	0.1	#####	Fenamiphos	22224-92-6	1.6E+01	n	2.1E+02	n					4.4E+00	n		4.3E-03	n	
				2.5E-02	I					1	0.1	#####	Fenpropathrin	39515-41-8	1.6E+03	n	2.1E+04	n					6.4E+01	n		2.9E+00	n	
				2.5E-02	I					1	0.1	#####	Fenvalerate	51630-58-1	1.6E+03	n	2.1E+04	n					5.0E+02	n		3.2E+02	n	
				1.3E-02	I					1	0.1	#####	Fluometuron	2164-17-2	8.2E+02	n	1.1E+04	n					2.4E+02	n		1.9E-01	n	
				4.0E-02	C	1.3E-02	C		1			#####	Fluoride	7694-48-8	3.1E+03	n	4.7E+04	n	1.4E+01	n	5.7E+01	n	8.0E+02	n	4.0E+03	1.2E+02	n	6.0E+02
				6.0E-02	I	1.3E-02	C			1		#####	Fluorine (Soluble Fluoride)	7782-41-4	4.7E+03	n	7.0E+04	n	1.4E+01	n	5.7E+01	n	1.2E+03	n	4.0E+03	1.8E+02	n	6.0E+02
				8.0E-02	I					1	0.1	#####	Fluridone	59756-60-4	5.1E+03	n	6.6E+04	n					1.4E+03	n		1.6E+02	n	
				4.0E-02	O					1	0.1	#####	Flurprimidol	56425-91-3	2.5E+03	n	3.3E+04	n					6.9E+02	n		3.1E+00	n	
				2.0E-03	O					1	0.1	#####	Flusilazole	85509-19-9	1.3E+02	n	1.6E+03	n					3.1E+01	n		5.1E+00	n	
				5.0E-01	O					1	0.1	#####	Flutolanil	66332-96-5	3.2E+04	n	4.1E+05	nm					7.9E+03	n		4.2E+01	n	
				1.0E-02	I					1	0.1	#####	Fluvalinate	69409-94-5	6.3E+02	n	8.2E+03	n					2.0E+02	n		2.9E+02	n	
				9.0E-02	O					1	0.1	#####	Folpet	133-07-3	5.7E+03	n	7.4E+04	n					1.6E+03	n		3.9E-01	n	
				2.5E-03	O					1	0.1	#####	Fomesafen	72178-02-0	1.6E+02	n	2.1E+03	n					4.8E+01	n		1.6E-01	n	
				2.0E-03	I					1	0.1	#####	Fonofos	944-22-9	1.3E+02	n	1.6E+03	n					2.4E+01	n		4.7E-02	n	
	1.3E-05	I		2.0E-01	I	9.8E-03	A	V	1			#####	Formaldehyde	50-00-0	1.7E+01	c*	7.3E-01	c*	2.2E-01	c*	9.4E-01	c*	4.3E-01	c*		8.7E-05	c*	
				9.0E-01	P	3.0E-04	X	V	1			#####	Formic Acid	64-18-6	2.9E+01	n	1.2E+02	n	3.1E-01	n	1.3E+00	n	6.3E-01	n		1.3E-04	n	
				2.5E+00	O					1	0.1	#####	Fosetyl-AL	39148-24-8	1.6E+05	nm	2.1E+06	nm					5.0E+04	n		6.6E+02	n	
				1.0E-03	X			V	1		0.03	#####	Furans	132-64-9	7.3E+01	n	1.0E+03	n					7.9E+00	n		1.5E-01	n	
				1.0E-03	I					1	0.03	#####	~Dibenzofuran	110-00-9	7.3E+01	n	1.0E+03	n					1.9E+01	n		7.3E-03	n	
3.8E+00	H			9.0E-01	I	2.0E+00	I	V	1		0.03	#####	~Tetrahydrofuran	109-99-9	1.8E+04	n	9.4E+04	n	2.1E+03	n	8.8E+03	n	3.4E+03	n		7.5E-01	n	
				3.0E-03	I	5.0E-02	H	V	1		0.1	#####	Furazolidone	67-45-8	1.4E-01	c	6.0E-01	c					2.0E-02	c		3.9E-05	c	
										1		#####	Furfural	98-01-1	2.1E+02	n	2.6E+03	n	5.2E+01	n	2.2E+02	n	3.8E+01	n		8.1E-03	n	
1.5E+00	C	4.3E-04	C							1	0.1	#####	Furium	531-82-8	3.6E-01	c	1.5E+00	c	6.5E-03	c	2.9E-02	c	5.1E-02	c		6.8E-05	c	
3.0E-02	I	8.6E-06	C							1	0.1	#####	Furmecycloz	60568-05-0	1.8E+01	c	7.7E+01	c	3.3E-01	c	1.4E+00	c	1.1E+00	c		1.2E-03	c	
				6.0E-03	O					1	0.1	#####	Glufosinate, Ammonium	77182-82-2	3.8E+02	n	4.9E+03	n					1.2E+02	n		2.6E-02	n	
				1.0E-01	A	8.0E-05	C		1		0.1	#####	Glutaraldehyde	111-30-8	6.0E+03	n	7.0E+04	n	8.3E-02	n	3.5E-01	n	2.0E+03	n		4.0E-01	n	
				4.0E-04	I	1.0E-03	H	V	1			#####	Glycidyl	765-34-4	2.3E+01	n	2.1E+02	n	1.0E+00	n	4.4E+00	n	1.7E+00	n		3.3E-04	n	
				1.0E-01	I					1	0.1	#####	Glyphosate	1071-83-6	6.3E+03	n	8.2E+04	n					2.0E+03	n	7.0E+02	8.8E+00	n	3.1E+00
				1.0E-02	X			V	1			#####	Guanidine	113-00-8	7.8E+02	n	1.2E+04	n					2.0E+02	n		4.5E-02	n	
				2.0E-02	P					1	0.1	#####	Guanidine Chloride	50-01-1	1.3E+03	n	1.6E+04	n					4.0E+02	n			n	
				3.0E-02	X					1	0.1	#####	Guanidine Nitrate	506-93-4	1.9E+03	n	2.5E+04	n					6.0E+02	n		1.5E-01	n	
				5.0E-05	I					1	0.1	#####	Haloxypol, Methyl	69806-40-2	3.2E+00	n	4.1E+01	n					7.6E-01	n		8.4E-03	n	
4.5E+00	I	1.3E-03	I	5.0E-04	I			V	1			#####	Heptachlor	76-44-8	1.3E-01	c	6.3E-01	c	2.2E-03	c	9.4E-03	c	1.4E-03	c	4.0E-01	1.2E-04	c	3.3E-02
9.1E+00	I	2.6E-03	I	1.3E-05	I			V	1			#####	Heptachlor Epoxide	1024-57-3	7.0E-02	c*	3.3E-01	c*	1.1E-03	c	4.7E-03	c	1.4E-03	c*	2.0E-01	2.8E-05	c*	4.1E-03
				3.0E-04	X	3.0E-03	X	V	1			#####	Heptanal, n-	111-71-7	2.4E+01	n	1.0E+02	n	3.1E+00	n	1.3E+01	n	6.3E+00	n		1.4E-03	n	
				2.0E-03	I	4.0E-01	P	V	1			#####	Heptane, N-	142-82-5	2.2E+01	n	2.9E+02	ns	4.2E+02	n	1.8E+03	n	6.0E+00	n		4.8E-02	n	
				2.0E-04	I					1	0.1	#####	Hexabromobenzene	87-82-1	1.6E+02	n	2.3E+03	n					4.0E+01	n		2.3E-01	n	
1.6E+00	I	4.6E-04	I	8.0E-04	I			V	1			#####	Hexabromodiphenyl ether, 2,2',4,4',5,5'-(BDE-153)	68631-49-2	1.3E+01	n	1.6E+02	n	6.1E-03	c	2.7E-02	c	9.8E-03	c	1.0E+00	1.2E-04	c	1.3E-02
7.8E-02																												

Key: I = IRIS; P = PPRTV; D = DWSHA; O = OPP; A = ATSDR; C = Cal EPA; X = APPENDIX PPRTV SCREEN (See FAQ #29); H = HEAST; F = See FAQ; E = see user guide Section 2.3.5; W = see user guide Section 2.3.6; L = see user guide on lead; M = mutagen; S = see user guide Section 5; V = volatile; R = RBA applied (See User Guide for Arsenic notice); c = cancer; n = noncancer; * = where n SL < 100X c SL; ** = where n SL < 10X c SL; SSL values are based on DAF=1; m = Concentration may exceed ceiling limit (See User Guide); s = Concentration may exceed Csat (See User Guide)

Toxicity and Chemical-specific Information													Contaminant		Screening Levels										SSLs				
SFO (mg/kg-day) ⁻¹	k _e (ug/m ³) ⁻¹	IUR (ug/m ³) ⁻¹	k _e (mg/kg-day)	RfD ₀ (mg/kg-day)	k _e (mg/m ³) ⁻¹	RfC ₁ (mg/m ³) ⁻¹	k _e (mg/m ³) ⁻¹	v _o I	muta	GIAB	ABS	C _{sat} (mg/kg)	Analyte	CAS No.	Resident Soil (mg/kg)	key	Industrial Soil (mg/kg)	key	Resident Air (ug/m ³)	key	Industrial Air (ug/m ³)	key	Tapwater (ug/L)	key	MCL (ug/L)	Risk-based SSL (mg/kg)	key	MCL-based SSL (mg/kg)	
9.5E-04	I		3.0E-01	I								#####	Isobutyl Alcohol	78-83-1	2.3E+04	ns	3.5E+05	s					5.9E+03	n		1.2E+00	n		
			2.0E-01	I	2.0E+00	C					1	0.1	Isophorone	78-59-1	5.7E+02	c*	2.4E+03	c*	2.1E+03	n	8.8E+03	n	7.8E+01	c*		2.6E-02	c*		
			1.5E-02	I									#####	Isopropalin	33820-53-0	1.2E+03	n	1.8E+04	n					4.0E+01	n		9.2E-01	n	
			2.0E+00	P	2.0E-01	P	V					1		Isopropanol	67-63-0	5.6E+03	n	2.4E+04	n	2.1E+02	n	8.8E+02	n	4.1E+02	n		8.4E-02	n	
			1.0E-01	I								1	0.1	Isopropyl Methyl Phosphonic Acid	1832-54-8	6.3E+03	n	8.2E+04	n					2.0E+03	n		4.3E-01	n	
			5.0E-02	I							1	0.1	Isoxaben	82568-50-7	3.2E+03	n	4.1E+04	n					7.3E+02	n		2.0E+00	n		
						3.0E-01	A	V				1		JP-7	E1737665	4.3E+08	nm	1.8E+09	nm	3.1E+02	n	1.3E+03	n	6.3E+02	n		6.3E+02	n	
				8.0E-03	O							1	0.1	Lactofen	77501-63-4	5.1E+02	n	6.6E+03	n					1.0E+02	n		4.6E+00	n	
8.5E-03	C	1.2E-05	C										Lactonitrile	78-97-7	1.3E+01	n	1.6E+02	n					4.0E+00	n		8.1E-04	n		
												1		Lead Compounds ~Lead Phosphate	7446-27-7	8.2E+01	c	3.8E+02	c	2.3E-01	c	1.0E+00	c	9.1E+00	c			c	
8.5E-03	C	1.2E-05	C										~Lead acetate	301-04-2	6.4E+01	c	2.7E+02	c	2.3E-01	c	1.0E+00	c	9.2E+00	c	1.5E+01	1.8E-03	c	1.4E+01	
8.5E-03	C	1.2E-05	C										~Lead and Compounds	7439-92-1	4.0E+02	L	8.0E+02	L	1.5E-01	L			1.5E+01	L			L		
													~Lead subacetate	1335-32-6	6.4E+01	c	2.7E+02	c	2.3E-01	c	1.0E+00	c	9.2E+00	c		2.0E-03	c		
			1.0E-07	I								#####	~Tetraethyl Lead	78-00-2	7.8E-03	n	1.2E-01	n					1.3E-03	n		4.7E-06	n		
				5.0E-06	P							1		Lewisite	541-25-3	3.9E-01	n	5.8E+00	n					9.0E-02	n		3.8E-05	n	
				7.7E-03	O							1	0.1	Linuron	330-55-2	4.9E+02	n	6.3E+03	n					1.3E+02	n		1.1E-01	n	
				2.0E-03	P							1		Lithium	7439-93-2	1.6E+02	n	2.3E+03	n					4.0E+01	n		1.2E+01	n	
				5.0E-04	I							1	0.1	MCPA	94-74-6	3.2E+01	n	4.1E+02	n					7.5E+00	n		2.0E-03	n	
			4.4E-03	O								1	0.1	MCPB	94-81-5	2.8E+02	n	3.6E+03	n					6.5E+01	n		2.6E-02	n	
				1.0E-03	I							1	0.1	MCPA	93-65-2	6.3E+01	n	8.2E+02	n					1.6E+01	n		4.7E-03	n	
				2.0E-02	I							1	0.1	Malathion	121-75-5	1.3E+03	n	1.6E+04	n					3.9E+02	n		1.0E-01	n	
				1.0E-01	I	7.0E-04	C					1	0.1	Maleic Anhydride	108-31-6	6.3E+03	n	8.0E+04	n	7.3E-01	n	3.1E+00	n	1.9E+03	n		3.8E-01	n	
				5.0E-01	I							1	0.1	Maleic Hydrazide	123-33-1	3.2E+04	n	4.1E+05	nm					1.0E+04	n		2.1E+00	n	
			1.0E-04	P							1	0.1	Malononitrile	109-77-3	6.3E+00	n	8.2E+01	n					2.0E+00	n		4.1E-04	n		
				3.0E-02	H						1	0.1	Mancozeb	8018-01-7	1.9E+03	n	2.5E+04	n					5.4E+02	n		7.6E-01	n		
				5.0E-03	I							1	0.1	Maneb	12427-38-2	3.2E+02	n	4.1E+03	n					9.8E+01	n		1.4E-01	n	
				1.4E-01	I	5.0E-05	I					1		Manganese (Diet)	7439-96-5	1.8E+03	n	2.6E+04	n	5.2E-02	n	2.2E-01	n	4.3E+02	n		2.8E+01	n	
				2.4E-02	S	5.0E-05	I				0.04			Manganese (Non-diet)	7439-96-5	1.8E+03	n	2.6E+04	n	5.2E-02	n	2.2E-01	n	4.3E+02	n		2.8E+01	n	
1.1E-02	P		9.0E-05	H							1	0.1	Mephosfolan	950-10-7	5.7E+00	n	7.4E-01	n					1.8E+00	n		2.6E-03	n		
			3.0E-02	I							1	0.1	Mepiquat Chloride	24307-26-4	1.9E+03	n	2.5E+04	n					6.0E+02	n		2.0E-01	n		
			4.0E-03	P							1	0.1	Mercaptoethanolthiazole, 2-	149-30-4	4.9E+01	c**	2.1E+02	c*					6.3E+00	c*		1.8E-02	c*		
													Mercury Compounds	7487-94-7	2.3E+01	n	3.5E+02	n	3.1E-01	n	1.3E+00	n	5.7E+00	n	2.0E+00		n		
													~Mercuric Chloride (and other Mercury salts)	7439-97-6	1.1E+01	ns	4.6E+01	ns	3.1E-01	n	1.3E+00	n	6.3E-01	n	2.0E+00	3.3E-02	n	1.0E-01	
													~Methyl Mercury	22967-92-6	7.8E+00	n	1.2E+02	n					2.0E+00	n		1.4E+01	n		
													~Phenylmercuric Acetate	62-38-4	5.1E+00	n	6.6E+01	n					1.6E+00	n		5.0E-04	n		
													Merphos	150-50-5	2.3E+00	n	3.5E+01	n					6.0E-01	n		5.9E-02	n		
													Merphos Oxide	78-48-8	6.3E+00	n	8.2E+01	n					2.8E-01	n		1.4E-03	n		
													Metallaxyl	57837-19-1	3.8E+03	n	4.9E+04	n					1.2E+03	n		3.3E-01	n		
													Methacrylonitrile	126-98-7	7.5E+00	n	1.0E+02	n	3.1E+01	n	1.3E+02	n	1.9E+00	n		4.3E-04	n		
													Methamidophos	10265-92-6	3.2E+00	nm	4.1E+01	nm					1.0E+00	n		2.1E-04	n		

Toxicity and Chemical-specific Information													Contaminant		Screening Levels										SSLs			
SFO (mg/kg-day) ¹	k _e (y)	IUR (ug/m ³ -y) ¹	k _e (y)	RfD ₀ (mg/kg-day)	k _e (y)	RfC ₀ (mg/m ³ -y)	k _e (y)	muta- gen	GIAB S	ABS 0.1	C _{sat} (mg/kg)		Analyte	CAS No.	Resident Soil (mg/kg)	key	Industrial Soil (mg/kg)	key	Resident Air (ug/m ³)	key	Industrial Air (ug/m ³)	key	Tapwater r (ug/L)	key	MCL (ug/L)	Risk-based SSL (mg/kg)	key	MCL-based SSL (mg/kg)
				7.0E-02	H	6.0E-04	V		1	1	#####		Methylenediphenyl Diisocyanate	101-88-8	8.5E+03	nm	3.6E-02	c	5.5E-04	c	2.4E-03	c	7.8E+02	n		1.2E+00	n	
				1.5E-01	I				1	0.1			Methylstyrene, Alpha-Metolachlor	98-83-9	9.5E+03	nm	8.2E+04	ns					2.7E+03	n		3.2E+00	n	
				2.5E-02	I				1	0.1			Metribuzin	21087-64-9	1.6E+03	n	2.1E+04	n					4.9E+02	n		1.5E-01	n	
				2.5E-01	I				1	0.1			Metsulfuron-methyl	74223-64-6	1.6E+04	nm	2.1E+05	nm					4.9E+03	n		1.9E+00	n	
				3.0E+00	P		V		1		3.42E-01		Mineral oils	8012-95-1	2.3E+05	s	3.5E+06	s					6.0E+04	n		2.4E+03	n	
1.8E+01	C	5.1E-03	C	2.0E-04	I		V		1				Mirex	2385-85-5	3.6E-02	c	1.7E-01	c	5.5E-04	c	2.4E-03	c	8.8E-04	n		6.3E-04	c	
				2.0E-03	I				1	0.1			Molinate	2212-67-1	1.3E+02	n	1.6E+03	n					3.0E+01	n		1.7E-02	n	
				5.0E-03	I				1				Molybdenum	7439-98-7	3.9E+02	n	5.8E+03	n					1.0E+02	n		2.0E+00	n	
				1.0E-01	I				1				Monochloramine	10599-90-3	7.8E+03	n	1.2E+05	nm					2.0E+03	n	4.0E+03		n	
				2.0E-03	P				1	0.1			Monomethylaniline	100-61-8	1.3E+02	n	1.6E+03	n					3.8E+01	n		1.4E-02	n	
				2.5E-02	I				1	0.1			Myclobutanil	88671-89-0	1.6E+03	n	2.1E+04	n					4.5E+02	n		5.6E+00	n	
				3.0E-04	X				1	0.1			N,N'-Diphenyl-1,4-benzenediamine	74-31-7	1.9E+01	n	2.5E+02	n					3.6E+00	n		3.7E-01	n	
				2.0E-03	I		V		1				Naled	300-76-5	1.6E+02	n	2.3E+03	n					4.0E+01	n		1.8E-02	n	
				3.0E-02	X	1.0E-01	P	V	1				Naphtha, High Flash Aromatic (HFAN)	64742-95-6	2.3E+03	n	3.5E+04	n	1.0E+02	n	4.4E+02	n	1.5E+02	n				
1.8E+00	C	0.0E+00	C	1.2E-01	O				1	0.1			Naphthylamine, 2-	91-59-8	3.0E+01	c	1.3E+00	c					3.9E-02	c		2.0E-04	c	
				2.8E-04	C	1.1E-02	C	1.4E-05	C		0.1		Napropamide	15299-99-7	7.6E+03	n	9.8E+04	n					2.0E+03	n		1.3E+01	n	
				2.6E-04	C	1.1E-02	C	1.4E-05	C		0.1		Nickel Acetate	373-02-4	6.7E+02	n	8.1E+03	n	1.1E-02	c**	4.7E-02	c**	2.2E+02	n		4.5E-02	n	
				2.6E-04	C	1.1E-02	C	1.4E-05	C	V			Nickel Carbonate	3333-67-3	7.6E+02	n	8.1E+03	n	1.1E-02	c**	4.7E-02	c**	2.2E+02	n				
				2.6E-04	C	1.1E-02	C	1.4E-05	C		0.04		Nickel Carbonyl	13463-39-3														

Key: I = IRIS; P = PPRTV; D = DWSHA; O = OPP; A = ATSDR; C = Cal EPA; X = APPENDIX PPRTV SCREEN (See FAQ #29); H = HEAST; F = See FAQ; E = see user guide Section 2.3.5; W = see user guide Section 2.3.6; L = see user guide on lead; M = mutagen; S = see user guide Section 5; V = volatile; R = RBA applied (See User Guide for Arsenic notice); c = cancer; n = noncancer; * = where: n SL < 100X c SL; ** = where n SL < 10X c SL; SSL values are based on DAF=1; m = Concentration may exceed ceiling limit (See User Guide); s = Concentration may exceed Csat (See User Guide)

Toxicity and Chemical-specific Information										Contaminant		Screening Levels										SSLs		
SFO (mg/kg-day) ¹	k _e IUR (ug/m ³) ⁻¹	k _e RfD ₀ (mg/kg-day) ¹	k _e RfC ₁ (mg/m ³) ¹	k _e v _o I _l (mg/m ³) ¹	mutagen	GIAB	ABS	C _{sat} (mg/kg)	Analyte	CAS No.	Resident Soil (mg/kg)	key	Industrial Soil (mg/kg)	key	Resident Air (ug/m ³)	key	Industrial Air (ug/m ³)	key	Tapwater (ug/L)	key	MCL (ug/L)	Risk-based SSL (mg/kg)	key	MCL-based SSL (mg/kg)
		7.0E-04	I			1			~Perchlorate and Perchlorate Salts	14797-73-0	5.5E+01	n	8.2E+02	n					1.4E+01	n	1.5E+01(F)		n	
		7.0E-04	I			1			~Potassium Perchlorate	7778-74-7	5.5E+01	n	8.2E+02	n					1.4E+01	n			n	
		7.0E-04	I			1			~Sodium Perchlorate	7601-89-0	5.5E+01	n	8.2E+02	n					1.4E+01	n			n	
		2.0E-02	P			1	0.1		Perfluorobutane sulfonic acid (PFBS)	375-73-5	1.3E+03	n	1.6E+04	n					4.0E+02	n		1.3E-01	n	
		2.0E-02	P			1	0.1		Perfluorobutanesulfonate	45187-15-3	1.3E+03	n	1.6E+04	n					4.0E+02	n		1.3E-01	n	
2.2E-03	C 6.3E-07 C	5.0E-02	I			1	0.1		Permethrin	52645-53-1	3.2E+03	n	4.1E+04	n					1.0E+03	n		2.4E+02	n	
		2.4E-01	O			1	0.1		Phenacetin	62-44-2	2.5E+02	c	1.0E+03	c	4.5E+00	c	1.9E+01	c	3.4E+01	c		9.7E-03	c	
						1	0.1		Phenmedipham	13684-63-4	1.5E+04	n	2.0E+05	nm					3.8E+03	n		2.1E+01	n	
		3.0E-01	I	2.0E-01 C		1	0.1		Phenol	108-95-2	1.9E+04	n	2.5E+05	nm	2.1E+02	n	8.8E+02	n	5.8E+03	n		3.3E+00	n	
		4.0E-03	I			1	0.1		Phenol, 2-(1-methylethoxy)-, methylcarbamate	114-26-1	2.5E+02	n	3.3E+03	n					7.8E+01	n		2.5E-02	n	
		5.0E-04	X			1	0.1		Phenothiazine	92-84-2	3.2E+01	n	4.1E+02	n					4.3E+00	n		1.4E-02	n	
		2.0E-04	X			1		#####	Phenyl Isothiocyanate	103-72-0	1.6E+01	n	2.3E+02	ns					2.6E+00	n		1.7E-03	n	
1.2E-01	P	6.0E-03	I			1	0.1		Phenylenediamine, m-	108-45-2	3.8E+02	n	4.9E+03	n					1.2E+02	n		3.2E-02	n	
		4.0E-03	P			1	0.1		Phenylenediamine, o-	95-54-5	4.5E+00	c*	1.9E+01	c					6.5E-01	c		1.7E-04	c	
		1.0E-03	X			1	0.1		Phenylenediamine, p-	106-50-3	6.3E+01	n	8.2E+02	n					2.0E+01	n		5.4E-03	n	
1.9E-03	H					1	0.1		Phenylphenol, 2-	90-43-7	2.8E+02	c	1.2E+03	c					3.0E+01	c		4.1E-01	c	
		2.0E-04	H			1	0.1		Phorate	298-02-2	1.3E+01	n	1.6E+02	n					3.0E+00	n		3.4E-03	n	
				3.0E-04 I V		1		#####	Phosgene	75-44-5	3.1E-01	n	1.3E+00	n	3.1E-01	n	1.3E+00	n						
		2.0E-02	I			1	0.1		Phosmet	732-11-6	1.3E+03	n	1.6E+04	n					3.7E+02	n				
		4.9E+01	P			1			Phosphates, Inorganic															
		4.9E+01	P			1			~Aluminum metaphosphate	13776-88-0	3.8E+06	nm	5.7E+07	nm					9.7E+05	n			n	
		4.9E+01	P			1			~Ammonium polyphosphate	68333-79-9	3.8E+06	nm	5.7E+07	nm					9.7E+05	n			n	
		4.9E+01	P			1			~Calcium pyrophosphate	7790-76-3	3.8E+06	nm	5.7E+07	nm					9.7E+05	n			n	
		4.9E+01	P			1			~Diammonium phosphate	7783-28-0	3.8E+06	nm	5.7E+07	nm					9.7E+05	n			n	
		4.9E+01	P			1			~Dicalcium phosphate	7757-93-9	3.8E+06	nm	5.7E+07	nm					9.7E+05	n			n	
		4.9E+01	P			1			~Dimagnesium phosphate	7782-75-4	3.8E+06	nm	5.7E+07	nm					9.7E+05	n			n	
		4.9E+01	P			1			~Dipotassium phosphate	7758-11-4	3.8E+06	nm	5.7E+07	nm					9.7E+05	n			n	
		4.9E+01	P			1			~Disodium phosphate	7558-79-4	3.8E+06	nm	5.7E+07	nm					9.7E+05	n			n	
		4.9E+01	P			1			~Monoaluminum phosphate	13530-50-2	3.8E+06	nm	5.7E+07	nm					9.7E+05	n			n	
		4.9E+01	P			1			~Monoammonium phosphate	7722-76-1	3.8E+06	nm	5.7E+07	nm					9.7E+05	n			n	
		4.9E+01	P			1			~Monocalcium phosphate	7758-23-8	3.8E+06	nm	5.7E+07	nm					9.7E+05	n			n	
		4.9E+01	P			1			~Monomagnesium phosphate	7757-86-0	3.8E+06	nm	5.7E+07	nm					9.7E+05	n			n	
		4.9E+01	P			1			~Monopotassium phosphate	7778-77-0	3.8E+06	nm	5.7E+07	nm					9.7E+05	n			n	
		4.9E+01	P			1			~Monosodium phosphate	7558-80-7	3.8E+06	nm	5.7E+07	nm					9.7E+05	n			n	
		4.9E+01	P			1			~Polyphosphoric acid	8017-16-1	3.8E+06	nm	5.7E+07	nm					9.7E+05	n			n	
		4.9E+01	P			1			~Potassium triphosphate	13845-36-8	3.8E+06	nm	5.7E+07	nm					9.7E+05	n			n	
		4.9E+01	P			1			~Sodium acid pyrophosphate	7758-16-9	3.8E+06	nm	5.7E+07	nm					9.7E+05	n			n	
		4.9E+01	P			1			~Sodium aluminum phosphate (acidic)	7785-88-8	3.8E+06	nm	5.7E+07	nm					9.7E+05	n			n	
		4.9E+01	P			1			~Sodium aluminum phosphate (anhydrous)	10279-59-1	3.8E+06	nm	5.7E+07	nm					9.7E+05	n			n	
		4.9E+01	P			1			~Sodium aluminum phosphate (tetrahydrate)	10305-76-7	3.8E+06	nm	5.7E+07	nm					9.7E+05	n			n	
		4.9E+01	P			1			~Sodium hexametaphosphate	10124-56-8	3.8E+06	nm	5.7E+07	nm					9.7E+05	n			n	
		4.9E+01	P			1			~Sodium polyphosphate	68915-31-1	3.8E+06	nm	5.7E+07	nm					9.7E+05	n			n	
		4.9E+01	P			1			~Sodium trimetaphosphate	7785-84-4	3.8E+06	nm	5.7E+07	nm					9.7E+05	n			n	
		4.9E+01	P			1			~Sodium triphosphate	7758-29-4	3.8E+06	nm	5.7E+07	nm					9.7E+05	n			n	
		4.9E+01	P			1			~Tetrapotassium phosphate	7320-34-5	3.8E+06	nm	5.7E+07	nm					9.7E+05	n			n	
		4.9E+01	P			1			~Tetrasodium pyrophosphate	7722-88-5	3.8E+06	nm	5.7E+07	nm					9.7E+05	n			n	
		4.9E+01	P			1			~Trialuminum sodium tetra decahydrogenoctaorthophosphate (dihydrate)	15136-87-5	3.8E+06	nm	5.7E+07	nm					9.7E+05	n			n	
		4.9E+01	P			1			~Tricalcium phosphate	7758-87-4	3.8E+06	nm	5.7E+07	nm					9.7E+05	n			n	
		4.9E+01	P			1			~Trimagnesium phosphate	7757-87-1	3.8E+06	nm	5.7E+07	nm					9.7E+05	n			n	
		4.9E+01	P			1			~Tripotassium phosphate	7778-53-2	3.8E+06	nm	5.7E+07	nm					9.7E+05	n			n	
		4.9E+01	P			1			~Trisodium phosphate	7601-54-9	3.8E+06	nm	5.7E+07	nm					9.7E+05	n			n	
		3.0E-04	I	3.0E-04 I V		1			Phosphine	7803-51-2	2.3E+01	n	3.5E+02	n	3.1E-01	n	1.3E+00	n	5.7E-01	n			n	
		4.9E+01	P	1.0E-02 I		1			Phosphoric Acid	7664-38-2	3.0E+06	nm	2.9E+07	nm	1.0E+01	n	4.4E+01	n	9.7E+05	n			n	
		2.0E-05	I			1			Phosphorus, White	7723-14-0	1.6E+00	n	2.3E+01	n					4.0E-01	n		1.5E-03	n	
						1			Phthalates															
1.4E-02	I 2.4E-06 C	2.0E-02	I			1	0.1		~Bis(2-ethylhexyl)phthalate	117-81-7	3.9E+01	c*	1.6E+02	c	1.2E+00	c	5.1E+00	c	5.6E+00	c*	6.0E+00	1.3E+00	c*	1.4E+00
1.9E-03	P	2.0E-01	I			1	0.1		~Butyl Benzyl Phthalate	95-58-7	2.9E+02	c*	1.2E+03	c					1.6E+01	c		2.4E-01	c	
		1.0E+00	I			1	0.1		~Butylphenyl Butylglycolate	85-70-1	6.3E+04	n	8.2E+05	nm					1.3E+04	n		3.1E+02	n	
		1.0E-01	I			1	0.1		~Dibutyl Phthalate	84-74-2	6.3E+03	n	8											

Key: I = IRIS; P = PPRTV; D = DWSHA; O = OPP; A = ATSDR; C = Cal EPA; X = APPENDIX PPRTV SCREEN (See FAQ #29); H = HEAST; F = See FAQ; E = see user guide Section 2.3.5; W = see user guide Section 2.3.6; L = see user guide on lead; M = mutagen; S = see user guide Section 5; V = volatile; R = RBA applied (See User Guide for Arsenic notice); c = cancer; n = noncancer; * = where: n SL < 100X c SL; ** = where n SL < 10X c SL; SSL values are based on DAF=1; m = Concentration may exceed ceiling limit (See User Guide); s = Concentration may exceed Csat (See User Guide)

Toxicity and Chemical-specific Information												Contaminant		Screening Levels												SSLs	
SFO (mg/kg-day) ⁻¹	k _e (ug/m ³) ⁻¹	IUR (mg/kg-day)	k _e (mg/m ³) ⁻¹	RfD ₀ (mg/kg-day)	k _e (mg/m ³) ⁻¹	RfC ₁ (mg/m ³) ⁻¹	k _e (mg/m ³) ⁻¹	muta- gen	GIAB	ABS	C _{sat} (mg/kg)	Analyte	CAS No.	Resident Soil (mg/kg)	key	Industrial Soil (mg/kg)	key	Resident Air (ug/m ³)	key	Industrial Air (ug/m ³)	key	Tapwater r (ug/L)	key	MCL (ug/L)	Risk-based SSL (mg/kg)	key	MCL-based SSL (mg/kg)
3.9E+00	E 1.1E-03	E 2.3E-05	E 1.3E-03	E V	1	0.14						~Hexachlorobiphenyl, 2,3,3',4,4',5'- (PCB 157)	69782-90-7	1.2E-01	c*	5.0E-01	c*	2.5E-03	c	1.1E-02	c	4.0E-03	c		1.7E-03	c	
3.9E+00	E 1.1E-03	E 2.3E-05	E 1.3E-03	E V	1	0.14						~Hexachlorobiphenyl, 2,3,3',4,4',5'- (PCB 156)	38380-08-4	1.2E-01	c*	5.0E-01	c*	2.5E-03	c	1.1E-02	c	4.0E-03	c		1.7E-03	c	
3.9E+03	E 1.1E+00	E 2.3E-08	E 1.3E-06	E V	1	0.14						~Hexachlorobiphenyl, 3,3',4,4',5,5'- (PCB 169)	32774-16-6	1.2E-04	c*	5.1E-04	c*	2.5E-06	c	1.1E-05	c	4.0E-06	c		1.7E-06	c	
3.9E+00	E 1.1E-03	E 2.3E-05	E 1.3E-03	E V	1	0.14						~Pentachlorobiphenyl, 2',3,4,4',5'- (PCB 123)	65510-44-3	1.2E-01	c*	4.9E-01	c*	2.5E-03	c	1.1E-02	c	4.0E-03	c		1.0E-03	c	
3.9E+00	E 1.1E-03	E 2.3E-05	E 1.3E-03	E V	1	0.14						~Pentachlorobiphenyl, 2,3',4,4',5'- (PCB 118)	31508-00-6	1.2E-01	c*	4.9E-01	c*	2.5E-03	c	1.1E-02	c	4.0E-03	c		1.0E-03	c	
3.9E+00	E 1.1E-03	E 2.3E-05	E 1.3E-03	E V	1	0.14						~Pentachlorobiphenyl, 2,3,3',4,4'- (PCB 105)	32598-14-4	1.2E-01	c*	4.9E-01	c*	2.5E-03	c	1.1E-02	c	4.0E-03	c		1.0E-03	c	
3.9E+00	E 1.1E-03	E 2.3E-05	E 1.3E-03	E V	1	0.14						~Pentachlorobiphenyl, 2,3,4,4',5'- (PCB 114)	74472-37-0	1.2E-01	c*	5.0E-01	c*	2.5E-03	c	1.1E-02	c	4.0E-03	c		1.0E-03	c	
1.3E+04	E 3.8E+00	E 7.0E-09	E 4.0E-07	E V	1	0.14						~Pentachlorobiphenyl, 3,3',4,4',5'- (PCB 126)	57465-28-8	3.6E-05	c*	1.5E-04	c*	7.4E-07	c	3.2E-06	c	1.2E-06	c		3.0E-07	c	
2.0E+00	I 5.7E-04	I		V	1	0.14						~Polychlorinated Biphenyls (high risk)	1336-36-3	2.3E-01	c	9.4E-01	c	4.9E-03	c	2.1E-02	c			5.0E-01			
4.0E-01	I 1.0E-04	I		V	1	0.14						~Polychlorinated Biphenyls (low risk)	1336-36-3					2.8E-02	c	1.2E-01	c	4.4E-02	c	5.0E-01	6.8E-03	c	7.8E-02
7.0E-02	I 2.0E-05	I		V	1	0.14						~Polychlorinated Biphenyls (lowest risk)	1336-36-3					1.4E-01	c	6.1E-01	c						
1.3E+01	E 3.8E-03	E 7.0E-06	E 4.0E-04	E	1	0.14						~Tetrachlorobiphenyl, 3,3',4,4'- (PCB 77)	32598-13-3	3.8E-02	c*	1.6E-01	c*	7.4E-04	c	3.2E-03	c	6.0E-03	c*		9.4E-04	c*	
3.9E+01	E 1.1E-02	E 2.3E-06	E 1.3E-04	E V	1	0.14						~Tetrachlorobiphenyl, 3,4,4',5'- (PCB 81)	70362-50-4	1.2E-02	c*	4.8E-02	c*	2.5E-04	c	1.1E-03	c	4.0E-04	c		6.2E-05	c	
			6.0E-04	I	1	0.1						Polymeric Methylene Diphenyl Diisocyanate (PMDI)	9016-87-9	8.5E+05	nm	3.6E+06	nm	6.3E-01	n	2.6E+00	n						
												Polynuclear Aromatic Hydrocarbons (PAHs)															
			6.0E-02	I	V	1	0.13					~Acenaphthene	83-32-9	3.6E+03	n	4.5E+04	n					5.3E+02	n		5.5E+00	n	
			3.0E-01	I	V	1	0.13					~Anthracene	120-12-7	1.8E+04	n	2.3E+05	nm					1.8E+03	n		5.8E+01	n	
1.0E-01	E 6.0E-05	E		V	M	1	0.13					~Benz[a]anthracene	56-55-3	1.1E+00	c	2.1E+01	c	1.7E-02	c	2.0E-01	c	3.0E-02	c		1.1E-02	c	
1.2E+00	C 1.1E-04	C		V	M	1	0.13					~Benzo[b]fluoranthene	205-82-3	4.2E-01	c	1.8E+00	c	2.6E-02	c	1.1E-01	c	6.5E-02	c		7.8E-02	c	
1.0E+00	I 6.0E-04	I	3.0E-04	I	2.0E-06	I	M	1	0.13			~Benzo[a]pyrene	50-32-8	1.1E-01	c	2.1E+00	c	1.7E-03	c**	8.8E-03	n	2.5E-02	c	2.0E-01	2.9E-02	c	2.4E-01
1.0E-01	E 6.0E-05	E		V	M	1	0.13					~Benzo[b]fluoranthene	205-99-2	1.1E+00	c	2.1E+01	c	1.7E-02	c	2.0E-01	c	2.5E-01	c		3.0E-01	c	
1.0E-02	E 6.0E-06	E		V	M	1	0.13					~Benzo[k]fluoranthene	207-08-9	1.1E+01	c	2.1E+02	c	1.7E-01	c	2.0E+00	c	2.5E+00	c		2.9E+00	c	
			8.0E-02	I	V	1	0.13					~Chloronaphthalene, Beta-	91-58-7	4.8E+03	n	6.0E+04	n					7.5E+02	n		3.9E+00	n	
1.0E-03	E 6.0E-07	E		V	M	1	0.13					~Chrysene	218-01-9	1.1E+02	c	2.1E+03	c	1.7E+00	c	2.0E+01	c	2.5E+01	c		9.0E+00	c	
1.0E+00	E 6.0E-04	E		V	M	1	0.13					~Dibenz[a,h]anthracene	53-70-3	1.1E-01	c	2.1E+00	c	1.7E-03	c	2.0E-02	c	2.5E-02	c		9.6E-02	c	
1.2E+01	C 1.1E-03	C		V	M	1	0.13					~Dibenzo[a,e]pyrene	192-65-4	4.2E-02	c	1.8E-01	c	2.6E-03	c	1.1E-02	c	6.5E-03	c		8.4E-02	c	
2.5E+02	C 7.1E-02	C		V	M	1	0.13					~Dimethylbenz[a]anthracene, 7,12-	57-97-6	4.6E-04	c	8.4E-03	c	1.4E-05	c	1.7E-04	c	1.0E-04	c		9.9E-05	c	
			4.0E-02	I	V	1	0.13					~Fluoranthene	206-44-0	2.4E+03	n	3.0E+04	n					8.0E+02	n		8.9E+01	n	
			4.0E-02	I	V	1	0.13					~Fluorene	86-73-7	2.4E+03	n	3.0E+04	n					2.9E+02	n		5.4E+00	n	
1.0E-01	E 6.0E-05	E		V	M	1	0.13					~Indeno[1,2,3-cd]pyrene	193-39-5	1.1E+00	c	2.1E+01	c	1.7E-02	c	2.0E-01	c	2.5E-01	c		9.8E-01	c	
2.9E-02	P		7.0E-02	A	V	1	0.13				#####	~Methylnaphthalene, 1-	90-12-0	1.8E-01	c	7.3E-01	c					1.1E+00	c		6.0E-03	c	
			4.0E-03	I	V	1	0.13					~Methylnaphthalene, 2-	91-57-6	2.4E+02	n	3.0E+03	n					3.6E+01	n		1.9E-01	n	
		3.4E-05	2.0E-02	I	3.0E-03	I	V	1	0.13			~Naphthalene	91-20-3	3.8E+00	c*	1.7E+01	c*	8.3E-02	c*	3.6E-01	c*	1.7E-01	c*		5.4E-04	c*	
1.2E+00	C 1.1E-04	C		V	M	1	0.13					~Nitropyrene, 4-	57835-92-4	4.2E-01	c	1.8E+00	c	2.6E-02	c	1.1E-01	c	1.9E-02	c		3.3E-03	c	
			3.0E-02	I	V	1	0.13					~Pyrene	129-00-0	1.8E+03	n	2.3E+04	n					1.2E+02	n		1.3E+01	n	
			2.0E-02	P		1	0.1					Potassium Perfluorobutane Sulfonate	29420-49-3	1.3E+03	n	1.6E+04	n					4.0E+02	n			n	
1.5E-01	I		9.0E-03	I	V	1	0.1					Prochloraz	67747-09-5	3.6E+00	c	1.5E+01	c					3.8E-01	c		1.9E-03	c	
			6.0E-03	H	V	1						Profluralin	26399-36-0	4.7E+02	n	7.0E+03	n					2.6E+01	n		1.6E+00	n	
			1.5E-02	I	V	1	0.1					Prometon	1610-18-0	9.5E+02	n	1.2E+04	n					2.5E+02	n		1.2E-01	n	
			4.0E-02	O		1	0.1					Prometryn	7287-19-6	2.5E+03	n	3.3E+04	n					6.0E+02	n		9.0E-01	n	
			1.3E-02	I	V	1	0.1					Propachlor	4918-16-7	8.2E+02	n	1.1E+04	n					2.5E+02	n		1.5E-01	n	
			5.0E-03	I	V	1	0.1					Propanil	709-98-8	3.2E+02	n	4.1E+03	n					8.2E+01	n		4.5E-02	n	
1.9E-01	O		4.0E-02	O		1	0.1					Propargite	2312-35-8	2.8E+00	c	1.2E+01	c					1.6E-01	c		1.1E-02	c	
			2.0E-03	I	V	1	0.1				#####	Propargyl Alcohol	107-19-7	1.6E+02	n	2.3E+03	n					4.0E+01	c		8.1E-03	n	
			2.0E-02	I	V	1	0.1					Propazine	139-40-2	1.3E+03	n	1.6E+04	n					3.4E+02	n		3.0E-01	n	
			2.0E-02	I	V	1	0.1					Propham	122-42-9	1.3E+03	n	1.6E+04	n					3.5E+02	n		2.2E-01	n	
			1.0E-01	O		1	0.1					Propiconazole	60207-90-1	6.3E+03	n	8.2E+04	n					1.6E+03	n				

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Toxicity and Chemical-specific Information										Contaminant		Screening Levels										SSLs			
SFO (mg/kg-day) ⁻¹	k _e IUR (ug/m ³) ⁻¹	k _e RfD ₀ (mg/kg-day) ⁻¹	k _e RfC ₁ (mg/m ³) ⁻¹	k _e v _o I	muta- gen	GIAB S	ABS I	C _{sat} (mg/kg)	Analyte	CAS No.	Resident Soil (mg/kg)	key	Industrial Soil (mg/kg)	key	Resident Air (ug/m ³)	key	Industrial Air (ug/m ³)	key	Tapwater (ug/L)	key	MCL (ug/L)	Risk-based SSL (mg/kg)	key	MCL-based SSL (mg/kg)	
2.4E-02	H	3.0E-02	I	6.0E-01	I	1	0.1		Stirofos (Tetrachlorovinphos)	961-11-5	2.3E+01	c*	9.6E+01	c					2.8E+00	c		8.2E-03	c		
		6.0E-01	I	2.0E-01	I	1	0.1		Strontium, Stable	7440-24-6	4.7E+04	n	7.0E+05	nm					1.2E+04	n		4.2E+02	n		
		3.0E-04	I	2.0E-01	I	1	0.1		Strychnine	57-24-9	1.9E+01	n	2.5E+02	n					5.9E+00	n		6.5E-02	n		
		2.0E-01	I	1.0E+00	I V	1	0.1	#####	Styrene	100-42-5	6.0E+03	ns	3.5E+04	ns	1.0E+03	n	4.4E+03	n	1.2E+03	n	1.0E+02	1.3E+00	n	1.1E-01	
		3.0E-03	P	2.0E-03	X	1	0.1		Styrene-Acrylonitrile (SAN) Trimer		1.9E+02	n	2.5E+03	n					4.8E+01	n			n		
		1.0E-03	P	2.0E-03	X	1	0.1		Sulfolane	126-33-0	6.3E+01	n	8.2E+02	n	2.1E+00	n	8.8E+00	n	2.0E+01	n		4.4E-03	n		
		8.0E-04	P	1.0E-03	C V	1	0.1		Sulfonylbis(4-chlorobenzene), 1,1'-	90-07-9	5.1E+01	n	6.6E+02	n					1.1E+01	n		6.5E-02	n		
				1.0E-03	C V	1			Sulfur Trioxide	7446-11-9	1.4E+06	nm	6.0E+06	nm	1.0E+00	n	4.4E+00	n	2.1E+00	n			n		
2.5E-02	I	7.1E-06	I	5.0E-02	H	1	0.1		Sulfuric Acid	7664-93-9	1.4E+06	nm	6.0E+06	nm	1.0E+00	n	4.4E+00	n		1.3E+00	c		1.5E-02	c	
		3.0E-02	H	2.0E-02	H	1	0.1		Sulfurous acid, 2-chloroethyl 2-[4-(1,1-dimethylethyl)phenoxy]-1-methylethyl	140-57-8	2.2E+01	c	9.2E+01	c	4.0E-01	c	1.7E+00	c					n		
		7.0E-02	I	1.3E-02	I	1	0.1		TCMTB	21564-17-0	1.9E+03	n	2.5E+04	n					4.8E+02	n		3.3E+00	n		
		2.0E-02	H	1.3E-02	I	1	0.1		Tebuthiuron	34014-18-1	4.4E+03	n	5.7E+04	n					1.4E+03	n		3.9E-01	n		
		1.3E-02	I	2.5E-05	H	1	0.1	#####	Temephos	3383-96-8	1.3E+03	n	1.6E+04	n					4.0E+02	n		7.6E+01	n		
		1.0E-03	I	1.0E-03	I	1	0.1		Terbacil	5902-51-2	8.2E+02	n	1.1E+04	n					2.5E+02	n		7.5E-02	n		
		3.0E-04	I	1.0E-03	I	1	0.1		Terbufos	13071-79-9	2.0E+00	n	2.9E+01	n					2.4E-01	n		5.2E-04	n		
		3.0E-02	I	1.0E-04	I	1	0.1		Terbutryn	886-50-0	6.3E+01	n	8.2E+02	n					1.3E+01	n		1.9E-02	n		
		3.0E-02	I	1.0E-04	I	1	0.1		Tetrabromodiphenyl ether, 2,2',4,4'-(BDE-47)	5436-43-1	6.3E+00	n	8.2E+01	n					2.0E+00	n		5.3E-02	n		
2.6E-02	I	7.4E-06	I	3.0E-02	I	1	0.1	#####	Tetrachlorobenzene, 1,2,4,5-	95-94-3	2.3E+01	n	3.5E+02	n					1.7E+00	n		7.9E-03	n		
2.0E-01	I	5.8E-05	C	2.0E-02	I	1	0.1	#####	Tetrachloroethane, 1,1,1,2-	630-20-6	2.0E+00	c	8.8E+00	c	3.8E-01	c	1.7E+00	c	5.7E-01	c		2.2E-04	c		
		2.0E-02	I	2.0E-02	I	1	0.1	#####	Tetrachloroethane, 1,1,2,2-	79-34-5	6.0E-01	c	2.7E+00	c	4.8E-02	c	2.1E-01	c	7.6E-02	c		3.0E-05	c		
2.1E-03	I	2.6E-07	I	6.0E-03	I	1	0.1	#####	Tetrachloroethylene	127-18-4	2.4E+01	c**	1.0E+02	c**	1.1E+01	c**	4.7E+01	c**	1.1E+01	c**	5.0E+00	5.1E-03	c**	2.3E-03	
		3.0E-02	I	4.0E-02	I V	1	0.1		Tetrachlorophenol, 2,3,4,6-	58-90-2	1.9E+03	n	2.5E+04	n					2.4E+02	n		1.8E-01	n		
2.0E+01	H	5.0E-04	I	8.0E+01	I V	1	0.1		Tetrachlorotoluene, p- alpha, alpha-	5216-25-1	3.5E-02	c	1.6E-01	c					1.3E-03	c		4.5E-06	c		
									Tetraethyl Dithiopyrophosphate	3689-24-5	3.2E+01	n	4.1E+02	n					7.1E+00	n		5.2E-03	n		
		2.0E-03	P	8.0E+01	I V	1	0.1	#####	Tetrafluoroethane, 1,1,1,2-	811-97-2	1.0E+05	s	4.3E+05	s	8.3E+04	n	3.5E+05	n	1.7E+05	n		9.3E+01	n		
		1.0E-05	S	1.0E-05	S	1	0.1	#####	Tetryl (Trinitrophenylmethyltrinitramine)	479-45-8	1.6E+02	n	2.3E+03	n					3.9E+01	n		3.7E-01	n		
		1.0E-05	X	1.0E-05	X	1	0.1		Thallic Oxide	1314-32-5	1.6E+00	n	2.3E+01	n					4.0E-01	n			n		
		1.0E-05	X	1.0E-05	X	1	0.1		Thallium (I) Nitrate	10102-45-1	7.8E-01	n	1.2E+01	n					2.0E-01	n			n		
		1.0E-05	X	1.0E-05	X	1	0.1		Thallium (Soluble Salts)	7440-28-0	7.8E-01	n	1.2E+01	n					2.0E-01	n	2.0E+00	1.4E-02	n	1.4E-01	
		1.0E-05	X	1.0E-05	X	1	0.1		Thallium Acetate	663-68-8	7.8E-01	n	1.2E+01	n					2.0E-01	n		4.1E-05	n		
		1.0E-05	X	1.0E-05	X	1	0.1		Thallium Carbonate	65331-73-9	1.6E+00	n	2.3E+01	n					4.0E-01	n		8.3E-05	n		
		1.0E-05	X	1.0E-05	X	1	0.1		Thallium Chloride	7791-12-0	7.8E-01	n	1.2E+01	n					2.0E-01	n			n		
		1.0E-05	S	1.0E-05	S	1	0.1		Thallium Selenite	12039-52-0	7.8E-01	n	1.2E+01	n					2.0E-01	n			n		
		4.3E-02	O	1.0E-02	I	1	0.1		Thallium Sulfate	7446-18-6	1.6E+00	n	2.3E+01	n					4.0E-01	n			n		
		1.0E-02	I	1.0E-02	I	1	0.1		Thiensiulfuron-methyl	79277-27-3	2.7E+03	n	3.5E+04	n					8.6E+02	n		2.6E-01	n		
		7.0E-02	X	3.0E-04	H	1	0.0075		Thiobencarb	28249-77-6	6.3E+02	n	8.2E+03	n					1.6E+02	n		5.5E-01	n		
		3.0E-04	H	1.0E-02	X	1	0.1		Thiodiglycol	111-48-3	5.4E+03	ns	7.9E+04	ns					1.4E+03	n		2.8E-01	n		
		3.0E-04	H	1.0E-02	X	1	0.1		Thiofanox	39196-18-4	1.9E+01	n	2.5E+02	n					5.3E+00	n		1.8E-03	n		
1.2E-02	O	2.7E-02	O	1.5E-02	O	1	0.1		Thiophanate, Methyl	23564-05-8	4.7E+01	c*	2.0E+02	c					6.7E+00	c*		5.7E-03	c*		
		1.5E-02	O	1.5E-02	O	1	0.1		Thiram	137-26-8	9.5E+02	n	1.2E+04	n					2.9E+02	n		4.2E-01	n		
		6.0E-01	H	1.0E-04	A V	1	0.1	#####	Tin	7440-31-5	4.7E+04	n	7.0E+05	nm					1.2E+04	n		3.0E+03	n		
		8.0E-02	I	1.0E-04	A V	1	0.1	#####	Titanium Tetrachloride	7550-45-0	1.4E+05	nm	6.0E+05	nm	1.0E-01	n	4.4E-01	n	2.1E-01	n	1.0E+03	7.6E-01	n	6.9E-01	
1.8E-01	X	1.1E-05	C	2.0E-04	X	1	0.1	#####	Toluene	108-88-3	4.9E+03	ns	4.7E+04	ns	5.2E+03	ns	2.2E+04	ns	1.1E+03	n		2.5E-04	n		
		5.0E-03	P	8.0E-06	C V	1	0.1	#####	Toluene-2,4-diisocyanate	584-84-9	6.4E+00	n	2.7E+01	n	8.3E-03	n	3.5E-02	n	1.7E-02	n			n		
									Toluene-2,5-diamine	95-70-5	3.0E+00	c**	1.3E+01	c*					4.3E-01	c**		1.3E-04	c**		
									Toluene-2,6-diisocyanate	91-08-7	5.3E+00	n	2.2E+01	n	8.3E-03	n	3.5E-02	n	1.7E-02	n		2.6E-04	n		
1.6E-02	P	5.1E-05	C	4.0E-03	X	1	0.1		Toluic Acid, p-	99-94-5	3.2E+02	n	4.1E+03	n					9.0E+01	n		2.3E-02	n		
3.0E-02	P					1	0.1		Toluidine, o- (Methylaniline, 2-)	95-53-4	3.4E+01	c	1.4E+02	c	5.5E-02	c	2.4E-01	c	4.7E+00	c		2.0E-03	c		
		3.0E+00	P	6.0E-01	P V	1	0.1	3.42E-01	Toluidine, p-	106-49-0	1.8E+01	c*	7.7E+01	c*					2.5E+00	c*		1.1E-03	c*		
									Total Petroleum Hydrocarbons (Aliphatic High)	E1790670	2.3E+05	s	3.5E+06	s					6.0E+04	n		2.4E+03	n		
		1.0E-02	X	1.0E-01	P V	1	0.1	#####	Total Petroleum Hydrocarbons (Aliphatic Low)	E1790666	5.2E+02	ns	2.2E+03	ns	6.3E+02	ns	2.6E+03	ns	1.3E+03	n		8.8E+00	n		
		4.0E-02	P	3.0E-02	P V	1	0.1	#####	Total Petroleum Hydrocarbons (Aliphatic Medium)	E1790668	9.6E+01	ns	4.4E+02	ns	1.0E+02	ns	4.4E+02	ns	1.0E+02	n		1.5E+00	n		

Key: I = IRIS; P = PPRTV; D = DWSHA; O = OPP; A = ATSDR; C = Cal EPA; X = APPENDIX PPRTV SCREEN (See FAQ #29); H = HEAST; F = See FAQ; E = see user guide Section 2.3.5; W = see user guide Section 2.3.6; L = see user guide on lead; M = mutagen; S = see user guide Section 5; V = volatile; R = RBA applied (See User Guide for Arsenic notice) ; c = cancer; n = noncancer; * = where: n SL < 100X c SL; ** = where n SL < 10X c SL; SSL values are based on DAF=1; m = Concentration may exceed ceiling limit (See User Guide); s = Concentration may exceed Csat (See User Guide)

Toxicity and Chemical-specific Information										Contaminant			Screening Levels										SSLs		
SFO (mg/kg-day) ¹	k _e IUR (ug/m ³ -y) ¹	k _e RfD ₀ (mg/kg-day) ¹	k _e RfC ₁ (mg/m ³) ¹	k _e V _o I _o gen	muta	GIAB	S	ABS	C _{sat} (mg/kg)	Analyte	CAS No.	Resident Soil (mg/kg)	key	Industrial Soil (mg/kg)	key	Resident Air (ug/m ³)	key	Industrial Air (ug/m ³)	key	Tapwater r (ug/L)	key	MCL (ug/L)	Risk-based SSL (mg/kg)	key	MCL-based SSL (mg/kg)
1.1E-02	I	3.1E-06	I	V	1	1	0.1	0.1	#####	Trichlorofluoromethane	75-69-4	2.3E+04	ns	3.5E+05	s					5.2E+03	n		3.3E+00	n	
										Trichlorophenol, 2,4,5-	95-95-4	6.3E+03	n	8.2E+04	n					1.2E+03	n		4.0E+00	n	
										Trichlorophenol, 2,4,6-	88-06-2	4.9E+01	c**	2.1E+02	c**	9.1E-01	c	4.0E+00	c	4.1E+00	c**		4.0E-03	c**	
3.0E+01	I	I	I	V	1	1	0.1	0.1	#####	Trichlorophenoxyacetic Acid, 2,4,5-	93-76-5	6.3E+02	n	8.2E+03	n					1.6E+02	n	5.0E+01	6.8E-02	n	
										Trichlorophenoxypropionic acid, -2,4,5	93-72-1	5.1E+02	n	6.6E+03	n					1.1E+02	n		6.1E-02	n	2.8E-02
										Trichloropropane, 1,1,2-	598-77-6	3.9E+02	n	5.8E+03	ns					8.8E+01	n		3.5E-02	n	
7.7E-03	I	7.5E-03	I	V	1	1	0.1	0.1	#####	Trichloropropane, 1,2,3-	96-18-4	5.1E-03	c	1.1E-01	c	3.1E-01	n	1.3E+00	n	7.5E-04	c		3.2E-07	c	
										Trichloropropene, 1,2,3-	96-19-5	7.3E-01	n	3.1E+00	n	3.1E-01	n	1.3E+00	n	6.2E-01	n		3.1E-04	n	
										Tricresyl Phosphate (TCP)	1330-78-5	1.3E+03	n	1.6E+04	n					1.6E+02	n		1.5E+01	n	
2.0E+00	P	7.0E-03	I	V	1	1	0.1	0.1	#####	Tridiphane	58138-08-2	1.9E+02	n	2.5E+03	n					1.8E+01	n		1.3E-01	n	
										Triethylamine	121-44-8	1.2E+02	n	4.8E+02	n	7.3E+00	n	3.1E+01	n	1.5E+01	n		4.4E-03	n	
										Triethylene Glycol	112-27-6	1.3E+05	nm	1.6E+06	nm					4.0E+04	n		8.8E+00	n	
2.0E-02	P	7.5E-03	I	V	1	1	0.1	0.1	#####	Trifluoroethane, 1,1,1-	420-46-2	1.5E+04	ns	6.2E+04	ns	2.1E+04	n	8.8E+04	n	4.2E+04	n		1.3E+02	n	
										Trifluralin	1582-09-8	9.0E+01	c**	4.2E+02	c*					2.6E+00	c*		8.4E-02	c*	
										Trimethyl Phosphate	512-56-1	2.7E+01	c*	1.1E+02	c*					3.9E+00	c*		8.6E-04	c*	
3.0E-02	I	3.0E-02	I	V	1	1	0.19	0.19	#####	Trimethylbenzene, 1,2,3-	526-73-8	3.4E+02	ns	2.0E+03	ns	6.3E+01	n	2.6E+02	n	5.5E+01	n		8.1E-02	n	
										Trimethylbenzene, 1,2,4-	95-63-6	3.0E+02	ns	1.8E+03	ns	6.3E+01	n	2.6E+02	n	5.6E+01	n		8.1E-02	n	
										Trimethylbenzene, 1,3,5-	108-67-8	2.7E+02	ns	1.5E+03	ns	6.3E+01	n	2.6E+02	n	6.0E+01	n		8.7E-02	n	
2.3E+00	C	6.6E-04	C	V	1	1	0.1	0.1	#####	Trimethylpentene, 2,4,4-	25167-70-8	7.8E+02	ns	1.2E+04	ns					6.5E+01	n		2.2E-01	n	
										Trinitrobenzene, 1,3,5-	99-35-4	2.2E+03	n	3.2E+04	n					5.9E+02	n		2.1E+00	n	
										Trinitrotoluene, 2,4,6-	118-96-7	2.1E+01	c**	9.6E+01	c**					2.5E+00	c**		1.5E-02	c**	
3.2E-03	P	7.0E-03	P	V	1	1	0.1	0.1	#####	Triphenylphosphine Oxide	791-28-6	1.3E+03	n	1.6E+04	n					3.6E+02	n		1.5E+00	n	
										Tris(1,3-Dichloro-2-propyl) Phosphate	13674-87-8	1.3E+03	n	1.6E+04	n					3.6E+02	n		8.0E+00	n	
										Tris(1-chloro-2-propyl)phosphate	13674-84-6	6.3E+02	n	8.2E+03	n					1.9E+02	n		6.5E-01	n	
1.0E+00	C	2.9E-04	C	V	1	1	0.1	0.1	#####	Tris(2,3-dibromopropyl)phosphate	126-72-7	2.8E-01	c	1.3E+00	c	4.3E-03	c	1.9E-02	c	6.8E-03	c		1.3E-04	c	
										Tris(2-chloroethyl)phosphate	115-96-8	2.7E+01	c*	1.1E+02	c*					3.8E+00	c*		3.8E-03	c*	
										Tris(2-ethylhexyl)phosphate	78-42-2	1.7E+02	c*	7.2E+02	c					2.4E+01	c*		1.2E+02	c*	
8.3E-03	P	9.0E-03	I	V	1	1	0.026	0.026	#####	Tungsten	7440-33-7	6.3E+01	n	9.3E+02	n					1.6E+01	n		2.4E+00	n	
										Uranium (Soluble Salts)	E715565	1.6E+01	n	2.3E+02	n	4.2E-02	n	1.8E-01	n	4.0E+00	n	3.0E+01	1.8E+00	n	1.4E+01
										Urethane	51-79-6	1.2E-01	c	2.3E+00	c	3.5E-03	c	4.2E-02	c	2.5E-02	c		5.6E-06	c	
7.2E-01	I	4.4E-06	I	V	1	1	0.1	0.1	#####	Vanadium Pentoxide	1314-62-1	4.6E+02	c**	2.0E+03	c**	3.4E-04	c*	1.5E-03	c*	1.5E+02	n			n	
										Vanadium and Compounds	7440-62-2	3.9E+02	n	5.8E+03	n	1.0E-01	n	4.4E-01	n	8.6E+01	n		8.6E+01	n	
										Vernolate	1929-77-7	7.8E+01	n	1.2E+03	n					1.1E+01	n		8.9E-03	n	
3.2E-05	H	3.0E-03	I	V	1	1	0.1	0.1	#####	Vinclozolin	50471-44-8	7.6E+01	n	9.8E+02	n					2.1E+01	n		1.6E-02	n	
										Vinyl Acetate	108-05-4	9.1E+02	n	3.8E+03	ns	2.1E+02	n	8.8E+02	n	4.1E+02	n		8.7E-02	n	
										Vinyl Bromide	593-60-2	1.2E-01	c*	5.2E-01	c*	8.8E-02	c*	3.8E-01	c*	1.8E-01	c*		5.1E-05	c*	
2.0E-01	S	1.0E-01	S	V	1	1	0.1	0.1	#####	Vinyl Chloride	75-01-4	5.9E-02	c	1.7E+00	c	1.7E-01	c	2.8E+00	c	1.9E-02	c	2.0E+00	6.5E-06	c	6.9E-04
										Warfarin	81-81-2	1.9E+01	n	2.5E+02	n					5.6E+00	n		5.9E-03	n	
										Xylene, P-	106-42-3	5.6E+02	ns	2.4E+03	ns	1.0E+02	n	4.4E+02	n	1.9E+02	n		1.9E-01	n	
2.0E-01	S	1.0E-01	S	V	1	1	0.1	0.1	#####	Xylene, m-	108-38-3	5.5E+02	ns	2.4E+03	ns	1.0E+02	n	4.4E+02	n	1.9E+02	n		1.9E-01	n	
										Xylene, o-	95-47-6	6.5E+02	ns	2.8E+03	ns	1.0E+02	n	4.4E+02	n	1.9E+02	n		1.9E-01	n	
										Xylenes	1330-20-7	5.8E+02	ns	2.5E+03	ns	1.0E+02	n	4.4E+02	n	1.9E+02	n	1.0E+04	1.9E-01	n	9.9E+00
5.0E-02	I	5.0E-02	I	V	1	1	0.1	0.1	#####	Zinc Phosphide	1314-84-7	2.3E+01	n	3.5E+02	n					6.0E+00	n			n	
										Zinc and Compounds	7440-66-6	2.3E+04	n	3.5E+05	nm					6.0E+03	n		3.7E+02	n	
										Zineb	12122-67-7	3.2E+03	n	4.1E+04	n					9.9E+02	n		2.9E+00	n	
8.0E-05	X									Zirconium	7440-67-7	6.3E+00	n	9.3E+01	n					1.6E+00	n		4.8E+00	n	